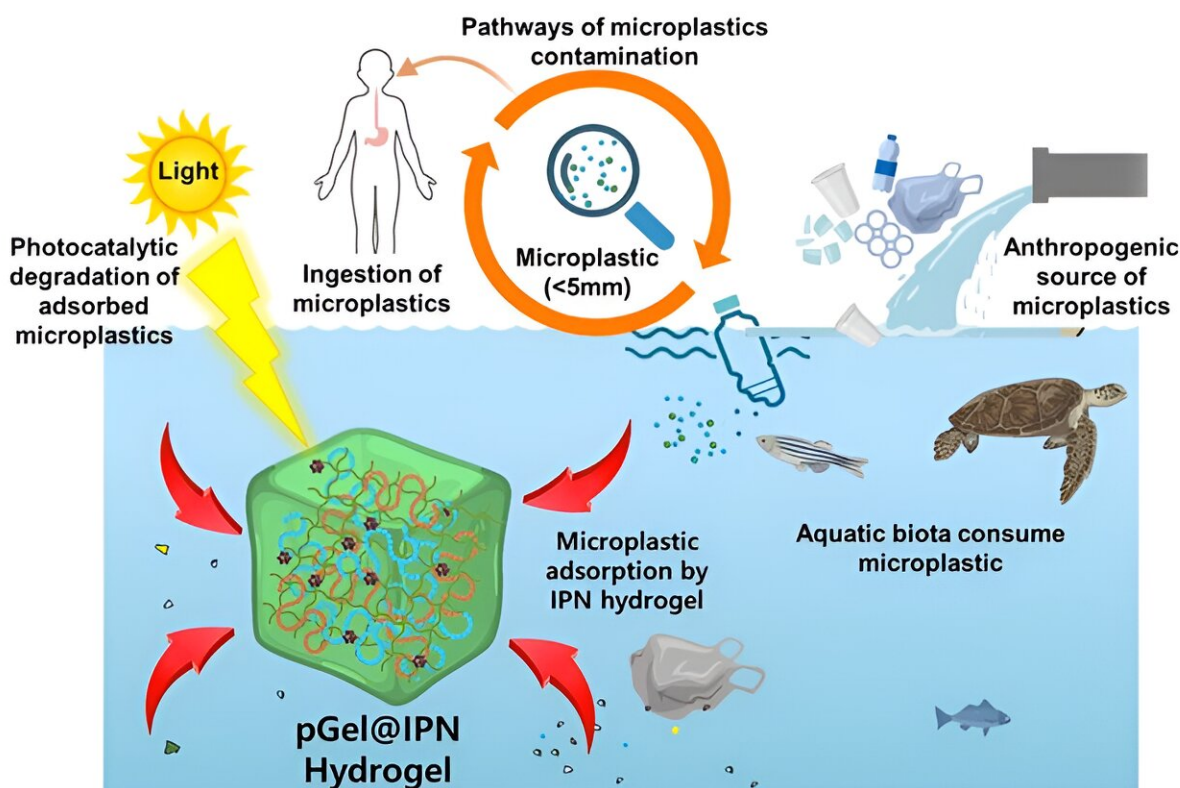


# Novel hydrogel removes microplastics from water

April 12 2024, by Shreya Gangwal



Addressing microplastic contamination in water with engineered 3D pGel@IPN hydrogel. Credit: Soumi Dutta

Microplastics pose a great threat to human health. These tiny plastic debris can enter our bodies through the water we drink and increase the

