Sea urchin protein provides insights into self-assembly of skeletal structures

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Calcium carbonate, or CaCO₃, comprises more than 4% of the earth's crust. Its most common natural forms are chalk, limestone, and marble, produced by the sedimentation of the shells of small fossilized snails, shellfish, and coral over millions of years.

New York University College of Dentistry (NYU Dentistry) researchers are studying how nature creates three-dimensional CaCO₃ inorganic/organic based materials to form seashells, invertebrate exoskeletons, and vertebrate bone, dentine, and enamel.

John Evans, DMD, PhD, a professor in NYU Dentistry's Department of Basic Science and Craniofacial Biology, oversees a research group focusing on the study of proteins that modulate the formation of biominerals, which in turn create new composite materials with unique properties, such as increased fracture and puncture resistances.

In a paper recently published in *Biochemistry*, Gaurav Jain, PhD, a postdoc in Dr. Evans's lab and coauthor of "A model sea urchin spicule matrix protein, rSpSM50, is a hydrogelator that modifies and organizes the mineralization process," looked at how the CaCO₃ matrix is organized inside a sea urchin spicule (See figure 1). At first, these spicules are nothing more than chalk, but when combined with sea urchin proteins, they form tiny stacks of "bricks," creating a structure that provides some of the toughest defense against predators and harsh ocean conditions.

"Primary mesenchyme cells (PMCs) inside a sea urchin embryo deposits amorphous CaCO₃ within the matrix of spicule proteins where these bricks are shaped into layers of calcium carbonate crystals," notes Dr. Jain. "However, the functional and assembly capabilities of individual spicule matrix proteins aren't clear. We are currently investigating one such protein found inside the spicules of a sea urchin embryo to understand what makes these proteins such efficient 'brick organizers.'"

The researchers looked at SM50, one of the most abundant and well-studied proteins found inside these spicules. They found that a recombinant version of the SM50 protein, rSpSM50, is a highly aggregation-prone protein that forms tiny jelly-like structures called hydrogels in solution. These 'jellies' capture tiny mineral nanoparticles and organize them into crystalline 'bricks.' Moreover, rSpSM50 causes surface texturing and forms randomly interconnected porous channels within these crystals.

"What is unique about rSpSM50 is that it fosters the formation and organization of two different forms of calcium carbonate—calcite and vaterite within the
'jellies' themselves, inducing fracture resistance to the overall structure," said Dr. Jain.

Researchers used a specific type of titration method that revealed the details about very early events in the spicule formation.

"rSpSM50 turns out to be a really important piece of the puzzle, as it slows the formation kinetics but neither stabilizes nor destabilizes the extremely tiny mineral particles that ultimately form these bricks," says coauthor Martin Pendola, PhD.

CaCo3 has always been a man's favorite construction material to make primitive tools, musical instruments, and craftware since the beginning of civilization. In modern times, CaCO3 is the most widely used mineral in the paper, plastics, paints and coatings industries both as a filler—and due to its special white color—as a coating pigment.

"Our current research, funded by U.S. Department of Energy, will enable scientists to better understand the mineralization and assembly process crucial to spicule formation in sea urchin," said Dr. Evans. "Our ultimate goal is to determine the molecular properties of these proteins that allow matrices to assemble, mineralize, and participate in the formation of naturally occurring organic/inorganic skeletal structures. The hope is that the comprehensive understanding of spicule proteins will enable the development of tunable fracture resistant materials that one day will find its use in developing lightweight 'armor' and 'sturdier' dental composites."


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