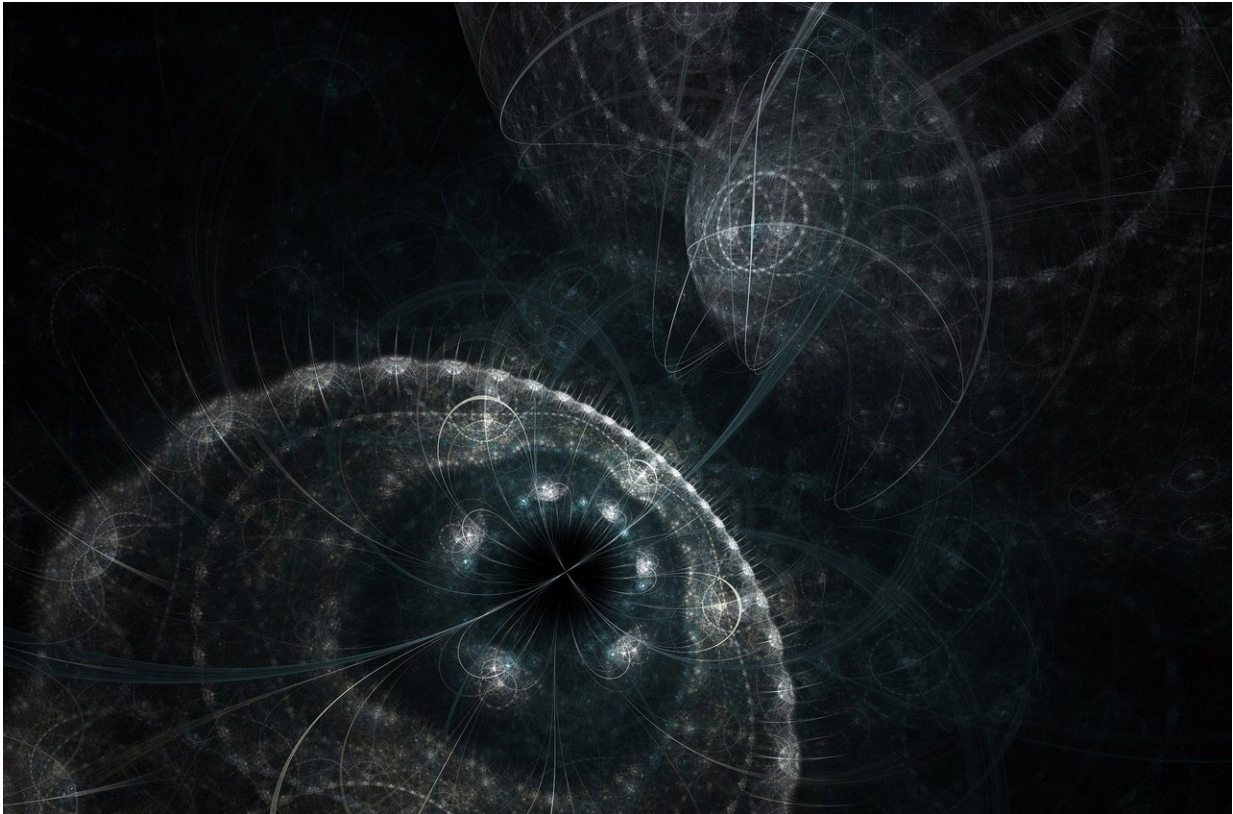


# Quantum chemistry on quantum computers

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Quantum computing and quantum information processing technology have attracted attention in recently emerging fields. Among many important and fundamental issues in science, solving the Schroedinger equation (SE) of atoms and molecules is one of the ultimate goals in chemistry, physics and their related fields. SE is the first principle of

non-relativistic quantum mechanics, whose solutions, termed wave functions, can afford any information of electrons within atoms and molecules, predicting their physicochemical properties and chemical reactions.

