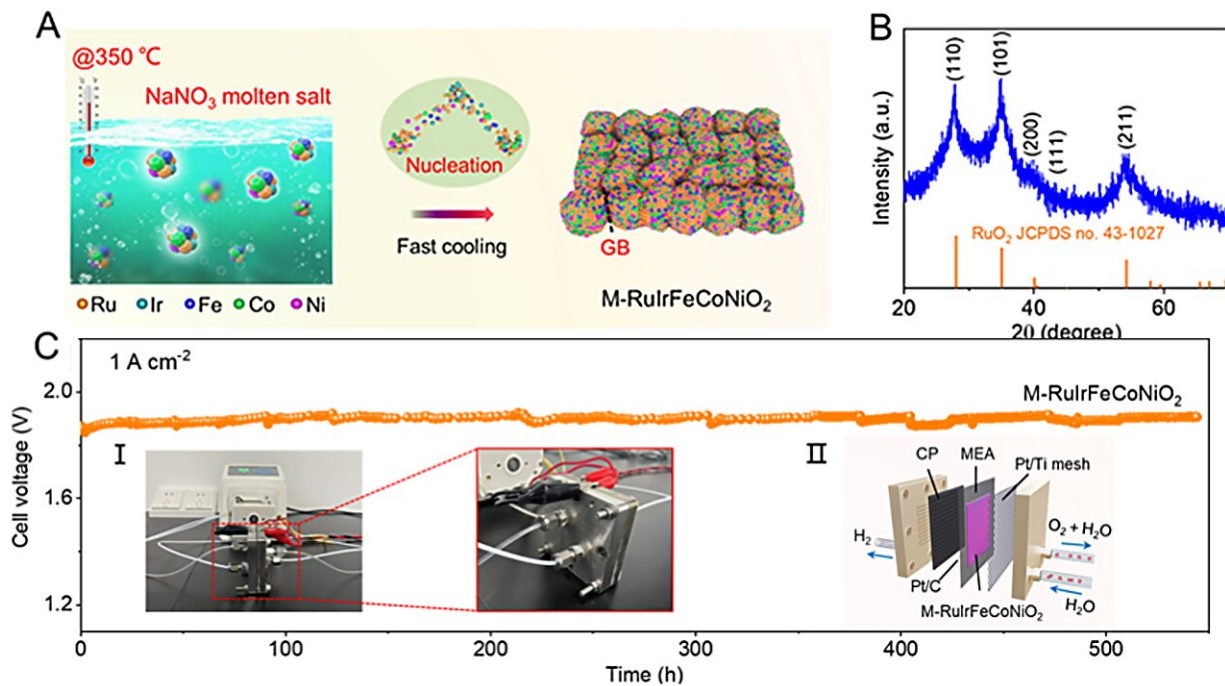


# Innovative catalyst advances acidic water splitting technology

September 18 2023, by Zhang Nannan



A. Schematic illustration of the fast, non-equilibrium synthetic process for synthesizing M-RuIrFeCoNiO<sub>2</sub>, showing the generation of GB; B. X-ray diffraction pattern of M-RuIrFeCoNiO<sub>2</sub>; C. Chronopotentiometric response of M-RuIrFeCoNiO<sub>2</sub> at 1 A cm<sup>-2</sup> in the PEM electrolyzer. Credit: Hu Chun

Researchers from the Shanghai Institute of Ceramics of the Chinese Academy of Sciences, together with collaborators, have made a significant breakthrough in electrocatalytic water splitting, a key

technology for converting intermittent solar and wind energy into clean hydrogen fuel.

According to the study published in [Science Advances](#), quinary high-entropy ruthenium iridium-based [oxide](#) holds promise for large-scale application in proton exchange membrane water electrolyzer (PEMWE).

In the pursuit of a hydrogen society, electrocatalytic water splitting has emerged as a potential solution. However, the acidic operating environment of the proton exchange membrane (PEM) has posed challenges for the long-term use of ruthenium oxide ( $\text{RuO}_2$ ). Now, the researchers led by Prof. Wang Xianying have discovered a quinary high-entropy five-membered ruthenium iridium-based oxide ( $\text{M-RuIrFeCoNiO}_2$ ), which has promising applications in PEMWE.

They developed a unique synthesis strategy for  $\text{M-RuIrFeCoNiO}_2$  to create abundant grain boundaries (GBs). This innovation significantly improves the [catalytic activity](#) and stability of  $\text{RuO}_2$  in acidic oxygen evolution reactions (OER), overcoming previous limitations.

The deliberate integration of foreign metal elements and GBs into the oxide catalyst played a pivotal role in improving OER activity and stability. This groundbreaking approach effectively solves the thermodynamic solubility problems associated with different metal elements.

Practical application tests showed remarkable results, as a PEMWE using the  $\text{M-RuIrFeCoNiO}_2$  catalyst maintained a high current density of  $1 \text{ A cm}^{-2}$  for more than 500 hours. This achievement marks a significant advancement in PEMWE technology and holds promise for the large-scale production of clean hydrogen fuel.

This study not only demonstrates a novel synthesis strategy for high

entropy oxides, but also provides valuable insights into their activity and stability in the context of PEMWE, contributing to the advancement of clean energy solutions.

**More information:** Chun Hu et al, Misoriented high-entropy iridium ruthenium oxide for acidic water splitting, *Science Advances* (2023).

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