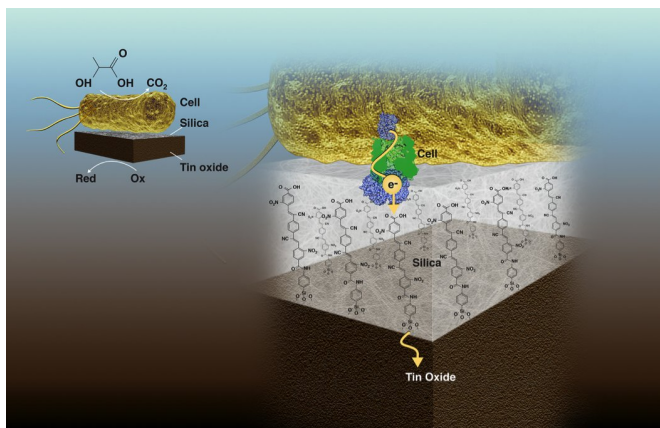


Scaling up biohybrid systems to synthesize fuels, chemicals

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Visualization of a bacterial cell (top) converting the chemical energy of organic molecules to electrons that are transferred to an inorganic tin oxide catalyst (bottom) via molecular wires embedded in an ultrathin silica layer (middle). The proton conducting silica membrane separates the chemically incompatible biological and inorganic environments thereby enabling electronic coupling of the catalysts on the shortest possible length scale, which is key to biohybrid performance and scalability. Credit: Zosia Rostomian/Berkeley Lab

Bioelectrochemical systems combine the best of both worlds – microbial cells with inorganic materials – to make fuels and other energy-rich chemicals with unrivaled efficiency. Yet technical difficulties have kept them impractical anywhere but in a lab. Now researchers at Lawrence Berkeley National Laboratory (Berkeley Lab) have developed a novel nanoscale membrane that could address these issues and pave the way for commercial scale-up.

The nanoscale membrane is embedded with molecular wires that simultaneously chemically separate, yet electrochemically couple, a microbial and an inorganic catalyst on the shortest possible length scale. This new modular architecture, described in a paper published recently in *Nature*

Communications, opens up a large design space for building scalable biohybrid electrochemical systems for a variety of applications, including electricity generation, waste remediation, and resource recovery, in addition to chemical synthesis.

The work was led by Heinz Frei, a senior scientist in Berkeley Lab's Molecular Biophysics and Integrated Bioimaging Division (MBIB), and Caroline Ajo-Franklin, a staff scientist with Berkeley Lab's Molecular Foundry who holds a secondary appointment in MBIB.

"This advance introduces a completely new architecture for bioelectrochemical systems based on nanoscale integration and provides a path forward to scaling up these systems to a commercially relevant level," said Frei. "What's more, it provides an example of how a key design principle inspired by biology is applied for solving a major scientific gap of engineered systems."

Biohybrid electrochemical systems employ separate microbial and inorganic catalysts in oxidation-reduction, or redox, reactions, to capitalize on the complementary strengths of each component. Microbes can synthesize complex molecules with high selectivity, while inorganic catalysts are the most efficient energy collectors. Such biohybrid systems are attractive as a sustainable technology to produce fuels and high-value chemicals using renewable energy.

But, a fundamental challenge in designing biohybrid systems is that the environments that support optimal function of living cells and [inorganic materials](#) are chemically incompatible, resulting in toxicity, corrosion, or efficiency-degrading cross reactions. To date, the approach has been to keep the biological and abiotic components physically separated by macroscopic (millimeter to centimeter) distances. However this exacts a high cost in terms of efficiency, due to resistance losses (on the order of 25 percent of the cell voltage)

caused by ion transport between the components, making scale-up to commercially relevant levels impractical.

In electrochemical systems, broadly speaking, an oxidation reaction at the anode and a reduction reaction at the cathode create a driving force for electrons to flow, thereby converting chemical energy into electrical energy or vice-versa. As a proof-of-concept, the researchers electrochemically coupled *Shewanella oneidensis*, an anaerobic bacterium, to an inorganic catalyst, tin dioxide (SnO₂). At 2 nanometers thick, the silica membrane enabled current flow while blocking oxygen and other small molecule transport.

This study builds on previous work by Frei's group in which they fabricated a square-inch sized artificial photosystem, in the form of an inorganic core-shell nanotube array, and by Ajo-Franklin's group in which insight at the molecular level revealed how the outer cell membrane protein interacts with an inorganic oxide surface.

More information: Jose A. Cornejo et al. Nanoscale membranes that chemically isolate and electronically wire up the abiotic/biotic interface, *Nature Communications* (2018). DOI: [10.1038/s41467-018-04707-6](https://doi.org/10.1038/s41467-018-04707-6)

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