

Clutch-stack-driven molecular gears in crystals could propel material innovation

January 19 2024



(Left) Schematic of the clutch stack arrangement. (Middle) Structure of the molecular gears in the clutch stack. (Right) Rotational direction of two adjacent molecular gears. Credit: Mingoo Jin, et al. *Journal of the American Chemical Society*. December 7, 2023

Gears are an essential component of everyday machines. The ability to shift gears, like in a car, allows for control of the degree or direction of motion generated, making machines more versatile.

Now, a team led by researchers at the Institute for Chemical Reaction



Design and Discovery (WPI-ICReDD) at Hokkaido University has reported a new design strategy for realizing molecular-sized gears in <u>crystals</u> and the first example of controllable molecular <u>gear</u> shifting in a <u>solid material</u>.

They developed a crystalline material that contains gear-like molecules that can be reversibly shifted between two types of motion. The design principle provides a blueprint for the development of versatile, new materials. The study is <u>published</u> in the *Journal of the American Chemical Society*.

Researchers utilized a gear-shaped molecule called triaryltriazine, which has a center triazine ring with three phenylene rings attached to it—which act like the teeth of a gear. By attaching bulky, stationary molecules to the phenylene rings, researchers induced a "clutch stack" arrangement, where adjacent triaryltriazine molecules are rotated 60° from each other, rather than stacking in the same orientation.

"The design of the clutch stack was inspired by the mechanical machinery system of the clutch in a car," said Associate Professor Mingoo Jin.

The attached stationary molecules also created enough space for the three phenylene rings to rotate between two positions in a flapping motion. The clutch stack arrangement of the triaryltriazine molecules enabled adjacent molecules to hook onto each other as the phenylene rings rotated, much like interlocking gears. This resulted in the correlated motion of all the molecules in the stack.

When the temperature was raised above a certain threshold, a different correlated motion was observed, in which phenylene rings underwent a 180° rotation. This change in motion was attributed to a phase transition in the crystal that created more space between adjacent <u>molecules</u>,



giving the phenylene rings more room to rotate.



(a) Structure of the molecular gears in the clutch stack. (b) Side view of molecular gears showing the two modes of motion observed at lower temperatures (left) and higher temperatures (right). Credit: Mingoo Jin, et al. Journal of the American Chemical Society. December 7, 2023

Researchers found this change in motion could be reversed by cooling the crystal, marking the first time such controllable molecular motion has been observed in a solid. The effect of the molecular gearshift could be fine-tuned by adjusting the size and structure of the stationary molecule attached to the gear molecule. This adjustability opens the door to the development of new functional materials that leverage crystalline molecular machines.

"The next direction for our research would be using geared molecular <u>motion</u> in crystals to manipulate different physical properties of solid-state materials, such as light emission or thermal behavior," stated Jin.



More information: Mingoo Jin et al, A Steric-Repulsion-Driven Clutch Stack of Triaryltriazines: Correlated Molecular Rotations and a Thermoresponsive Gearshift in the Crystalline Solid, *Journal of the American Chemical Society* (2023). DOI: 10.1021/jacs.3c08909

Provided by Hokkaido University

Citation: Clutch-stack-driven molecular gears in crystals could propel material innovation (2024, January 19) retrieved 27 April 2024 from <u>https://phys.org/news/2024-01-clutch-stack-driven-molecular-gears.html</u>

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