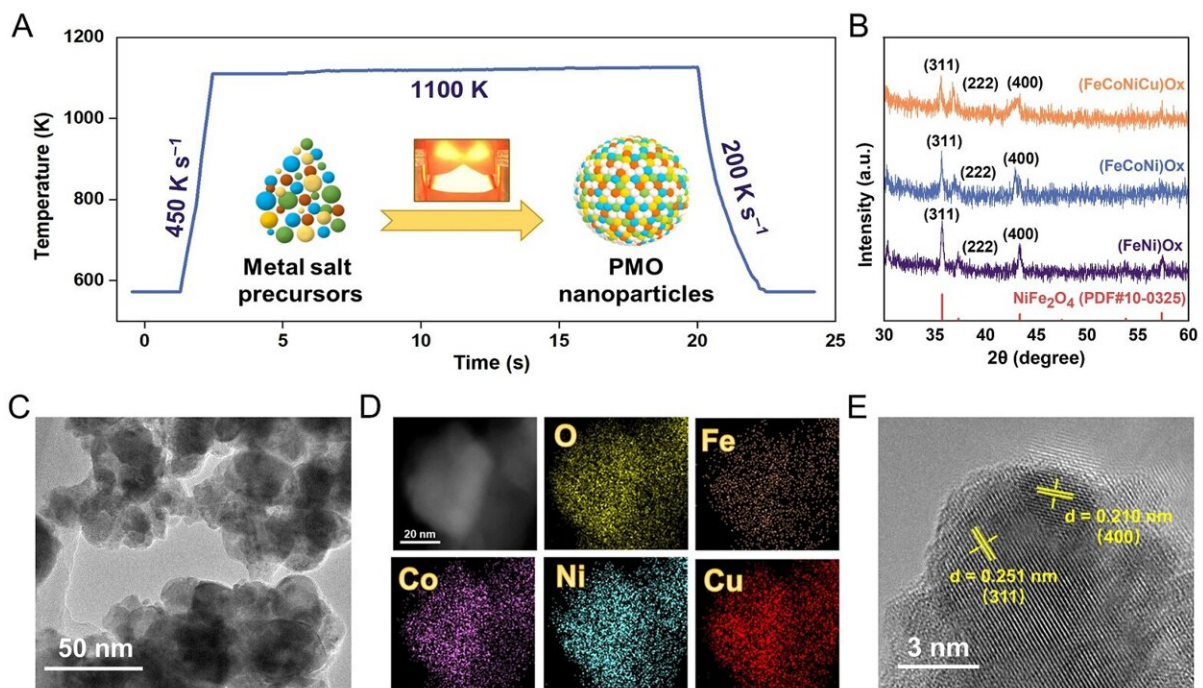


# Stud offers insights for the future design of highly efficient multi-element electrocatalysts

July 25 2024



The (FeCoNiCu)Ox electrocatalyst was first prepared using the rapid Joule-heating method within a short duration at the temperature of 1100 K. XRD results suggest that the (FeCoNiCu)Ox electrocatalyst exhibits a characteristic pattern consistent with NiFe<sub>2</sub>O<sub>4</sub>, confirming its typical spinel structure. TEM image displays that the (FeCoNiCu)Ox electrocatalyst consists of many nanoparticles in tens of nanometers in diameter. The STEM and EDS elemental mapping images verify the homogeneous distribution of Fe, Co, Ni, Cu and O elements in the nanoparticles. The HR-TEM image of the (FeCoNiCu)Ox electrocatalyst mainly indexed to NiFe<sub>2</sub>O<sub>4</sub> (311) and (400), which is in good

