Graphene-based materials kill bacteria through one of two possible mechanisms

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Graphene oxide (top) kills bacterial cells through cell-wrapping, while reduced graphene oxide (bottom) kills bacterial cells through cell-trapping. © 2012 ACS

One particularly interesting aspect of this subject is the toxic effects of graphene-based materials on the microscopic world of bacteria. For this very reason, Jun Wei at the A*STAR Singapore Institute of Manufacturing Technology and his co-workers have now compared the antibacterial activity of graphite, graphite oxide, graphene oxide and reduced graphene oxide using the model bacterium Escherichia coli. They showed that the two graphene-based materials kill substantially more bacteria than two graphite-based materials—with graphene oxide being the top performer.

Interestingly, graphene oxide particles were the smallest of all the four graphene materials as measured by dynamic light scattering. Wei and co-workers believe that particles of reduced graphene oxide were larger because they aggregated both laterally and in three dimensions.

In fact, the size of the particles could well be the key to why graphene oxide is so deadly to bacteria. When the researchers studied the affected cells using scanning electron microscopy, they saw that most of the E.coli cells were individually wrapped by layers of graphene oxide. In contrast, E. coli cells were usually embedded in the larger reduced-graphene-oxide aggregates (see image). A similar cell-trapping mechanism was operational in the graphite-based materials.

So why does cell-wrapping kill more cells than cell-trapping? The researchers believe that the direct contact of cell surface with graphene causes membrane stress and irreversible damage.

Wei and co-workers also investigated chemical mechanisms by which the materials could disrupt and kill bacteria. They found that the oxidation of glutathione, an important cellular antioxidant, occurred on exposure to graphite and reduced graphene oxide. "It might be that these structures act as conducting bridges extracting electrons from glutathione molecules and releasing them into the

The discovery of graphene has brought much excitement to the nanotechnology community. Much of this excitement is due to the possibility of deriving graphene-based materials with applications in electronics, energy storage, sensing and biomedical devices. Despite the potential, however, there is a real concern that graphene-based materials may have deleterious effects on human health and the natural environment.
external environment," says Wei.

Intriguingly, while the effect of the membrane-disrupting mechanisms dies away after four hours of incubation, the oxidation mechanism shows only minor changes.

"With the knowledge obtained in this study, we envision that physicochemical properties of graphene-based materials, such as the density of functional groups, size and conductivity can be better tailored to either reduce environmental risks or increase application potential," says Wei.


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