Dr. K. Sugisaki, Profs. K. Sato and T. Takui and coworkers, all researchers from Osaka City University (OCU) in Japan, have found a novel [quantum](#) algorithm enabling us to perform full configuration interaction (Full-CI) calculations suitable for "[chemical reactions](#)" without exponential/combinatorial explosion. Full-CI gives the exact numerical solutions of SE, which are intractable problems even for supercomputers. Such a quantum algorithm contributes to the acceleration of implementing practical quantum computers. Since 1929, chemistry and physics have sought to predict complex [chemical](#) reactions by invoking Full-CI approaches, but they have never been successful until now. Full-CI calculations are potentially capable of predicting chemical reactions. The researchers of the current study report a new Full-CI approach implemented on quantum computers for the first time.

The paper is published in *ACS Central Science*.

They write, "As Dirac claimed in 1929 when quantum mechanics was established, the exact application of mathematical theories to solve SE leads to equations too complicated to be solvable. In fact, the number of variables to be determined in the Full-CI method grows exponentially against the system size, and it easily runs into astronomical figures such as exponential explosion. For example, the dimension of the Full-CI [calculation](#) for benzene molecule  $C_6H_6$ , in which only 42 electrons are involved, amounts to  $10^{44}$ , which is impossible to be dealt with by any supercomputer. Worse, molecular systems during the dissociation process are characterized by extremely complex electronic structures (multiconfigurational nature), and relevant numerical calculations are

impossible on any supercomputer."

According to the OCU research group, quantum computers date back to Feynman's suggestion in 1982 that quantum mechanics can be simulated by a [computer](#) itself built of quantum mechanical elements that obey quantum mechanical laws. More than 20 years later, Prof. Aspuru-Guzik, Harvard Univ. (Toronto Univ. since 2018) and coworkers have proposed a quantum algorithm capable of calculating the energies of atoms and molecules not exponentially but polynomially against the number of the variables of the systems, making a breakthrough in the field of quantum chemistry on quantum computers.

When Aspuru's quantum algorithm is applied to the Full-CI calculations on quantum computers, good approximate [wave functions](#) close to the exact wave functions of SE under study are required. Otherwise, bad wave functions need an extreme number of steps of repeated calculations to reach the exact ones, hampering the advantages of [quantum computing](#). This problem becomes extremely serious for the analyses of chemical reactions, which have a multiconfigurational nature due to electrons not participating in chemical bonding during the bond dissociation. The OCU researchers have tackled this problem, one of the most intractable issues in quantum science and chemistry, and made a breakthrough in implementing a new quantum algorithm generating particular wave functions called configuration state functions (CSFs) in polynomial computing time.

The previously proposed algorithms for quantum computing, however, inevitably involve the dissociation and formation of many chemical bonds, and as a result, generate many electrons not participating in chemical bonds, making the quantum algorithms difficult to apply. This is termed the "Quantum Dilemma."

The OCU researchers have introduced a diradical character,  $y_i^{(0 \sim 1)}$ , to

measure and characterize the nature of open shell electronic structures, and have exploited the diradical characters to construct multiconfigurational wave functions required for chemical reactions, executing the Full-CI calculations along the whole reaction pathway on quantum computers. This new procedure requires no time-consuming post-Hartree-Fock calculations, avoiding the exponential explosion of the calculation, solving the "Quantum Dilemma" for the first time. The OCU group writes, "This is the first example of a practical quantum algorithm that makes quantum chemical calculations for predicting chemical reaction pathways realizable on quantum computers equipped with a sizable number of qubits. The implementation empowers practical applications of quantum chemical calculations on quantum computers in many important fields of chemistry and materials science."

**More information:** Kenji Sugisaki et al, Quantum Chemistry on Quantum Computers: A Method for Preparation of Multiconfigurational Wave Functions on Quantum Computers without Performing Post-Hartree–Fock Calculations, *ACS Central Science* (2018). [DOI: 10.1021/acscentsci.8b00788](https://doi.org/10.1021/acscentsci.8b00788)

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