

# A molecular moon lander: Insight into molecular motion on surfaces at the nanoscale

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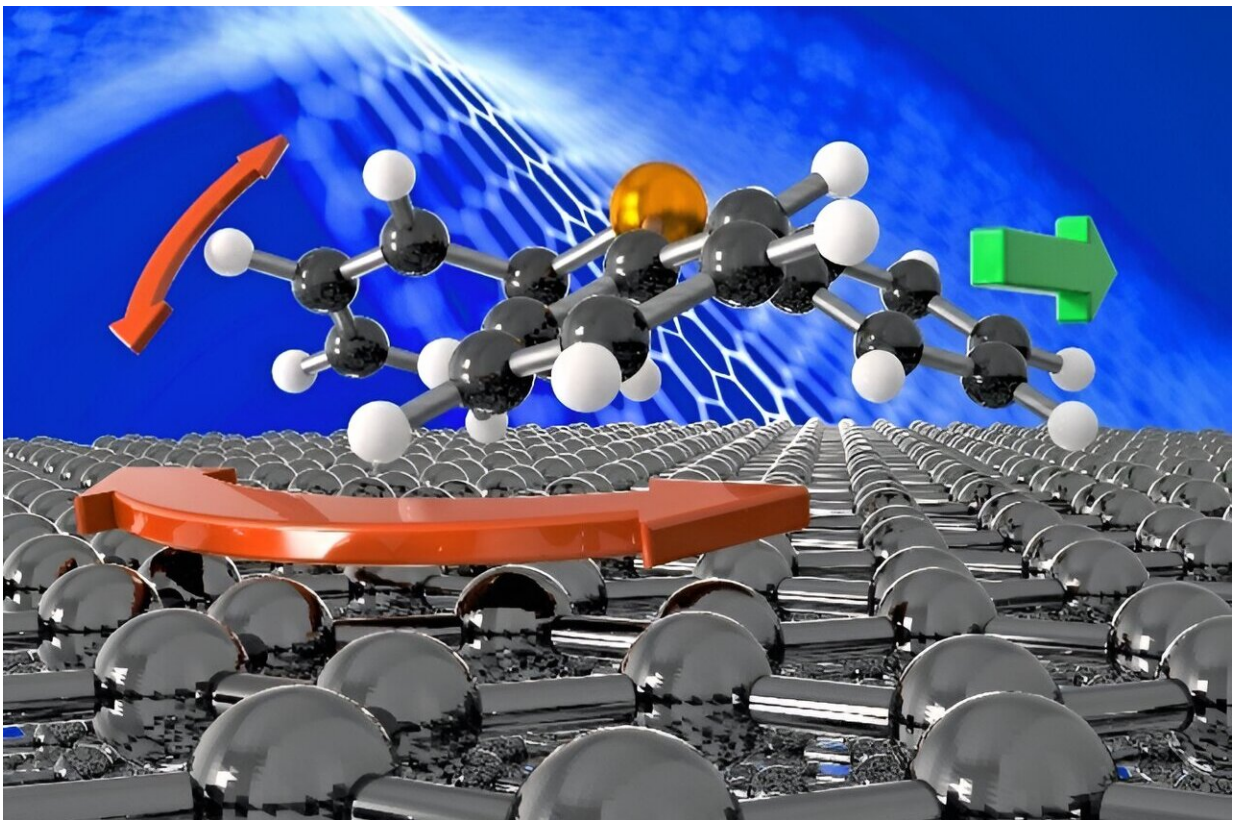


Illustration showing a single triphenylphosphine molecule over graphite. Credit: TU Graz

For years, scientists have been intrigued by how molecules move across

surfaces. The process is critical to numerous applications, including catalysis and the manufacturing of nanoscale devices.

Now, using neutron spectroscopy experiments performed at Institut Laue-Langevin (ILL) and advanced [theoretical models](#) and computer simulations, a team led by Anton Tamtögl from Graz University of Technology has unveiled the unique movement of triphenylphosphine ( $\text{PPh}_3$ ) [molecules](#) on graphite surfaces, a behavior akin to a nanoscopic moon lander.

The work is [published](#) in the journal *Communications Chemistry*.

In fact,  $\text{PPh}_3$  molecules exhibit a remarkable form of motion, rolling and translating in ways that challenge previous understandings. This moon lander-like motion seems to be facilitated by their unique geometry and three-point binding with the surface.

"Delving into the complex world of molecular motion on graphite surfaces has been an exciting journey," reveals Anton Tamtögl.

"Measurements and simulation unveiled a sophisticated motion and 'dance' of the molecules, providing us with a deeper understanding of surface dynamics and opening up new horizons for [materials science](#) and nanotechnology."

Triphenylphosphine is an important molecule for the synthesis of organic compounds and nanoparticles with numerous industrial applications. The molecule exhibits a peculiar geometry:  $\text{PPh}_3$  is pyramidal with a propeller-like arrangement of its three cyclic groups of atoms.

