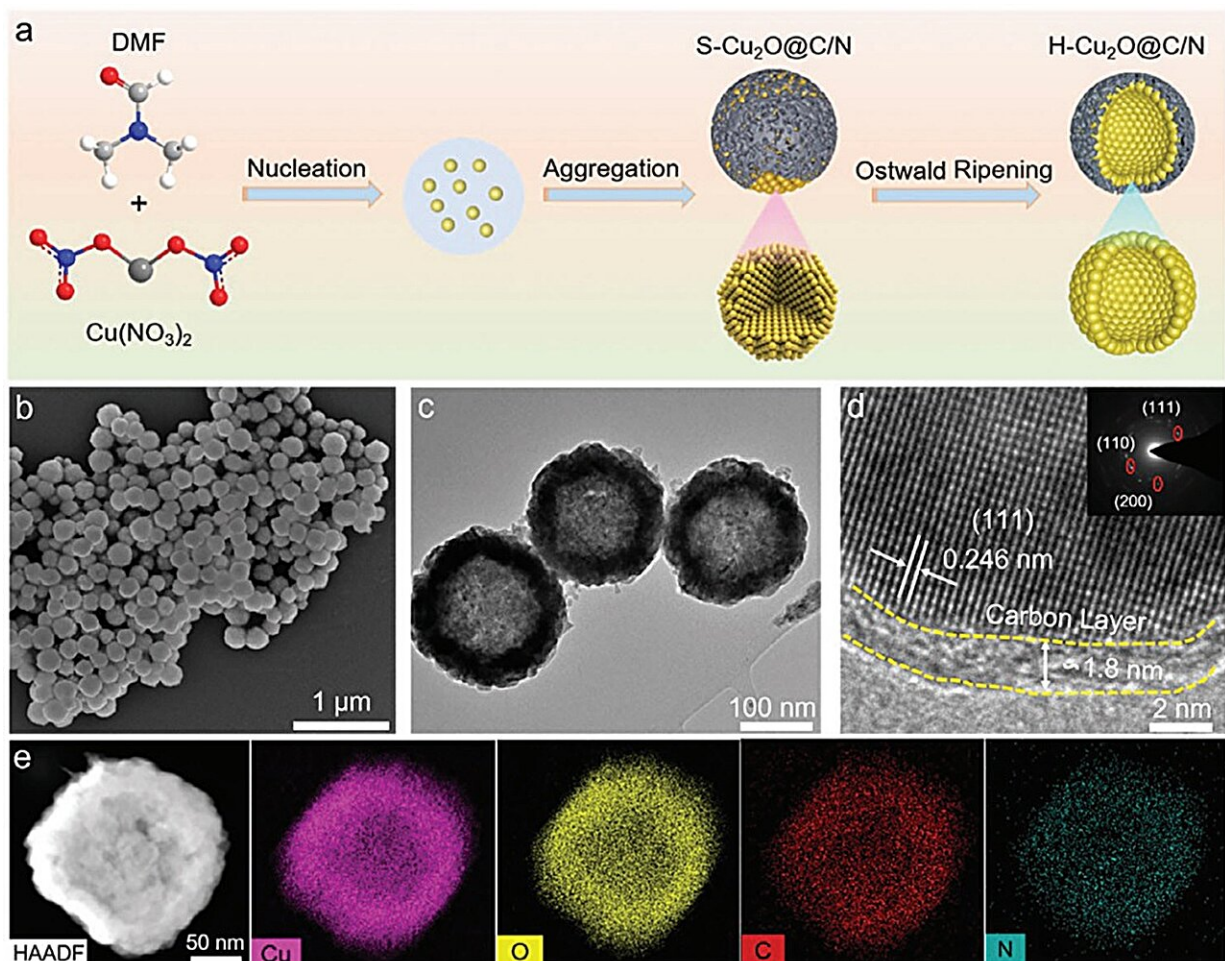


# Novel catalyst strategy offers solution for efficient CO<sub>2</sub> reduction reaction

March 20 2024, by Zhang Nannan



Synthesis schematic and characterization of H-Cu<sub>2</sub>O@C/N. a. Schematic of the synthetic process for H-Cu<sub>2</sub>O@C/N; b. SEM image; c. TEM image; d. HRTEM image, and the corresponding selective area electron diffraction pattern (inset) of H-Cu<sub>2</sub>O@C/N. Credit: Wang Hui

With the help of nuclear magnetic resonance spectrometer, a research group led by Prof. Wang Hui from the Hefei Institutes of Physical Science of the Chinese Academy of Sciences has prepared a carbon-covered hollow cuprous oxide high-efficiency catalyst by using the solvent autocarbonylation reduction strategy, which provided a new solution for the electrocatalytic carbon dioxide reduction reaction (CO<sub>2</sub>RR) in the preparation of multicarbon (C<sub>2+</sub>) products.

The results were published in [\*Advanced Functional Materials\*](#).

Excessive carbon dioxide emissions are a global problem. Converting CO<sub>2</sub> into chemicals and fuels through CO<sub>2</sub>RR not only helps the environment but also supports China's "dual-carbon" target. Progress has been made in producing single carbon (C1) products such as [carbon monoxide](#) and [formic acid](#) from CO<sub>2</sub>RR. However, current CO<sub>2</sub>RR efficiency in producing C<sub>2+</sub> products is low, creating a need for catalysts that can improve efficiency and selectivity.

In this study, the researchers developed a specialized nanoreactor called nitrogen-doped carbon shell-protected hollow cuprous oxide (H-Cu<sub>2</sub>O@C/N) using a solvent autocarbonation reduction strategy.

This nanoreactor enhancement helps increase the amount of key intermediates (\*CO) on the [catalyst](#) surface, which accelerates the production of C<sub>2+</sub> products through a chemical reaction.

When tested in a membrane electrode assembly (MEA) electrolyzer, the H-Cu<sub>2</sub>O@C/N nanoreactor achieved impressive results, with a 75.9% efficiency in producing C<sub>2+</sub> products and a high current density of 248.8 mA·cm<sup>-2</sup>. This demonstrates the effectiveness of the catalysts in CO<sub>2</sub>RR conversion.

To further understand this process, the research team conducted [detailed](#)

[studies](#). These results confirmed that the C/N inclusions prepared by solvent autocarbon reduction strategy can effectively protect the  $\text{Cu}^+$  active species and ensure their catalytic stability.

This work provides an efficient and feasible way to optimize the catalyst structure for highly selective  $\text{CO}_2\text{RR}$  preparation of  $\text{C}_{2+}$  products.

**More information:** Xiangfu Meng et al, Steering C–C Coupling by Hollow  $\text{Cu}_2\text{O}@C/N$  Nanoreactors for Highly Efficient Electroreduction of  $\text{CO}_2$  to  $\text{C}_2^+$  Products, *Advanced Functional Materials* (2024). [DOI: 10.1002/adfm.202312719](https://doi.org/10.1002/adfm.202312719)

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