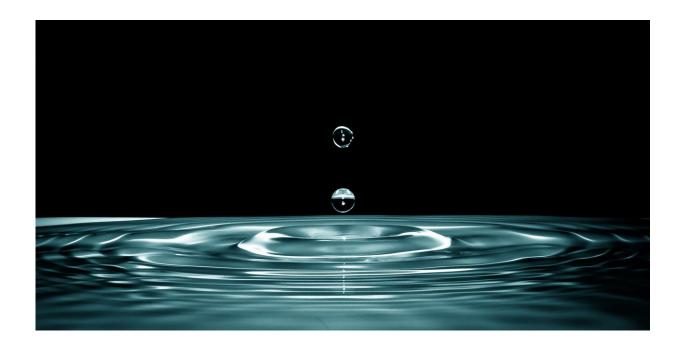


Scientists demonstrate the survival of quantum coherence in a chemical reaction involving ultracold molecules

May 16 2024



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If you zoom in on a chemical reaction to the quantum level, you'll notice that particles behave like waves that can ripple and collide. Scientists have long sought to understand quantum coherence, the ability of particles to maintain phase relationships and exist in multiple states simultaneously; this is akin to all parts of a wave being synchronized. It has been an open question whether quantum coherence can persist



through a chemical reaction where bonds dynamically break and form.

Now, for the first time, a team of Harvard scientists has demonstrated the survival of quantum coherence in a chemical reaction involving <u>ultracold molecules</u>. These findings highlight the potential of harnessing <u>chemical reactions</u> for future applications in quantum information science.

"I am extremely proud of our work investigating a very fundamental property of a chemical reaction where we really didn't know what the result would be," said senior co-author Kang-Kuen Ni, Theodore William Richards Professor of Chemistry and Professor of Physics. "It was really gratifying to do an experiment to find out what Mother Nature tells us."

In the paper, published in *Science*, the researchers detail how they studied a specific atom-exchange chemical reaction in an ultra-cold environment involving ⁴⁰K⁸⁷Rb bialkali molecules, where two potassium-rubidium (KRb) molecules react to form potassium (K₂) and rubidium (Rb₂) products.

The team prepared the initial nuclear spins in KRb molecules in an entangled state by manipulating magnetic fields and then examined the outcome with specialized tools. In the ultra-cold environment, the Ni Lab was able to track the nuclear spin degrees of freedom and observe the intricate quantum dynamics underlying the reaction process and outcome.

The work was undertaken by several members of Ni's Lab, including Yi-Xiang Liu, Lingbang Zhu, Jeshurun Luke, J.J. Arfor Houwman, Mark C. Babin, and Ming-Guang Hu.

Utilizing <u>laser cooling</u> and magnetic trapping, the team was able to cool



their molecules to just a fraction of a degree above absolute zero. In this ultracold environment, of just 500 nanoKelvin, molecules slow down, enabling scientists to isolate, manipulate, and detect individual quantum states with remarkable precision. This control facilitates the observation of quantum effects such as superposition, entanglement, and coherence, which play fundamental roles in the behavior of molecules and chemical reactions.

By employing sophisticated techniques, including coincidence detection where the researchers can pick out the exact pairs of reaction products from individual reaction events, the researchers were able to map and describe the reaction products with precision. Previously, they observed the partitioning of energy between the rotational and translational motion of the <u>product molecules to be chaotic</u>. Therefore, it is surprising to find quantum order in the form of coherence in the same underlying reaction dynamics, this time in the nuclear spin degree of freedom.

The results revealed that quantum coherence was preserved within the nuclear spin degree of freedom throughout the reaction. The survival of coherence implied that the product molecules, K_2 and Rb_2 , were in an entangled state, inheriting the entanglement from the reactants. Furthermore, by deliberately inducing decoherence in the reactants, the researchers demonstrated control over the reaction product distribution.

Going forward, Ni hopes to rigorously prove that the product molecules were entangled, and she is optimistic that <u>quantum coherence</u> can persist in non-ultracold environments.

"We believe the result is general and not necessarily limited to low temperatures and could happen in more warm and wet conditions," Ni said. "That means there is a mechanism for chemical reactions that we just didn't know about before."



First co-author and graduate student Lingbang Zhu sees the experiment as an opportunity to expand people's understanding about chemical reactions in general.

"We are probing phenomena that are possibly occurring in nature," Zhu said. "We can try to broaden our concept to other chemical reactions. Although the electronic structure of KRb might be different, the idea of interference in reactions could be generalized to other chemical systems as well."

More information: Yi-Xiang Liu et al, Quantum interference in atom-exchange reactions, *Science* (2024). <u>DOI: 10.1126/science.adl6570</u>. <u>www.science.org/doi/10.1126/science.adl6570</u>

Provided by Harvard University

Citation: Scientists demonstrate the survival of quantum coherence in a chemical reaction involving ultracold molecules (2024, May 16) retrieved 20 June 2024 from https://phys.org/news/2024-05-scientists-survival-quantum-coherence-chemical.html

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