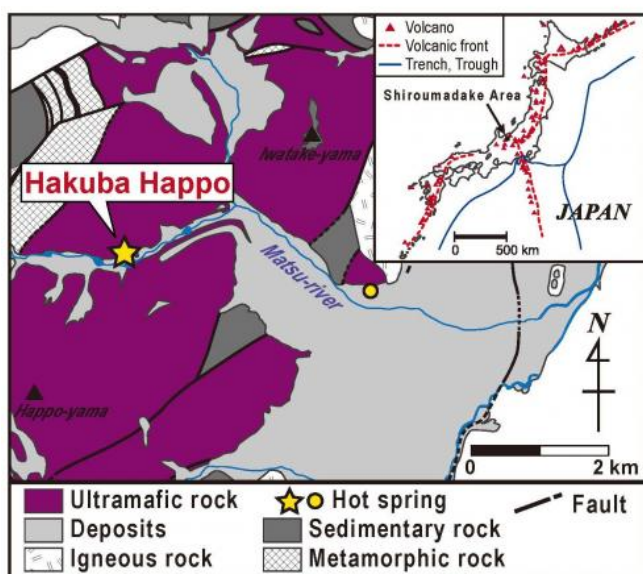


Organic chemical origins in hydrothermal systems

22 January 2014

Fig.1. The Shiroumadake area in central Japan



The researchers compared water samples from a series of [hot springs](#) in the Shiroumadake area in Japan. Here due to recent volcanic activity they could study ongoing serpentinisation processes.

They measured the pH and temperature as well as the gas and ion content of the water samples in terms of both concentration and the ratio of different isotopes of the chemical constituents. Different isotopes of the same chemical differ in the number of neutrons in the nucleus. Each reaction yield characteristic isotope ratio because reaction rate of each isotopes are slightly different depending on processes.

Suda and colleagues found unexpected values for the ratio of different isotopes in the methane (CH₄) and molecular hydrogen (H₂) dissolved in the water, and the water itself (H₂O) at the hot spring Hakuba Happono. In serpentinite-hosted hydrothermal systems methane was thought to form from synthesis with molecular hydrogen. However the researchers found that the ratio of different isotopes and chemicals could not be explained for this process in the temperature and pH conditions they had measured.

They conclude, "Based on a comparison of the hydrogen isotope systematics of our results with those of other serpentinite-hosted hydrothermal systems, we suggest that abiotic CH₄ production directly from H₂O (without mediation by H₂) may be more common in serpentinite-hosted systems."

Background

- Methane and organic compounds

Organic compounds are carbon based chemicals. The simplest [organic compounds](#) are strings of carbon atoms bonded to hydrogen. These hydrogen bonds can substitute with other atoms and molecules to provide the wide ranging organic chemicals that are found in living organisms.

Researchers at Tokyo Institute of Technology reveal mechanisms for the formation of methane, which may have been a crucial stage in the origin of life on Earth.

Serpentinite-hosted hydrothermal systems have been suggested as likely sites for the formation of organic compounds in the abiotic conditions of early Earth, that is, in the absence of living organisms. "Such compounds were likely crucial for the chemical evolution of life," explain Konomi Suda and colleagues at Tokyo Institute of Technology, Japan Agency of Marine-Earth Science and Technology and the Open University of Japan. Their latest research identifies mechanisms in the abiotic formation of the organic compound methane in serpentinite-hosted hydrothermal systems, a process that so far has not been satisfactorily understood.

Methane is the simplest organic compound comprising just one carbon atom bonded to four hydrogens. In the absence of living organisms methane synthesis can occur through abiotic mechanisms, which likely played a significant role in the early Earth environment. Possible abiotic mechanisms include the formation of methane directly from H₂O or H₂. The formation mechanism and conditions such as temperature and pH determine the relative levels of different isotopes.

- The hot spring Hakuba Happo

The researchers compared [water samples](#) from five hot springs in the Shiroumadake area in Japan. One of these sites is Hakuba Happo, a newly discovered serpentinite-hosted system. Serpentinite is a rock that results from the geochemical processes of hydration and metamorphic transformation of ultramafic rock from the Earth's mantle.

The water at Hakuba Happo is pumped up from two drilling wells Happo #1 and Happo #3. It is one of the most alkaline hot springs in Japan and the concentration of CH₄ was 10-100 times that of the other hot springs.

- Isotopic fractionation and fractionation equilibrium

Different chemical isotopes that differ by the number of neutrons in the atomic nuclei form the same chemical compounds. For example both hydrogen (no neutrons in the nucleus) and deuterium (one neutron in the nucleus) can form [molecular hydrogen](#) (H₂), water (H₂O) and methane (CH₄).

Processes described as 'fractionation' affect the relative abundance of different isotopes in the chemical compounds in a given system. Fractionation equilibrium describes the system when the abundance of isotopes in the different chemicals no longer changes with time. Comparing known fractionation equilibrium values with the measured isotopic abundance provides clues of processes that have taken place in the system.

More information: Konomi Suda, Yuichiro

Ueno, Motoko Yoshizakia, et al. "Origin of methane in serpentinite-hosted hydrothermal systems: The CH₄-H₂-H₂O hydrogen isotope systematics of the Hakuba Happo hot spring" *Earth and Planetary Science Letters*, 2014, 386 112-125.

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