

Capturing carbon dioxide with electricity: A microbial enzyme inspires electrochemistry

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The gas conversion process by an electrode-based enzymatic reaction. Credit: Lemaire/Belhamri/Wagner, Max Planck Institute for Marine Microbiology

Humans continuously emits greenhouse gases, worsening global warming. For example, carbon dioxide (CO₂) accumulates dramatically over the years and is chemically very stable. Yet, some microbes capture CO₂ using highly efficient enzymes. Scientists from the Max Planck Institute for Marine Microbiology in Bremen together with the

Universities of Geneva and Radboud isolated one of these enzymes.

When the enzyme was electronically branched on an electrode, they observed the conversion of CO₂ to formate with perfect efficiency. This phenomenon will inspire new CO₂-fixation systems because of its remarkable directionality and rates. The results are now published in *Angewandte Chemie*.

Seeking microorganisms that efficiently capture the greenhouse gas CO₂

"The enzymes employed by the microorganisms represent a fantastic playground for scientists as they allow highly specific reactions at fast rates," says Tristan Wagner, head of the Max Planck Research Group Microbial Metabolism at the Max Planck Institute for Marine Microbiology (MPIMM).

Some of these enzymes have an interesting way of capturing CO₂: They transform it into formate, a stable and safe compound that can be used to store energy or to synthesize various molecules for industrial or pharmaceutical purposes. One example is *Methanocaldococcus shengliensis*, a methanogen (a microbe producing methane) isolated from an oilfield and growing at 50 °C.

It has been cultivated and studied over the past years by Julia Kurth and Cornelia Welte at Radboud University in the Netherlands. At the Max Planck Institute for Marine Microbiology, Olivier Lemaire, Mélissa Belhamri and Tristan Wagner dissected the microbe to find its CO₂-capturing enzyme and measure how fast and efficiently it can transform CO₂.

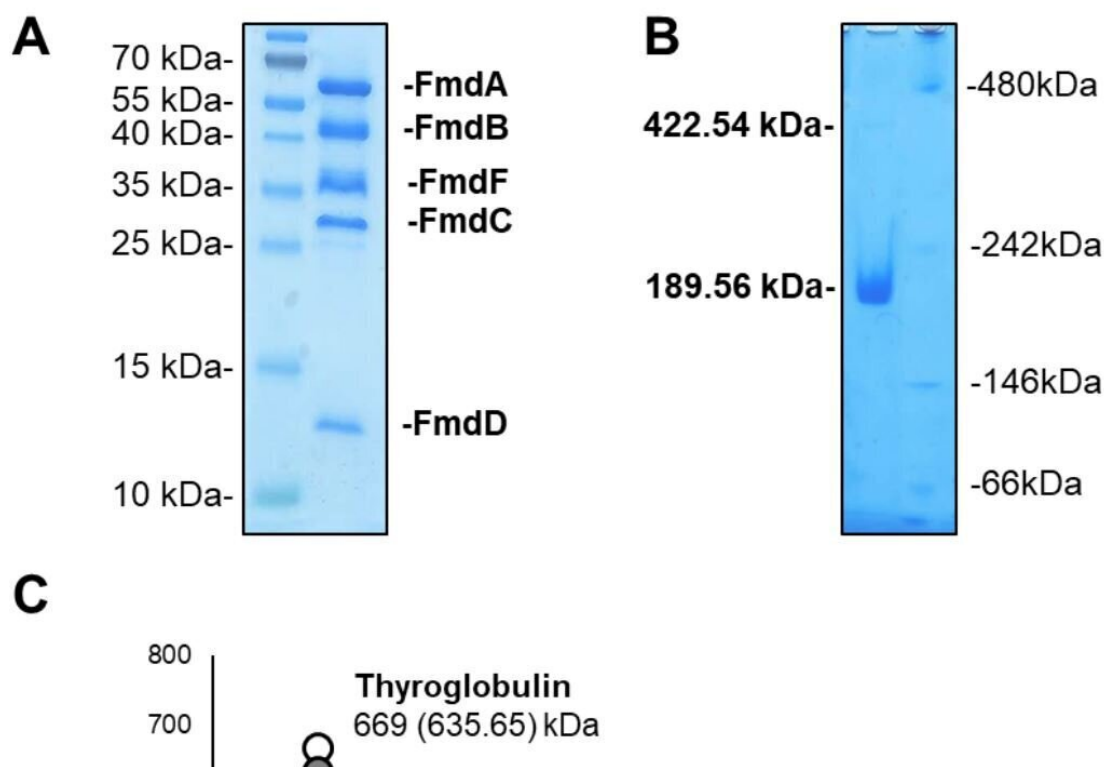
A CO₂-converting enzyme with great potential

The Max Planck-scientists undertook the challenging task to isolate the microbial enzyme. "Since we knew that such enzymes are sensitive to oxygen, we had to work inside an anaerobic tent devoid of ambient air to separate it from the other proteins—quite complicated, but we succeeded," says Olivier Lemaire. Once isolated, the scientists characterized the enzyme's properties.