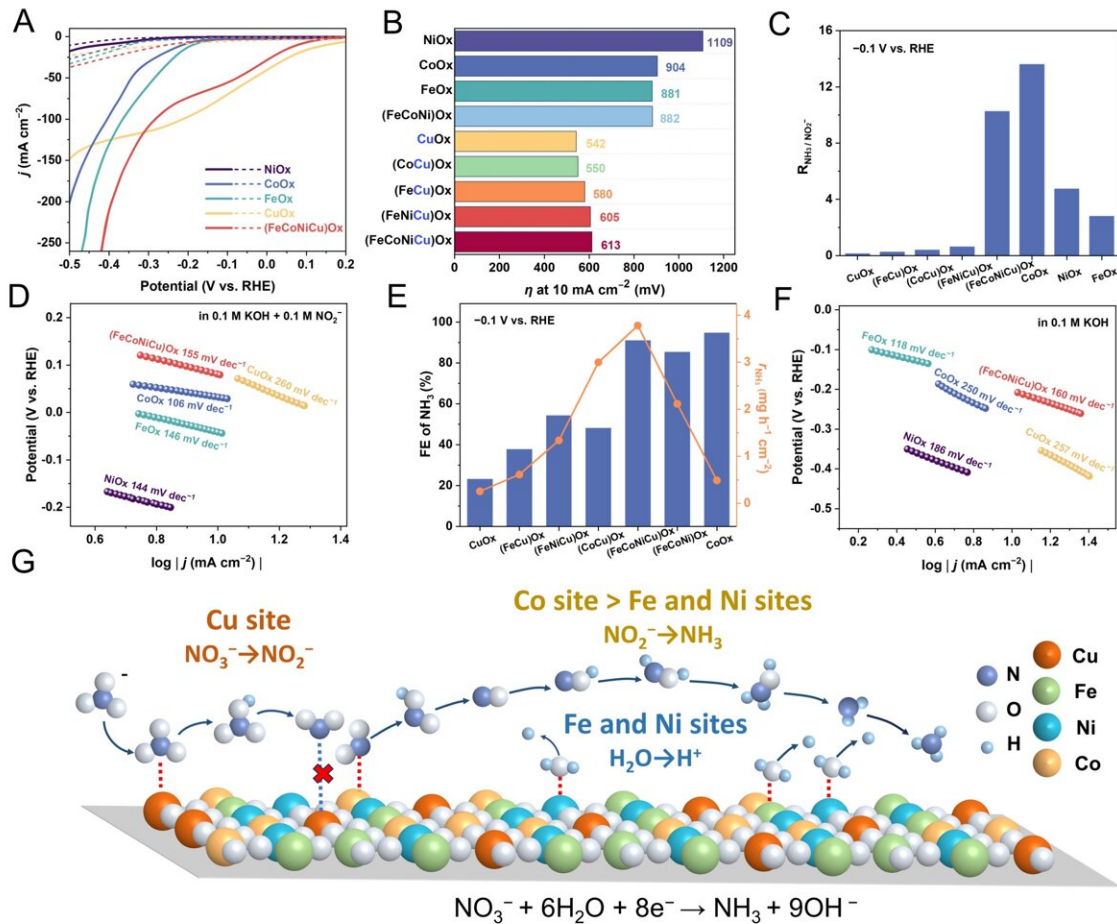
agreement with the XRD results. Credit: Science China Press

Professor Zhe Weng and Chunpeng Yang from Tianjin University [published](#) a paper titled "Unveiling multi-element synergy in polymetallic oxides for efficient nitrate reduction to ammonia" in the journal *Science China Materials*.

