Team performs first laboratory simulation of exoplanet atmospheric chemistry
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Lead author Sarah Hörst, right, and assistant research scientist Chao He examine samples of simulated atmospheres in a dry nitrogen glove box, where they are stored to avoid contamination from Earth's atmosphere. Credit: Will Kirk/JHU

Scientists have conducted the first lab experiments on haze formation in simulated exoplanet atmospheres, an important step for understanding upcoming observations of planets outside the solar system with the James Webb Space Telescope.

The simulations are necessary to establish models of the atmospheres of far-distant worlds, models that can be used to look for signs of life outside the solar system. Results of the studies appeared this week in *Nature Astronomy*.

"One of the reasons why we're starting to do this work is to understand if having a haze layer on these planets would make them more or less habitable," said the paper's lead author, Sarah Hörst, assistant professor of Earth and planetary sciences at the Johns Hopkins University.

With telescopes available today, planetary scientists and astronomers can learn what gases make up the atmospheres of exoplanets. "Each gas has a fingerprint that's unique to it," Hörst said. "If you measure a large enough spectral range, you can look at how all the fingerprints are superimposed on top of each other."

Current telescopes, however, do not work as well with every type of exoplanet. They fall short with exoplanets that have hazy atmospheres. Haze consists of solid particles suspended in gas, altering the way light interacts with the gas. This muting of spectral fingerprints makes measuring the gas composition more challenging.

Hörst believes this research can help the exoplanet science community determine which types of atmospheres are likely to be hazy. With haze clouding up a telescope's ability to tell scientists which gases make up an exoplanet's atmosphere - if not the amounts of them - our ability to detect life elsewhere is a murkier prospect.

Hörst uses a flashlight to look inside the experimental chamber when the experiment is running, to see if haze is being formed. Credit: Will Kirk/JHU

Planets larger than Earth and smaller than
Neptune, called super-Earths and mini-Neptunes, are the predominant types of exoplanets, or planets outside our solar system. As this class of planets is not found in our solar system, our limited knowledge makes them more difficult to study.

With the coming launch of the James Webb Space Telescope, scientists hope to be able to examine the atmospheres of these exoplanets in greater detail. JWST will be capable of looking back even further in time than Hubble with a light collecting area around 6.25 times greater. Orbiting around the sun a million miles from Earth, JWST will help researchers measure the composition of extrasolar planet atmospheres and even search for the building blocks of life.

"Part of what we're trying to help people figure out is basically where you would want to look," said Hörst of future uses of the James Webb Space Telescope.

Given that our solar system has no super-Earths or mini-Neptunes for comparison, scientists don't have "ground truths" for the atmospheres of these exoplanets. Using computer models, Hörst's team was able to put together a series of atmospheric compositions that model super-Earths or mini-Neptunes. By varying levels of three dominant gases (carbon dioxide, hydrogen, gaseous water), four other gases (helium, carbon monoxide, methane, nitrogen) and three sets of temperatures, they assembled nine different "planets."

The computer modeling proposed different percentages of gases, which the scientists mixed in a chamber and heated. Over three days, the heated mixture flowed through a plasma discharge, a setup that initiated chemical reactions within the chamber.

"The energy breaks up the gas molecules that we start with. They react with each other and make new things and sometimes they'll make a solid particle [creating haze] and sometimes they won't," Hörst said.

"The fundamental question for this paper was: Which of these gas mixtures - which of these atmospheres - will we expect to be hazy?" said Hörst.

The researchers found that all nine variants made haze in varying amounts. The surprise lay in which combinations made more. The team found the most haze particles in two of the water-dominant atmospheres. "We had this idea for a long time that methane chemistry was the one true path to make a haze, and we know that's not true now," said Hörst, referring to compounds abundant in both hydrogen and carbon.

Furthermore, the scientists found differences in the colors of the particles, which could affect how much heat is trapped by the haze. "Having a haze layer can change the temperature structure of an atmosphere," said Hörst. "It can prevent really energetic photons from reaching a surface."

Like the ozone layer that now protects life on Earth from harmful radiation, scientists have speculated a primitive haze layer may have shielded life in the very beginning. This could be meaningful in our search for external life.

For Hörst's group, the next steps involve analyzing
the different hazes to see how the color and size of the particles affect how the particles interact with light. They also plan to try other compositions, temperatures, energy sources and examine the composition of the haze produced.

"The production rates were the very, very first step of what's going to be a long process in trying to figure out which atmospheres are hazy and what the impact of the haze particles is," Hörst said.


Provided by Johns Hopkins University

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