Neutrons offer unique possibilities in the study of materials' structure and dynamics. In a typical experiment, [neutrons](#) scattered off the sample are measured as a function of the change in their direction and energy.

Due to their low energy neutrons are an excellent probe for studying low energy excitations such as molecular rotations and diffusion. Neutron spectroscopy measurements were performed at ILL Instruments IN5 (TOF spectrometer) and IN11 (neutron spin-echo spectrometer).

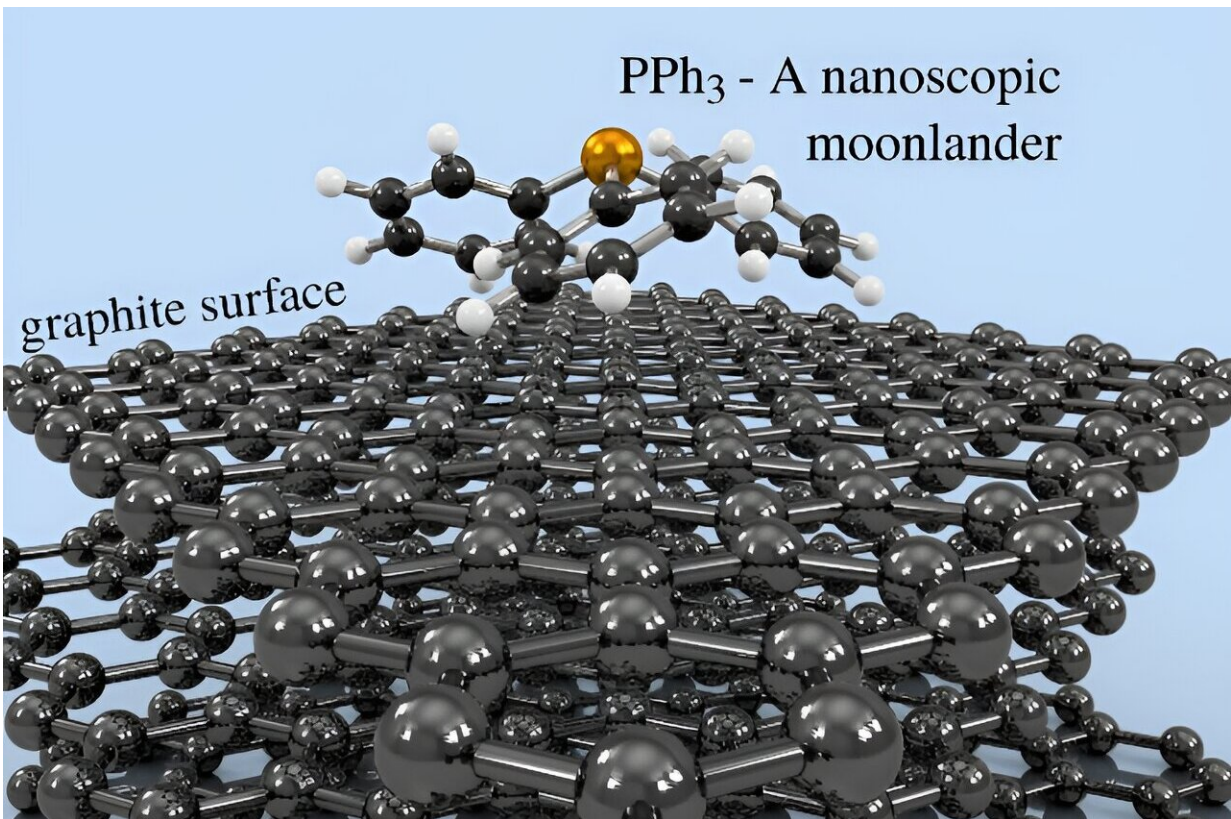


Illustration showing a single triphenylphosphine molecule over graphite. Credit: TU Graz

"It's amazing to see how ILL's powerful spectrometers allow us to follow the dynamics of these fascinating molecular systems even if the amount of sample is tiny," says ILL scientist Peter Fouquet. "Neutron beams do not destroy these sensitive samples and allow for a perfect comparison with [computer simulations](#)."

The study shows that PPh<sub>3</sub> molecules interact with the graphite surface in a manner that allows them to move with surprisingly low energy barriers. The movement is characterized by rotations and translations (jump-motions) of the molecules. While rotations and intramolecular motion dominate up to about 300 K, the molecules follow an additional translational jump-motion across the [surface](#) from 350-500 K.

Understanding the detailed mechanisms of molecular [motion](#) at the nanoscale opens up new avenues for the fabrication of advanced materials with tailored properties. Apart from the fundamental interest, the movement of PPh<sub>3</sub> and related compounds on graphite surfaces is of great importance for applications.

**More information:** Anton Tamtögl et al, Molecular motion of a nanoscopic moonlander via translations and rotations of triphenylphosphine on graphite, *Communications Chemistry* (2024). [DOI: 10.1038/s42004-024-01158-7](#)

Provided by Institut Laue-Langevin

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