

Research team develops polymers that can kill bacteria

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Escherichia coli. Credit: Rocky Mountain Laboratories, NIAID, NIH

Antibiotic-resistant bacteria have become a rapidly growing threat to public health. Each year, they account for more than 2.8 million infections, according to the U.S. Centers for Disease Control and

Prevention. Without new antibiotics, even common injuries and infections harbor the potential to become lethal.

Scientists are now one step closer to eliminating that threat, thanks to a Texas A&M University-led collaboration that has developed a new family of polymers capable of killing [bacteria](#) without inducing antibiotic resistance by disrupting the membrane of these microorganisms.

"The new polymers we synthesized could help fight [antibiotic resistance](#) in the future by providing antibacterial molecules that operate through a mechanism against which bacteria do not seem to develop resistance," said Dr. Quentin Michaudel, an assistant professor in the Department of Chemistry and lead investigator in the research, [published](#) Dec. 11 in the *Proceedings of the National Academy of Sciences*.

Working at the interface of organic chemistry and [polymer](#) science, the Michaudel Laboratory was able to synthesize the new polymer by carefully designing a positively charged molecule that can be stitched many times to form a large molecule made of the same repeating charged motif using a carefully selected catalyst called AquaMet.

According to Michaudel, that catalyst proves key, given that it has to tolerate a high concentration of charges and also be water-soluble—a feature he describes as uncommon for this type of process.

After achieving success, the Michaudel Lab put its polymers to the test against two main types of [antibiotic-resistant bacteria](#)—E. coli and Staphylococcus aureus (MRSA)—in collaboration with Dr. Jessica Schiffman's group at the University of Massachusetts Amherst. While awaiting those results, the researchers also tested their polymers' toxicity against human red blood cells.

"A common issue with antibacterial polymers is a lack of selectivity between bacteria and human cells when targeting the cellular membrane," Michaudel explained. "The key is to strike a right balance between effectively inhibiting bacteria growth and killing several types of cells indiscriminately."

Michaudel credits the multidisciplinary nature of scientific innovation and the generosity of dedicated researchers across the Texas A&M campus and country as factors in his team's success in determining the perfect catalyst for their molecule assembly.

"This project was several years in the making and would not have been possible without the help of several groups, in addition to our UMass collaborators," Michaudel said.

"For instance, we had to ship some samples to the Letteri Lab at the University of Virginia to determine the length of our polymers, which required the use of an instrument that few labs in the country have. We are also tremendously grateful to [biochemistry Ph.D. candidate] Nathan Williams and Dr. Jean-Philippe Pellois here at Texas A&M, who provided their expertise in our assessment of toxicity against red blood cells."

Michaudel says the team will now focus on improving the activity of its polymers against bacteria—specifically, their selectivity for bacterial cells versus human cells—before moving on to in vivo assays.

"We are in the process of synthesizing a variety of analogs with that exciting goal in mind," he said.

The team's paper features Michaudel Lab member and Texas A&M chemistry Ph.D. graduate Dr. Sarah Hancock as first author. Other key contributors from the Michaudel Lab are chemistry graduate student An

Tran, postdoctoral scholar Dr. Arunava Maity and former postdoctoral scholar Dr. Nattawut Yuntawattana, who is now an assistant professor of materials science at Kasetsart University in Thailand.

More information: Sarah N. Hancock et al, Ring-opening metathesis polymerization of N -methylpyridinium-fused norbornenes to access antibacterial main-chain cationic polymers, *Proceedings of the National Academy of Sciences* (2023). [DOI: 10.1073/pnas.2311396120](https://doi.org/10.1073/pnas.2311396120)

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