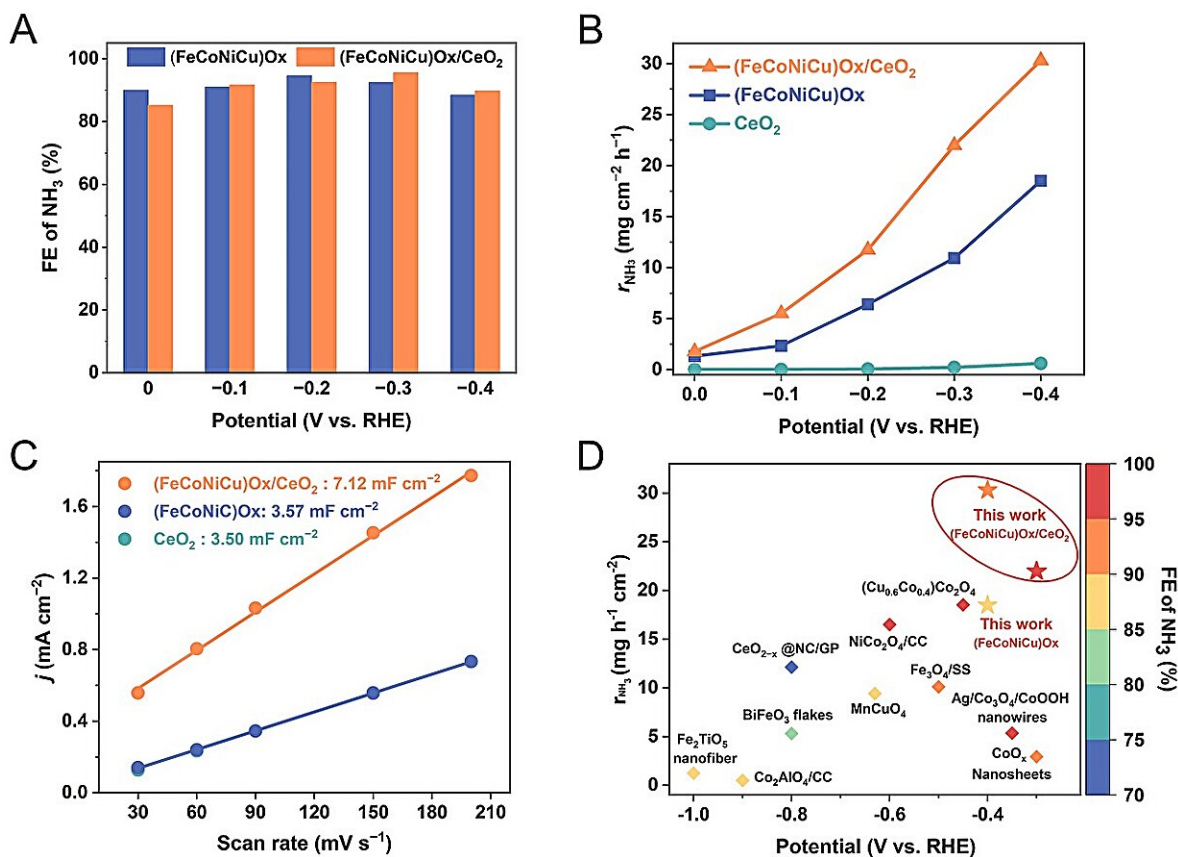
In this study, the (FeCoNiCu)O<sub>x</sub>/CeO<sub>2</sub> electrocatalyst for the reduction of NO<sub>3</sub><sup>-</sup> to NH<sub>3</sub> was prepared using the rapid Joule-heating method within a short duration. Electrochemical measurements revealed that the (FeCoNiCu)O<sub>x</sub>/CeO<sub>2</sub> electrocatalyst exhibited a high Faradaic efficiency for NH<sub>3</sub> exceeding 90% in the potential range of 0 to -0.4 V vs.

RHE, along with a high NH<sub>3</sub> yield rate of 30.3 mg h<sup>-1</sup> cm<sup>-2</sup>. Moreover, the (FeCoNiCu)O<sub>x</sub>/CeO<sub>2</sub> electrocatalyst demonstrated excellent long-term stability for more than 10 h at 200 mA cm<sup>-2</sup>.

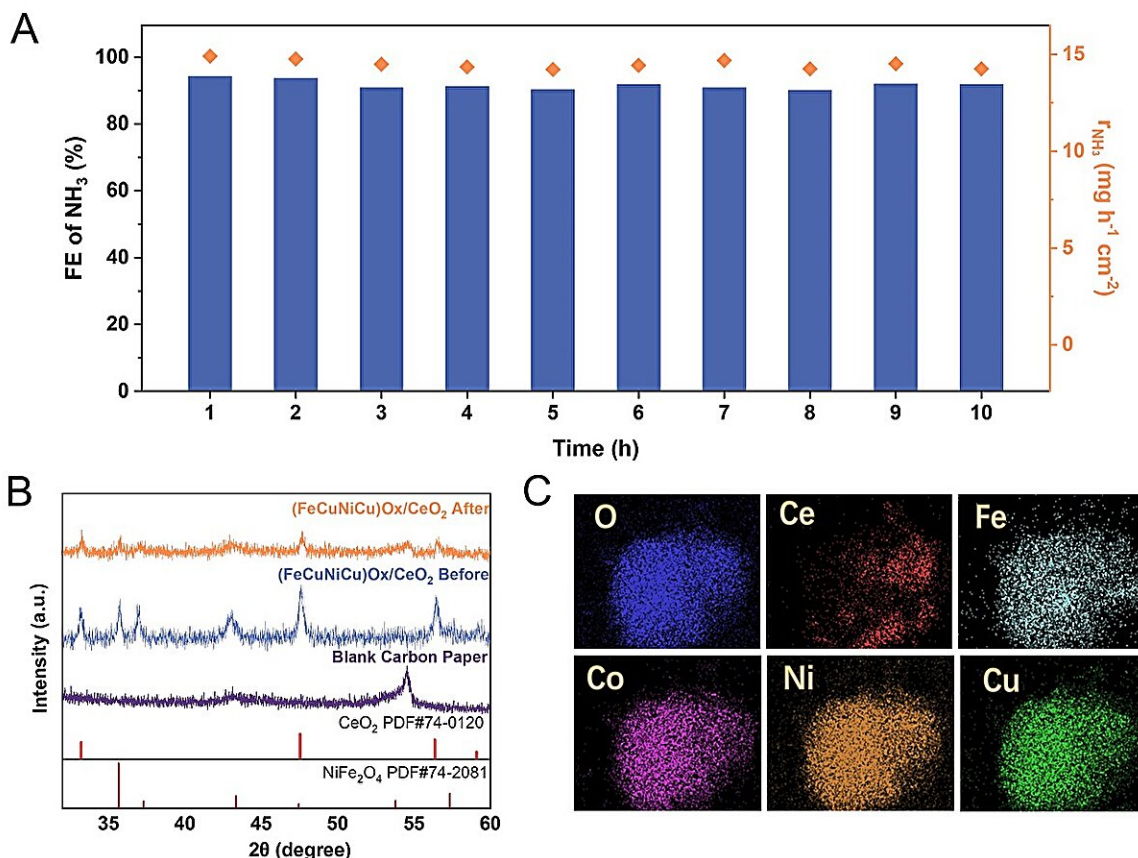
Through a series of comprehensive experiments, the individual contributions of each element and their synergistic effect have been clearly elucidated.



