

Hide and seek between atoms: Findings illuminate mechanism of high-nickel cathodes for lithium-ion batteries

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Schematic depiction of the structural composition of a high-nickel cathode material and findings from quantitative analysis of cation mixing on the surface of a high-nickel cathode material conducted through HAADF-STEM in conjunction with deep learning. Credit: POSTECH

Researchers have made a breakthrough in understanding the stabilization mechanism of surface structures in high-capacity, high-nickel cathode materials through single-element doping in their collaborative research through quantitative analysis. Their work was <u>published</u> in *Chemical*



Engineering Journal.

In the quest to extend the driving range of electric vehicles, there's a growing need for cathode materials with a higher capacity to store more power. Nickel (Ni) is widely used in electric vehicle batteries due to its high energy density. High-<u>nickel</u> compounds like $\text{LiNi}_{0.8}\text{Co}_{0.1}\text{Mn}_{0.1}\text{O}_2$ are common cathode materials, boasting substantial nickel content.

However, as the concentration of nickel rises, a concerning phenomenon emerges: nickel ions infiltrate the lithium layer by exchanging positions with similarly sized nickel and lithium ions along certain surfaces. This excessive cation mixing has been linked to diminished battery performance.

To address this issue, recent research has focused on incorporating <u>metal</u> <u>ions</u> as dopants. These metal cations are placed within the <u>transition</u> <u>metal</u> or lithium layers of high-nickel cathode materials. Precise the doping sites is crucial to understand their effect on the structural stability of the cathode materials. However, the small quantity of metal cations added to enhance cathode performance poses challenges in pinpointing their exact locations and studying the stabilization mechanism.

In this research, the team developed a deep learning AI technique to quantitatively analyze cation mixing using atomic structure images. They combined this approach with atomic-scale electron microscopy (HAADF-STEM), allowing them to visualize, for the first time, the location of aluminum (Al), titanium (Ti), and zirconium (Zr) metal dopants at sub-molar concentrations (mol %) in high-nickel cathode materials. Through this method, they were able to examine how these dopants affect the surface structure and electrochemical properties of the cathode material.

The examination revealed that the introduction of three metal cations



into the transition metal layer fortified the bonds between nickel and <u>oxygen atoms</u>, thereby curbing cation mixing and enhancing structural stability. Among aluminum, titanium, and zirconium, all contributed to increased discharge capacity and retention in the high-capacity nickel cathode material with titanium exhibiting the most pronounced effect. This marks the first quantitative assessment and analysis of cation mixing defects, a domain previously restricted to qualitative examination.

POSTECH Professor Si-Young Choi who led the research stated, "We developed a deep learning technology for the <u>quantitative analysis</u> of cation mixing in high-nickel cathode materials, enhancing the effectiveness of atomic-scale structural analysis."

"Our aim is to lay the groundwork for technologies that analyze highly sensitive materials, thereby advancing the understanding of performance enhancement mechanisms for next-generation <u>cathode materials</u>."

The research team includes Professor Si-Young Choi, and So-Yeon Kim and Yu-Jeong Yang, Ph.D. candidates, from the Department of Materials Science and Engineering at Pohang University of Science and Technology (POSTECH) along with Dr. Sungho Choi from the Korea Research Institute of Chemical Technology (KRICT) and Dr. Sora Lee and Chiho Jo from LG Energy Solution.

More information: So-Yeon Kim et al, Site selectivity of single dopant in high-nickel cathodes for lithium-ion batteries, *Chemical Engineering Journal* (2024). DOI: 10.1016/j.cej.2024.148869

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