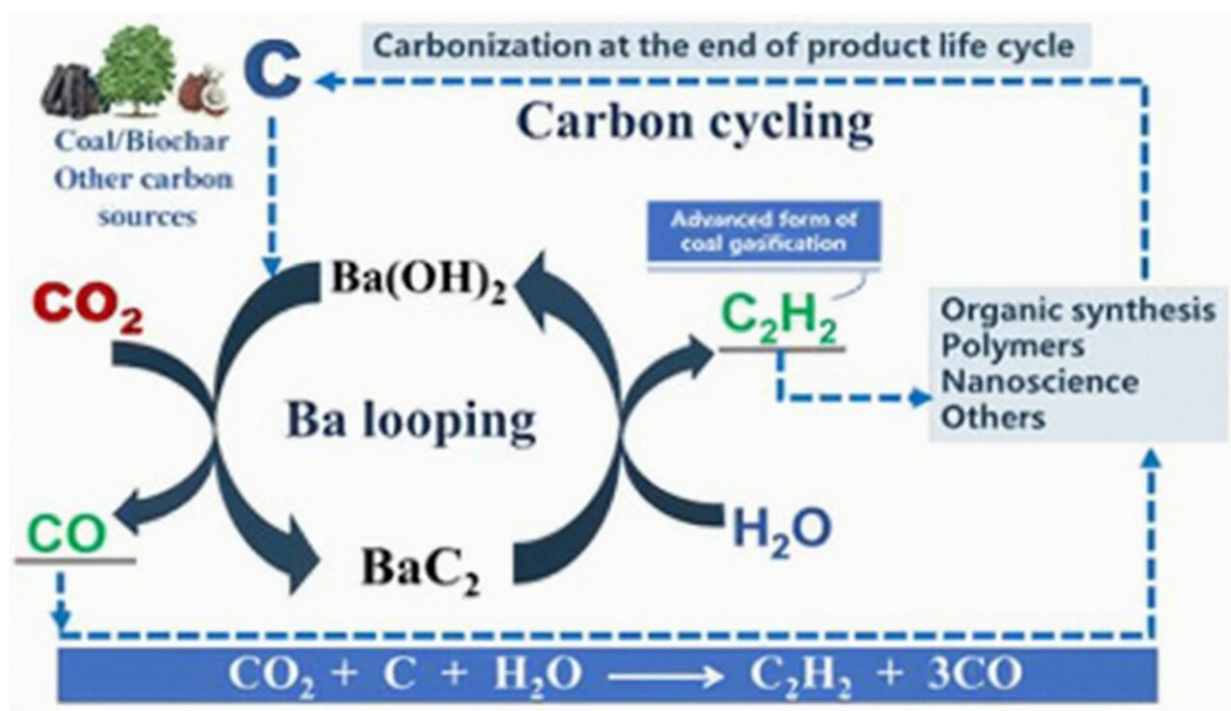


# Novel sustainable coupling technology proposed for carbon-to-acetylene process

October 10 2023, by Li Yuan



Credit: *Green Chemistry* (2023). DOI: 10.1039/D3GC01775C

The carbide-based carbon-to-acetylene ( $\text{C}_2\text{H}_2$ ) process is a simple pathway to convert various sources of carbon into acetylene and carbon monoxide directly. However, the current industrial process based on calcium carbide ( $\text{CaC}_2$ ) is restricted by high energy consumption, significant amount of carbon dioxide and industrial solid waste emission.

Recently, a research team led by Prof. Zhao Hong and Prof. Jiang Biao from the Shanghai Advanced Research Institute of the Chinese Academy of Sciences has proposed a sustainable [acetylene](#) and [carbon monoxide](#) coproducing process based on  $\text{BaCO}_3$ - $\text{BaC}_2$ - $\text{Ba}(\text{OH})_2$ - $\text{BaCO}_3$  barium cycle, which can simultaneously realize  $\text{CO}_2$  capture and acetylene-[carbon](#) monoxide co-production at mild dynamic conditions with lower energy consumption and less [waste](#) emission.

The results were published in [Green Chemistry](#) on Aug. 16.

The researchers found that  $\text{BaC}_2$  could be efficiently solid-phase synthesized at about  $1,500^\circ\text{C}$  by using carbon and  $\text{BaCO}_3$  as [raw materials](#) without  $\text{CO}_2$  emission, which is more than  $600^\circ\text{C}$  lower than the production temperature of  $\text{CaC}_2$ .

In addition,  $\text{Ba}(\text{OH})_2$  produced by the gasification of calcium carbide into acetylene was easily recovered and converted into  $\text{BaCO}_3$  by absorbing  $\text{CO}_2$ , which was then used to synthesize carbide, verifying the coupling process between carbon-to-acetylene and carbon dioxide capture based on Ba loop, reducing the waste of carbide slag.

The results suggested that  $\text{BaC}_2$  was the more suitable intermediate for carbon-to-acetylene process than  $\text{CaC}_2$ , because of the milder formation temperature, the faster reaction rate, and the more convenient barium recover to carbide production.

Featuring low cost, less wastes and high efficiency of co-producing of acetylene and carbon monoxide, this technology is expected to synthesize various of chemicals by using  $\text{C}_2\text{H}_2$  and  $\text{CO}$  as platform chemicals instead of  $\text{CO}$  and  $\text{H}_2$  produced by carbon gasification.

**More information:** Miao Li et al, Reengineering of the carbon-to-acetylene process featuring negative carbon emission, *Green Chemistry*

(2023). [DOI: 10.1039/D3GC01775C](https://doi.org/10.1039/D3GC01775C)

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