Several single/binary/ternary MOx electrocatalysts (M = Fe, Co, Ni, Cu, FeCo, FeNi, FeCu, CoCu, FeCoNi and FeNiCu) as control samples were also prepared using the same Joule-heating synthesis method and electrochemical tested.  
 Credit: Science China Press



The NO<sub>3</sub><sup>-</sup>RR performance of the (FeCoNiCu)Ox electrocatalyst was studied under different reduction potentials. The (FeCoNiCu)Ox electrocatalyst demonstrates a high FE<sub>NH<sub>3</sub></sub> of over 90% in the wide potential range of 0 to -0.4 V vs. RHE. However, the r<sub>NH<sub>3</sub></sub> of the (FeCoNiCu)Ox is not as competitive compared to other oxide electrocatalysts reported in previous literatures. TEM and SEM images reveal particle agglomeration, which could affect the dispersion of active components. Previous studies have reported that incorporating CeO<sub>2</sub> as a support in catalysts improves the uniform dispersion of active components and significantly increases the active surface area of the catalysts. Moreover, Ce has a much larger atomic radius than the other elements (Fe, Co, Ni and Cu) and thus Ce cannot incorporate the crystal structure of (FeCoNiCu)Ox even by the rapid Joule-heating method. Accordingly, a (FeCoNiCu)Ox/CeO<sub>2</sub> composite catalyst was further successfully synthesized. Credit: Science China Press



During the 10 h of electrocatalysis at 200 mA cm<sup>-2</sup>, the FE<sub>NH<sub>3</sub></sub> and r<sub>NH<sub>3</sub></sub> demonstrate remarkable stability. The XRD pattern reveals that the position and shape of the peaks remain largely unaltered before and after electrocatalysis. The EDS elemental mapping of the (FeCoNiCu)Ox/CeO<sub>2</sub> electrocatalyst after electrocatalysis clearly displays identical element distribution as before electrocatalysis. These results suggests that excellent stability of the (FeCoNiCu)Ox/CeO<sub>2</sub> electrocatalyst. Credit: Science China Press

Specifically, the Cu active sites efficiently reduce NO<sub>3</sub><sup>-</sup> to [nitrite](#) (NO<sub>2</sub><sup>-</sup>) at low overpotential, while the adjacent Co sites facilitate the deep reduction of intermediate NO<sub>2</sub><sup>-</sup>.

The Fe and Ni sites play a crucial role in promoting water dissociation to ensure sufficient proton supply. Simultaneously, the CeO<sub>2</sub> component increases the active surface area of the (FeCoNiCu)O<sub>x</sub> electrocatalyst and improves the NH<sub>3</sub> yield rate, making it suitable for [industrial applications](#). This work offers significant insights into the design of highly efficient multi-element electrocatalysts.

**More information:** Yaning Qie et al, Unveiling multi-element synergy in polymetallic oxides for efficient nitrate reduction to ammonia, *Science China Materials* (2024). [DOI: 10.1007/s40843-024-3017-4](https://doi.org/10.1007/s40843-024-3017-4)

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