They showed that it efficiently generates formate from CO_2 but performs the reverse reaction at very slow rates and poor yield. "Similar enzymes belonging to the family of formate dehydrogenases are well known to operate in both directions, but we showed that the enzyme from *Methermicoccus shengliensis* is nearly unidirectional and could not efficiently convert the formate back into CO_2 ," reports Méliissa Belhamri. "We were quite thrilled by this phenomenon, occurring only in the absence of oxygen," she adds.

"Since the formate generated from CO_2 -fixation cannot be transformed back and therefore accumulates, such a system would be a highly interesting candidate for CO_2 -capture, especially if we could branch it on an electrode," Tristan Wagner points out.

The advantage of that: With the enzyme naturally or chemically attached to an electrode, the "energy" required to capture the CO_2 will be directly delivered by the electrode, without electric current loss or the need for expensive or toxic chemical compounds as relays. Consequently, the enzyme-bound electrodes are efficient and attractive systems for gas conversion procedures. Thus, the purified enzyme was sent to the University of Geneva to set up an electrode-based CO_2 -capture system.



Composition and molecular weight of the purified MsFmd sample. A. SDS PAGE profile of the purified MsFmd complex. B. Native PAGE profile of the complex. The estimated molecular weight of the bands, calculated based on the ladder, is indicated. C. Determination of the molecular weight of MsFmd by size exclusion chromatography. The calibration curve was determined with a calibration kit (GE Healthcare). The theoretical (calculated) molecular weight of the protein standards are indicated. Theoretical and calculated molecular weight are shown as white and gray dots, respectively. Red dots show the molecular weight determined during two distinct purification process. The MsFmd size is averaged from these two experiments. Credit: *Angewandte Chemie International Edition* (2023). DOI: 10.1002/anie.202311981

Electricity-based gas conversion

Selmihan Sahin and Ross Milton from the University of Geneva are

specialists in electrochemistry. They use electrodes connected to electric current to perform chemical reactions. The electrode-based formate generation from CO_2 often requires polluting and [rare metals](#), and that is why they tried to replace these metals with the enzyme extracted in the group of Tristan Wagner at the MPIMM.

The procedure of enzyme binding on an electrode is not always as efficient as expected, but the enzyme from Wagner's research group has specific characteristics that could facilitate the process. The scientists from Switzerland managed to fix the enzyme on a graphite electrode, where it performed the gas conversion.

The measured rates were comparable to those obtained with classic formate dehydrogenases. "The strength of this biological system coupled to the electrode lies in its efficiency in transferring the electrons from the electricity towards CO_2 transformation," highlights Lemaire.

Sahin and Milton also confirmed that the system performs the reverse reaction poorly, as previously observed in the reaction tube. Consequently, the modified electrode continuously converted the greenhouse gas to formate without any detectable side-products generated or electric current loss.

Towards a new solution for atmospheric CO_2 utilization

The collaborative work provides a new molecular tool to the [scientific community](#): An enzyme converting CO_2 by transferring electricity with high efficiency. Renewable green energy (e.g., wind or solar) could provide electricity to the electrode-based system that would turn CO_2 into formate, a molecule directly usable for applications or to store energy.

"Before us, no one ever tried to study an enzyme from such a

methanogen for an [electrode](#)-based gas conversion," says Tristan Wagner. "Yet, methanogens are natural outstanding gas converters."

As powerful as they could be, employing enzymes for large-scale processes would also require similar-scale enzyme production systems, a considerable investment. Therefore, while the discovered strategy could, in theory, significantly improve CO₂ transformation, a deep knowledge of the [enzyme](#) mechanism is necessary before its application, and the team of researchers will now have to dissect in depth the molecular secrets of the reaction.

More information: Selmihan Sahin et al, Bioelectrocatalytic CO₂ Reduction by Mo-Dependent Formylmethanofuran Dehydrogenase, *Angewandte Chemie International Edition* (2023). [DOI: 10.1002/anie.202311981](https://doi.org/10.1002/anie.202311981)

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