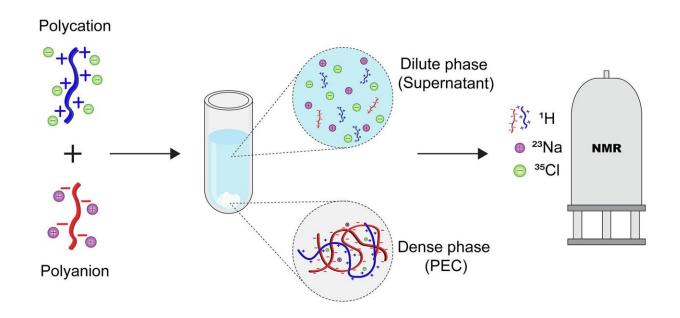


Scientists use NMR spectroscopy to map polyelectrolyte distribution and solve a century-old chemistry riddle

June 17 2024, by K. W. Wesselink-Schram



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When two solutions of oppositely charged polymers (polyelectrolytes) are mixed, phase separation occurs and leads to the formation of a polymer-rich phase and a supernatant phase. The precise distribution of all components, that is polyelectrolytes and counterions, had not yet been determined.



Researchers at the University of Twente can now measure quantitatively exactly where these components are distributed using <u>nuclear magnetic</u> <u>resonance</u> (NMR) spectroscopy.

For more than 90 years, chemists have been fascinated by polyelectrolyte complexes that form after solutions of two oppositely charged polymers are mixed. A better understanding of these complexes is crucial for the development of materials based on these complexes.

These materials include <u>porous membranes</u> for water filtration, saloplastics that are used as ion exchange membranes, and aqueousbased extraction media.

New method

To improve the applicability of these materials, scientists have developed a method to achieve a better understanding of the complexes. UT Ph.D. student and first author of the paper Giulia Allegri explains, "We introduced a new method based on NMR spectroscopy. It allows us to fully quantify the composition of the system, and measure how all the components are distributed after complexation occurs."

This study marks the first time that all components have been quantitatively determined in both phases.

Furthermore, NMR spectroscopy brings more advantages. "The protocol we developed is very fast; we can do a full analysis in less than 40 minutes," says co-author Ricardo Martinho.

Allegri adds that this method is label-free and non-invasive. She says, "Polyelectrolyte complexes are very common in nature. We envision that our methodology will contribute to a better understanding of both natural polyelectrolyte complexes and will help to improve polyelectrolyte



complex materials."

Allegri is a Ph.D. student in the Molecular Nanofabrication research group (Faculty of S&T) supervised by Dr. Saskia Lindhoud. Dr. Martinho is a researcher at the Biomolecular Nanotechnology research group (Faculty of S&T).

Together with Prof. Dr. Jurriaan Huskens, they published their findings in an <u>article</u> titled "Distribution of polyelectrolytes and counterions upon polyelectrolyte complexation," in the *Journal of Colloid and Interface Science*.

More information: Giulia Allegri et al, Distribution of polyelectrolytes and counterions upon polyelectrolyte complexation, *Journal of Colloid and Interface Science* (2024). <u>DOI:</u> 10.1016/j.jcis.2024.06.062

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