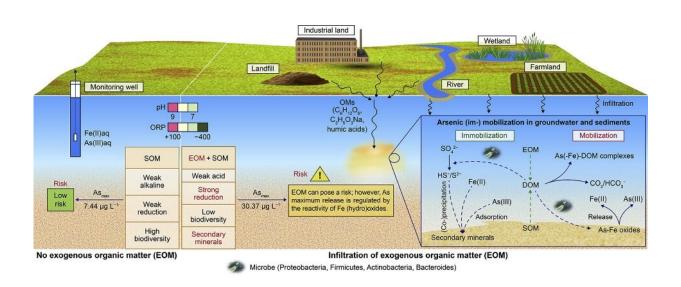


New study uncovers potential risk of arsenic release from sediment under organic matter influence

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Graphical abstract. Credit: Environmental Science and Ecotechnology

Researchers from the Chinese Research Academy of Environmental Sciences have conducted a study to assess the impact of environmental factors and microbial communities on the mobilization of arsenic (As). The findings, published in the journal *Environmental Science and Ecotechnology*, reveal important insights into the biogeochemical processes involved in As release.

The study focused on processes such as desorption, reduction,



complexation, and co-precipitation that affect the As behavior in the environment. The interaction between Fe (hydro) oxides and organic matters (OMs), particularly dissolved <u>organic matter</u> (DOM), was identified as a crucial control mechanism. The OMs were characterized using fluorescence indices, which indicated sustained biological activities throughout the experimental period.

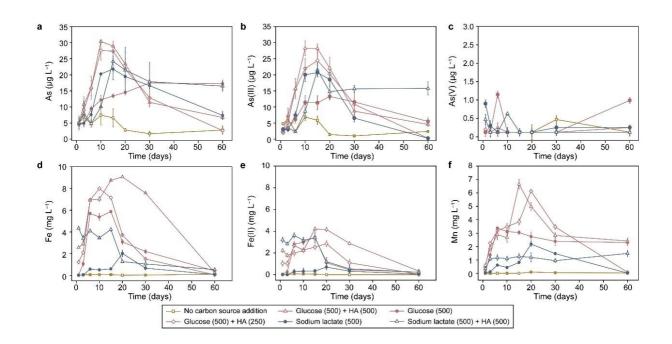
The microbial community analysis unveiled the presence of bacteria capable of reducing Fe, Mn, and As, as well as bacteria involved in metabolic transformations using EOM. When bio-reactive and chemically reactive OMs were introduced, they created a reduction environment facilitating the release of As, Fe, and Mn, particularly at high OM concentrations. Glucose and sodium lactate, easily metabolized by microorganisms, resulted in higher releases than the <u>control group</u> without OMs.

The addition of humic acid (HA), a chemically reactive OM, significantly influenced the release of Fe and Mn, albeit with a lesser impact on As. The study also observed the formation of secondary Fe minerals, such as siderite and mackinawite, which incorporated As and contributed to the decline in As, Fe, and Mn concentrations in the aqueous phase.

Microbial decomposition altered the characteristics of DOM, leading to the production of amino acids and the presence of polysaccharides, as indicated by specific functional groups. Furthermore, the research employed canonical correspondence analysis (CCA) and redundancy analysis (RDA) to examine the relationship between <u>environmental</u> <u>factors</u>, the microbial community, and As mobilization.

Positive correlations were found between As(III), Fe, and Mn, while a negative correlation was observed with oxidation-reduction potential (ORP). Several bacterial genera associated with As metabolism were





identified, underscoring their role in the release process.

With the addition of OM, As, Fe, and Mn were significantly released from the sediment. In general, As, Fe, and Mn concentrations increased first and then decreased during the experiment. Samples with As(V) or As(III) below the detection limit of 0.25 μ g L–1 were plotted as half of the detection limit (0.125 μ g L–1) to make the comparison. The error bar is the standard deviation (n = 4). Credit: *Environmental Science and Ecotechnology*

The research significantly enhances our understanding of the intricate factors influencing the release of arsenic (As) and sheds light on the microbial processes involved in As mobilization in aquatic environments. The findings have important implications for managing and mitigating groundwater pollution caused by the infiltration of EOM.

Specific sites, such as landfills, petrochemical sites, and managed aquifer recharge projects, are identified as particularly vulnerable to



contamination. Further investigations are necessary to explore the effects of hydrodynamics and hydrogeochemical environments in practical applications. These findings underscore the need for comprehensive strategies to effectively control and mitigate the <u>environmental risks</u> associated with EOM infiltration, aiming to safeguard groundwater quality.

More information: Fan Feng et al, Exogenous-organic-matter-driven mobilization of groundwater arsenic, *Environmental Science and Ecotechnology* (2023). DOI: 10.1016/j.ese.2023.100243

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