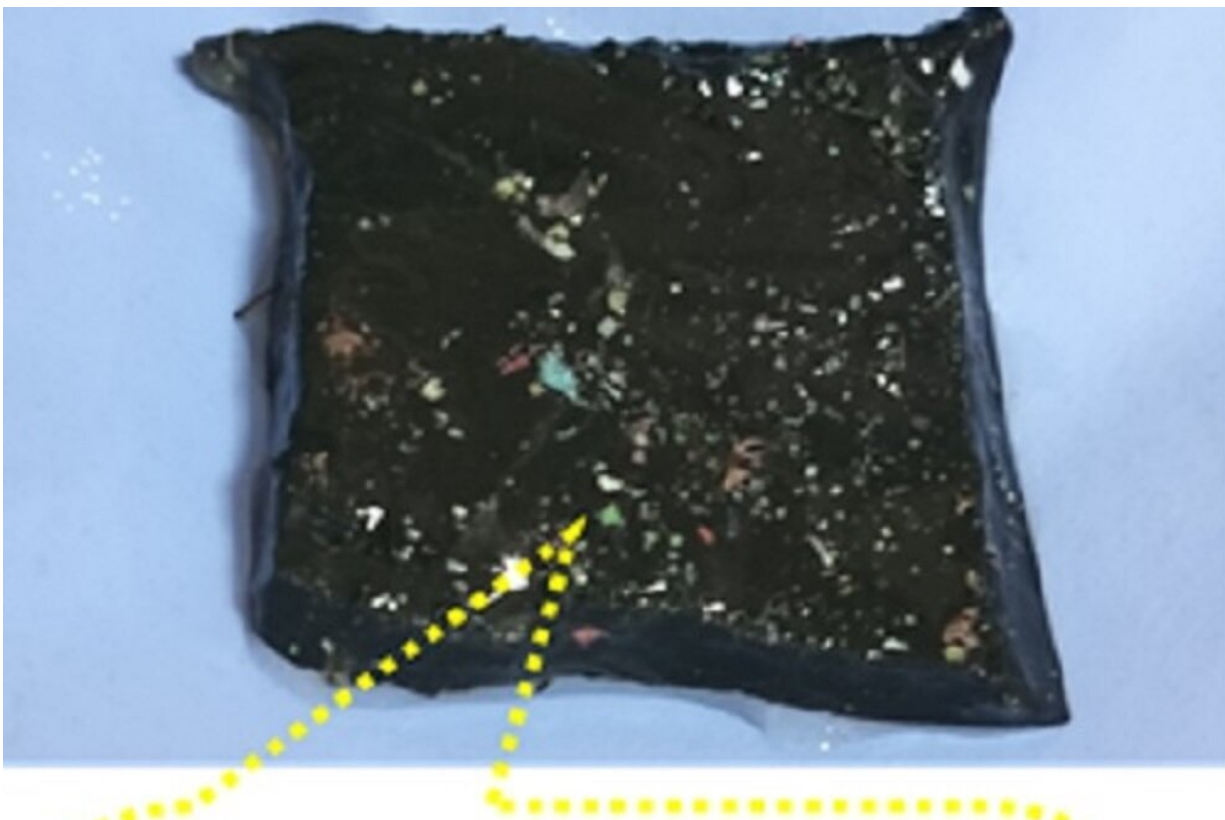
risk of illnesses. They are also an environmental hazard; found even in remote areas like polar ice caps and deep ocean trenches, they endanger aquatic and terrestrial lifeforms.

To combat this emerging pollutant, researchers at the Indian Institute of Science (IISc) have designed a sustainable hydrogel to remove microplastics from water. The material has a unique intertwined polymer network that can bind the contaminants and degrade them using UV light irradiation. The research is [published](#) in the journal *Nanoscale*.

Scientists have previously tried using filtering membranes to remove microplastics. However, the membranes can become clogged with these tiny particles, rendering them unsustainable. Instead, the IISc team led by Suryasarathi Bose, Professor at the Department of Materials Engineering, decided to turn to 3D hydrogels.

The novel hydrogel developed by the team consists of three different polymer layers—chitosan, polyvinyl alcohol and polyaniline—intertwined together, making an Interpenetrating Polymer Network (IPN) architecture. The team infused this matrix with nanoclusters of a material called copper substitute polyoxometalate (Cu-POM).

These nanoclusters are catalysts that can use UV light to degrade the microplastics. The combination of the polymers and nanoclusters resulted in a strong hydrogel with the ability to adsorb and degrade large amounts of microplastics.



Microplastic adsorption on hydrogel and fluorescent-tagged microplastic detection via spectroscopy. Credit: Soumi Dutta

Most microplastics are a product of the incomplete breakdown of household plastics and fibers. To mimic this in the lab, the team crushed food container lids and other daily-use plastic products to create two of the most common microplastics existing in nature: polyvinyl chloride and polypropylene.

"Along with treatment or removal of microplastics, another major problem is detection. Because these are very [small particles](#), you cannot see them with the naked eye," explains Soumi Dutta, first author of the study and SERB National Post-doctoral fellow at the Department of Materials Engineering.

To solve this problem, the researchers added a [fluorescent dye](#) to the microplastics to track how much was being adsorbed and degraded by the hydrogel under different conditions. "We checked the removal of microplastics at different pH levels of water, different temperatures, and different concentrations of microplastics," explains Dutta.

The hydrogel was found to be highly efficient—it could remove about 95% and 93% of the two different types of microplastics in water at near-neutral pH ( $\sim 6.5$ ). The team also carried out several experiments to test how durable and strong the material was. They found that the combination of the three polymers made it stable under various temperatures.



Suryasarathi Bose and Soumi Dutta, Department of Materials Engineering, IISc.  
Credit: Soumi Dutta

"We wanted to make a material that is more sustainable and can be used repetitively," explains Bose. The hydrogel could last for up to five cycles of [microplastic](#) removal without significant loss of efficacy. What's more, Bose points out, is that once it has outlived its use, the [hydrogel](#) can be repurposed into carbon nanomaterials that can remove [heavy metals](#) like hexavalent chromium from polluted water.

Moving forward, the researchers plan to work with collaborators to develop a device that can be deployed on a large scale to help clean up microplastics from various water sources.

**More information:** Soumi Dutta et al, Polyoxometalate nanocluster-infused triple IPN hydrogels for excellent microplastic removal from contaminated water: detection, photodegradation, and upcycling, *Nanoscale* (2024). [DOI: 10.1039/D3NR06115A](https://doi.org/10.1039/D3NR06115A)

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