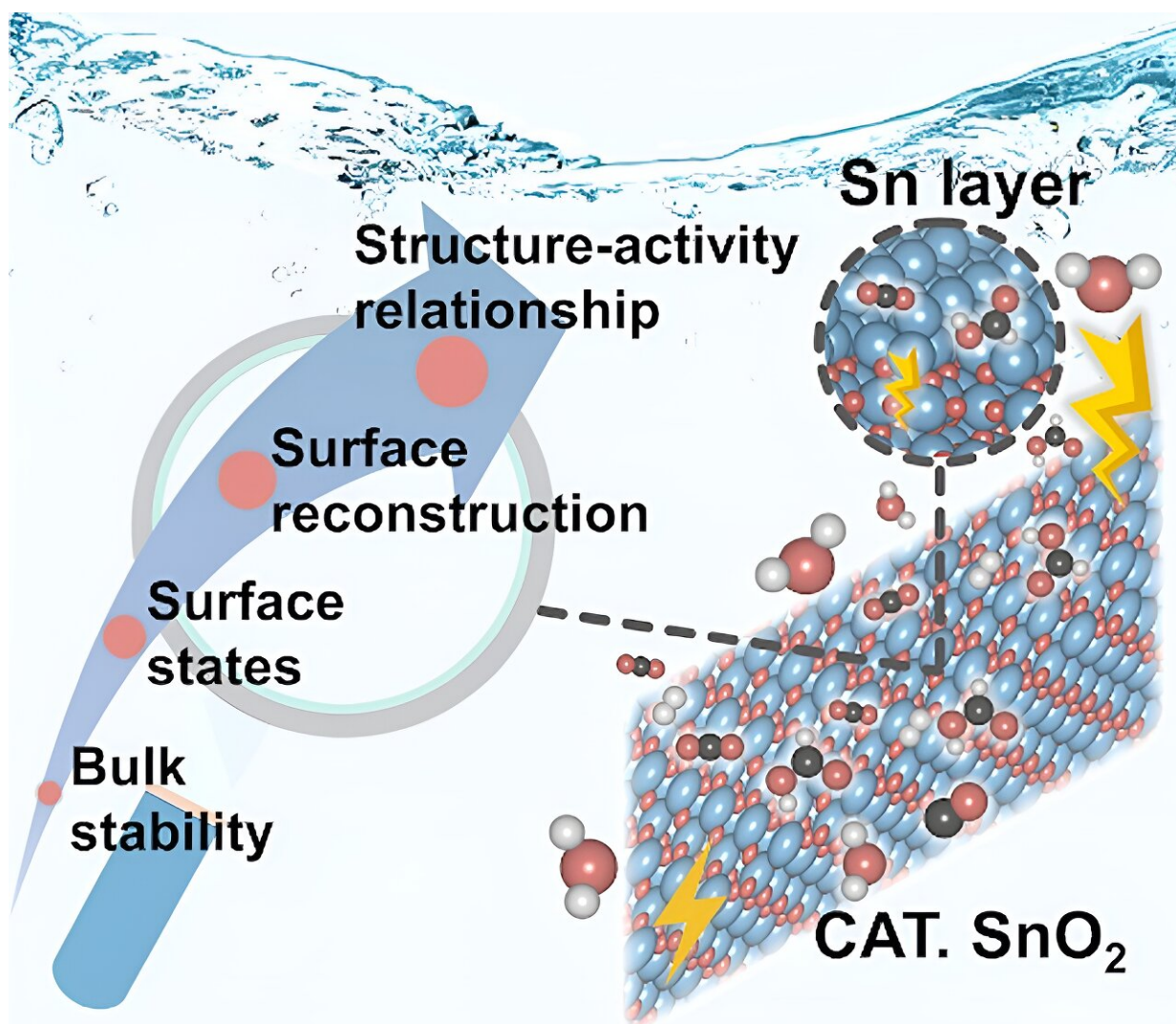


# Deciphering catalysts: Unveiling structure-activity correlations

March 7 2024



The standard research paradigm uncovers the structure-property-activity relationships for the electrochemical CO<sub>2</sub> reduction reaction (CO<sub>2</sub>RR) over SnO<sub>2</sub>. This picture illustrates the surface reconstruction induced by oxygen

vacancies (1/1 ML coverage) and surface-active species (Sn layer) accountable for selective HCOOH production. Credit: *Angewandte Chemie International Edition* (2024). DOI: 10.1002/anie.202319913

In a new step toward combating climate change and transitioning to sustainable solutions, a group of researchers has developed a research paradigm that makes it easier to decipher the relationship between catalyst structures and their reactions.

Details of the researchers' breakthrough were [published](#) in the journal *Angewandte Chemie International Edition* on January 29, 2024.

Understanding how a [catalyst](#)'s surface affects its activity can aid the design of efficient catalyst structures for specific reactivity requirements. However, grasping the mechanisms behind this relationship is no straightforward task given the complicated interface microenvironment of electrocatalysts.

"To decipher this, we honed in on the electrochemical CO<sub>2</sub> reduction reaction (CO<sub>2</sub>RR) in tin-oxide-based (Sn–O) catalysts," points out Hao Li, associate professor at Tohoku University's Advanced Institute for Materials Research (WPI-AIMR) and corresponding author of the paper. "In doing so, we not only uncovered the active surface species of SnO<sub>2</sub>-based catalysts during CO<sub>2</sub>RR but also established a clear correlation between surface speciation and CO<sub>2</sub>RR performance."

CO<sub>2</sub>RR is recognized as a promising method for reducing CO<sub>2</sub> emissions and producing high-value fuels, with [formic acid](#) (HCOOH) being a noteworthy product because of its various applications in industries such as pharmaceuticals, metallurgy, and environmental remediation.

The proposed method helped identify the genuine surface states of SnO<sub>2</sub> responsible for its performance in CO<sub>2</sub> reduction reactions under specific electrocatalytic conditions. Moreover, the team corroborated their findings through experiments using various SnO<sub>2</sub> shapes and advanced characterization techniques.

Li and his colleagues developed their methodology by combining theoretical studies with experimental electrochemical techniques.

"We bridged the gap between the theoretical and experimental, offering a comprehensive understanding of catalyst behavior under real-world conditions in the process," adds Li.

The research team is now focused on applying this methodology to a variety of electrochemical reactions. In doing those, they hope to uncover more about unique structure-activity correlations, accelerating the design of high-performance and scalable electrocatalysts.

**More information:** Zhongyuan Guo et al, Deciphering Structure-Activity Relationship Towards CO<sub>2</sub> Electroreduction over SnO<sub>2</sub> by A Standard Research Paradigm, *Angewandte Chemie International Edition* (2024). [DOI: 10.1002/anie.202319913](https://doi.org/10.1002/anie.202319913)

Provided by Tohoku University

Citation: Deciphering catalysts: Unveiling structure-activity correlations (2024, March 7) retrieved 28 April 2024 from <https://phys.org/news/2024-03-deciphering-catalysts-unveiling.html>

This document is subject to copyright. Apart from any fair dealing for the purpose of private study or research, no part may be reproduced without the written permission. The content is provided for information purposes only.