

Quantum dots: the tiny 'rainbow' crystals behind chemistry Nobel

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Scientists can carefully tailor quantum dots to any color they want, resulting in the nanotech being used in next-generation TV screens.

Quantum dots are tiny crystals that scientists can tune to different colors, giving an extra-vivid pop to next-generation TV screens or illuminating



tumors inside bodies so surgeons can hunt them down.

Three scientists won the <u>Nobel Chemistry Prize on Wednesday</u> for their work turning an idea first theorized in the 1930s into a reality that now has pride of place in living rooms across the world.

What are they?

Quantum dots are semiconducting particles just one thousandth the width of a human hair.

In 1937, the physicist Herbert Froehlich predicted that once particles were small enough—so-called nanoparticles—they would come under the strange spell of quantum mechanics.

To explain this <u>quantum phenomenon</u>, American Chemical Society president Judith Giordan said to "think of it like a little box".

When a particle is shrunk down small enough, the electron is "going to whack into the sides of the box," she told AFP.

In a larger box, the electrons would whack the sides less often, meaning they have less energy.

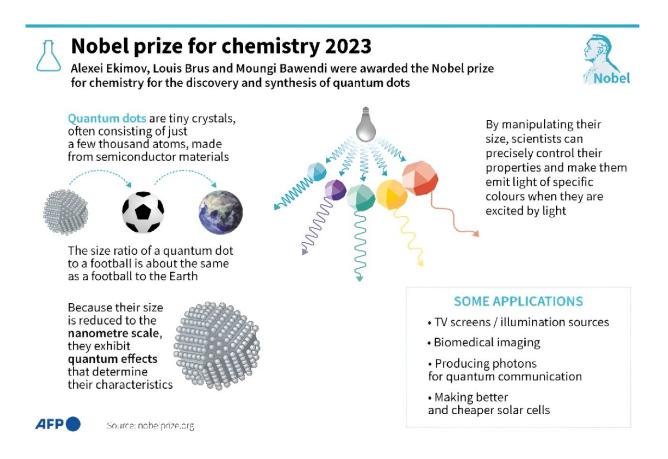
For quantum dots, the larger boxes emit <u>red light</u>, while the smaller ones show up blue.

This means that by controlling the size of the particle, scientists can make their crystals red, blue and everything in between.

Leah Frenette, an expert on quantum dots at Imperial College London, told AFP that working with the nanomaterial was like "watching rainbows all day".



But it would be 40 years after Froehlich's prediction that anyone was able to actually observe this phenomenon.



The discoveries made by the winners of the 2023 Nobel Prize for chemistry.

Who discovered what?

In the <u>early 1980s</u>, Russian-born physicist Alexei Ekimov—one of Wednesday's new laureates—melted colored glass and X-rayed the results.

He noticed that the smaller particles were more blue, also recognizing



that this was a quantum effect.

But being glass, the material was not easy to manipulate—and being published in a Soviet scientific journal meant few noticed.

At around the same time in the United States, another new laureate Louis Brus—oblivious of Ekimov's work—became the first to discover this colorful quantum effect in a liquid solution.

"For a long time, nobody thought you could ever actually make such small particles, yet this year's laureates succeeded," Nobel Committee member Johan Aqvist said.

"However, for quantum dots to become really useful, you needed to be able to make them in solution with exquisite control of their size and surface."

The third new Nobel winner, French-born Moungi Bawendi, found a way to do just this in his lab at the Massachusetts Institute of Technology in 1993.

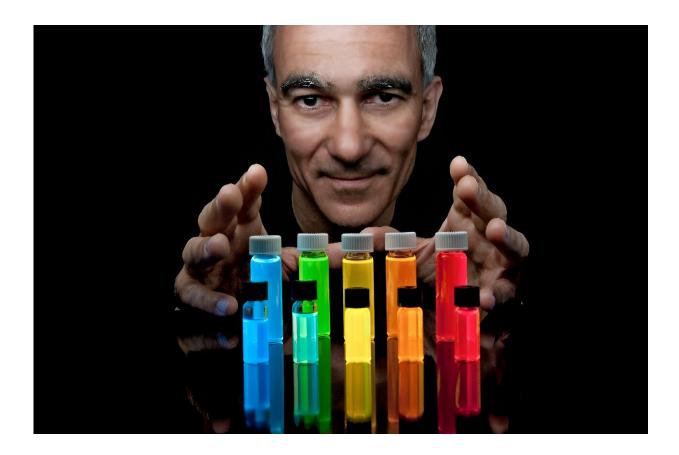
By precisely controlling the temperature of a liquid mixture of <u>particles</u> called colloid, Bawendi was able to grow nanocrystals to the exact size he wanted, paving the way for mass production.

What are they used in?

The most common everyday use of quantum dots is probably in "QLED" televisions.

Cyril Aymonier, head of France's Institute of Condensed Matter Chemistry, told AFP that the nanocrystals "improve the resolution of the screen and preserve the quality of the color for longer".





French-born scientist Moungi Bawendi found a way to grow nanocrystals just to the exact size he wanted, giving him control over quantum dots.

Doctors also use their bright fluorescence to highlight organs or tumors in the bodies of patients.

Frenette said she is working on <u>diagnostic tests</u> which would use the dots as "little beacons" for diseases in medical samples.

One problem is that most quantum dots are made using cadmium, a toxic heavy metal.

Both Aymonier and Frenette said they are working on quantum dots that



are not toxic.

Future use?

In the future, quantum dots could have the potential to double the efficiency of solar cells, Giordan said.

Their strange quantum powers could produce twice as many electrons as existing technology, she explained.

"That's amazing, because we are coming closer to the limit of current solar materials," she added.

Past use?

While quantum dots are considered on the cutting edge of science, people have probably been using them for centuries without knowing it.

The reds and yellows in stained glass windows as far as back as the 10th century show that artists of the time unwittingly took advantages of techniques that resulted in <u>quantum dots</u>, according to scientists.

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