

# Simple new process stores carbon dioxide in concrete without compromising strength

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By using a carbonated — rather than a still — water-based solution during the concrete manufacturing process, engineers discovered a new way to store carbon dioxide. Credit: Alessandro Rotta Loria/Northwestern University

By using a carbonated—rather than a still—water-based solution during the concrete manufacturing process, a Northwestern University-led team of engineers has discovered a new way to store carbon dioxide (CO<sub>2</sub>) in the ubiquitous construction material.

Not only could the new process help sequester CO<sub>2</sub> from the ever-warming atmosphere, it also results in concrete with uncompromised strength and durability.

In laboratory experiments, the process achieved a CO<sub>2</sub> sequestration efficiency of up to 45%, meaning that nearly half of the CO<sub>2</sub> injected during concrete manufacturing was captured and stored. The researchers hope their new process could help offset CO<sub>2</sub> emissions from the [cement](#) and concrete industries, which are [responsible for 8%](#) of [global greenhouse gas emissions](#).

The study was [published today](#) in *Communications Materials*, a journal published by Nature Portfolio.

"The cement and concrete industries significantly contribute to human-caused CO<sub>2</sub> emissions," said Northwestern's Alessandro Rotta Loria, who led the study. "We are trying to develop approaches that lower CO<sub>2</sub> emissions associated with those industries and, eventually, could turn cement and concrete into massive 'carbon sinks.' We are not there yet, but we now have a new method to reuse some of the CO<sub>2</sub> emitted as a result of concrete manufacturing in this very same material. And our solution is so simple technologically that it should be relatively easy for industry to implement."

"More interestingly, this approach to accelerate and accentuate the carbonation of cement-based materials provides an opportunity to engineer new clinker-based products where CO<sub>2</sub> becomes a key ingredient," said study co-author Davide Zampini, vice president of global research and development at CEMEX.

Rotta Loria is the Louis Berger Assistant Professor of Civil and Environmental Engineering at Northwestern's McCormick School of Engineering. The study was a collaboration between Rotta Loria's

laboratory and CEMEX, a global building materials company dedicated to sustainable construction.

## **Limitations of previous processes**

A non-negotiable part of infrastructure, concrete is one of the world's most consumed materials—second only to water. To make concrete in its simplest form, workers combine water, fine aggregates (like sand), coarse aggregates (like gravel) and cement, which binds all the ingredients together. Since the 1970s, previous researchers have explored various ways to store CO<sub>2</sub> inside concrete.

"The idea is that cement already reacts with CO<sub>2</sub>," Rotta Loria explained. "That's why concrete structures naturally absorb CO<sub>2</sub>. But, of course, the absorbed CO<sub>2</sub> is a small fraction of the CO<sub>2</sub> emitted from producing the cement needed to create concrete."

Processes to store CO<sub>2</sub> fall into one of two categories: hardened concrete carbonation or fresh concrete carbonation. In the hardened approach, solid concrete blocks are placed into chambers where CO<sub>2</sub> gas is injected at high pressures. In the fresh version, workers inject CO<sub>2</sub> gas into the mixture of water, cement and aggregates while concrete is being produced.

In both approaches, some of the injected CO<sub>2</sub> reacts with the cement to become solid calcium carbonate crystals. Both techniques, however, share deal-breaking limitations. They are hindered by low CO<sub>2</sub> capture efficiency and high energy consumption. Even worse: The resulting concrete is often weakened, hampering its applicability.

## **Uncompromised strength**

In Northwestern's new approach, the researchers leveraged the fresh concrete carbonation process. But, instead of injecting CO<sub>2</sub> while mixing all the ingredients together, they first injected CO<sub>2</sub> gas into water mixed with a small amount of cement powder. After mixing this carbonated suspension with the rest of the cement and aggregates, they achieved a concrete that actually absorbed CO<sub>2</sub> during its manufacturing.

"The cement suspension carbonated in our approach is a much lower viscosity fluid compared to the mix of water, cement and aggregates that is customarily employed in present approaches to carbonate fresh concrete," Rotta Loria said. "So, we can mix it very quickly and leverage a very fast kinetics of the chemical reactions that result in calcium carbonate minerals. The result is a concrete product with a significant concentration of calcium carbonate minerals compared to when CO<sub>2</sub> is injected into the fresh concrete mix."

After analyzing their carbonated concrete, Rotta Loria and his colleagues found its strength rivaled the durability of regular concrete.

"A typical limitation of carbonation approaches is that strength is often affected by the chemical reactions," he said. "But, based on our experiments, we show the strength might actually be even higher. We still need to test this further, but, at the very least, we can say that it's uncompromised. Because the strength is unchanged, the applications also don't change. It could be used in beams, slabs, columns, foundations—everything we currently use concrete for."

"The findings of this research underline that although carbonation of cement-based materials is a well-known reaction, there is still room to further optimize the CO<sub>2</sub> uptake through better understanding of the mechanisms tied to materials processing," Zampini said.

The study, "Storing CO<sub>2</sub> while strengthening concrete by carbonating its

cement in suspension," was supported by CEMEX Innovation Holding Ltd.

**More information:** Xiaoxu Fu et al, Storing CO<sub>2</sub> while strengthening concrete by carbonating its cement in suspension, *Communications Materials* (2024). [DOI: 10.1038/s43246-024-00546-9](https://doi.org/10.1038/s43246-024-00546-9)

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