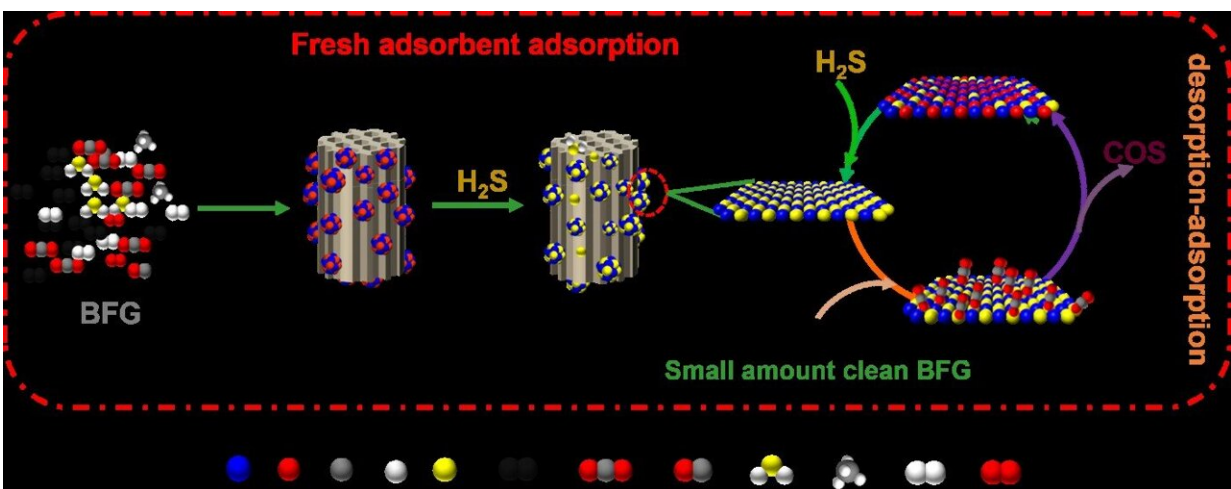


Ammonia induction strategy for preparation of transition metal oxides / zeolite H₂S adsorbent

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H₂S removal and regeneration mechanism of NH₃-CuO/13X-3 adsorbent.
Credit: Erping Cao, et al.

Blast furnace gas (BFG) is an important by-product energy for the iron and steel industry and has been widely used for heating and electricity generation. However, the undesirable contaminants, such as COS, CS₂, and H₂S, in BFG generate harmful environmental emissions.

Due to the stringent ultra-low emission standards, the desulfurization of BFG is urgent for integrated steel plants. Compared with other

desulfurization materials, zeolite-based adsorbents represent a viable option with low costs and long service life. However, the sulfur capacity of zeolite is relatively low and needs to improve.

Impregnation of transition metal oxides onto zeolite is a common strategy to prepare H₂S adsorbent. However, this method usually results in the agglomeration of metal particles during calcination, forming relatively large metal particles. The large metal particle may increase the gas diffusion resistance in adsorbent and inhibit the desulfurization performance. Therefore, minimizing the metal particles on zeolite with a high loading is the key to the preparation of an adsorbent with high sulfur capacity.

To this end, a team of researchers from the Institute of Process Engineering, Chinese Academy of Sciences, has proposed an [ammonia](#) induction strategy. In the process of loading copper oxide onto 13X zeolite by the impregnation method, ammonia was introduced, and a Cu-based complex formed first and then adsorbed on zeolite, which was converted to CuO in the subsequent calcination process.

"The introduction of ammonia effectively inhibits the agglomeration and increases the dispersibility of CuO particles during calcination, prevents the plugging of zeolite pores, improves the diffusion of H₂S during desulfurization, and thus increases the adsorption rate and sulfur capacity of H₂S adsorbent," said Erping Cao, lead author of the study [published](#) in *Green Energy and Environment*.

"The H₂S adsorption capacity of NH₃-CuO/13X adsorbent prepared by ammonia induction is more than twice that of CuO/13X adsorbent."

Notably, similar results were obtained when the ammonia induction strategy was applied to the other kind of zeolite-based adsorbents.

"Based on the ammonia induction strategy, we have provided a general approach for the preparation of transition metal oxide/[zeolite](#) adsorbents with high sulfur capacity," said corresponding author Yanbin Cui.

More information: Erping Cao et al, Ammonia-induced CuO/13X for H₂S removal from simulated blast furnace gas at low temperature, *Green Energy & Environment* (2024). [DOI: 10.1016/j.gee.2024.02.002](https://doi.org/10.1016/j.gee.2024.02.002)

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