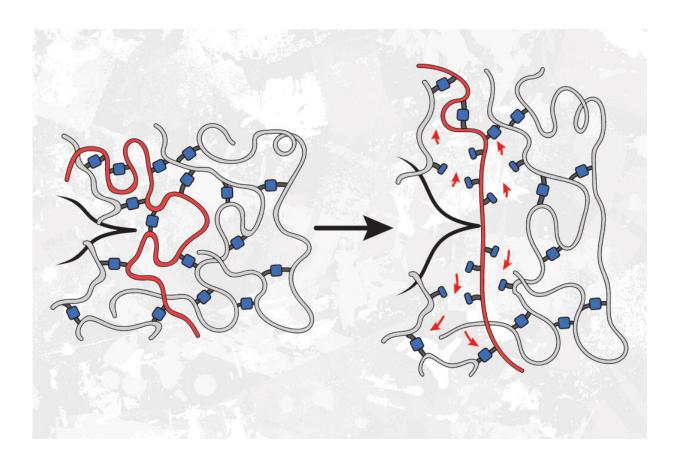


Adding weaker bonds can enhance polymer's resistance to tearing

June 22 2023



As this polymer network is stretched, weaker crosslinking bonds (blue) break more easily than any of the strong polymer strands, making it more difficult for a crack to propagate through the material. Credit: Massachusetts Institute of Technology

A team of chemists from MIT and Duke University has discovered a



counterintuitive way to make polymers stronger: introduce a few weaker bonds into the material.

Working with a type of polymer known as polyacrylate elastomers, the researchers found that they could increase the materials' resistance to tearing up to tenfold, simply by using a weaker type of crosslinker to join some of the polymer building blocks.

These rubber-like polymers are commonly used in car parts, and they are also often used as the "ink" for 3D-printed objects. The researchers are now exploring the possible expansion of this approach to other types of materials, such as rubber tires.

"If you could make a rubber tire 10 times more resistant to tearing, that could have a dramatic impact on the lifetime of the tire and on the amount of microplastic waste that breaks off," says Jeremiah Johnson, a professor of chemistry at MIT and one of the senior authors of the study, which appears in *Science*.

A significant advantage of this approach is that it doesn't appear to alter any of the other physical properties of the polymers.

"Polymer engineers know how to make materials tougher, but it invariably involves changing some other property of the material that you don't want to change. Here, the toughness enhancement comes without any other significant change in physical properties—at least that we can measure—and it is brought about through the replacement of only a small fraction of the overall material," says Stephen Craig, a professor of chemistry at Duke University who is also a senior author of the paper.

This project grew out of a longstanding collaboration between Johnson, Craig, and Duke University Professor Michael Rubinstein, who is also a



senior author of the paper. The paper's lead author is Shu Wang, an MIT postdoc who earned his Ph.D. at Duke.

The weakest link

Polyacrylate elastomers are polymer networks made from strands of acrylate held together by linking molecules. These building blocks can be joined together in different ways to create materials with different properties.

One architecture often used for these polymers is a star polymer network. These polymers are made from two types of building blocks: one, a star with four identical arms, and the other a chain that acts as a linker. These linkers bind to the end of each arm of the stars, creating a network that resembles a volleyball net.

In a 2021 study, Craig, Rubinstein, and MIT Professor Bradley Olsen teamed up to measure the strength of these polymers. As they expected, they found that when weaker end-linkers were used to hold the polymer strands together, the material became weaker. Those weaker linkers, which contain cyclic molecules known as cyclobutane, can be broken with much less force than the linkers that are usually used to join these building blocks.

As a follow-up to that study, the researchers decided to investigate a different type of polymer network in which <u>polymer</u> strands are cross-linked to other strands in random locations, instead of being joined at the ends.

This time, when the researchers used weaker linkers to join the acrylate building blocks together, they found that the material became much more resistant to tearing.



This occurs, the researchers believe, because the weaker bonds are randomly distributed as junctions between otherwise strong strands throughout the material, instead of being part of the ultimate strands themselves. When this material is stretched to the breaking point, any cracks propagating through the material try to avoid the stronger bonds and go through the weaker bonds instead. This means the crack has to break more bonds than it would if all of the bonds were the same strength.

"Even though those bonds are weaker, more of them end up needing to be broken, because the crack takes a path through the weakest bonds, which ends up being a longer path," Johnson says.

Tough materials

Using this approach, the researchers showed that polyacrylates that incorporated some weaker linkers were nine to 10 times harder to tear than polyacrylates made with stronger crosslinking molecules. This effect was achieved even when the weak crosslinkers made up only about 2% of the overall composition of the material.

The researchers also showed that this altered composition did not alter any of the other properties of the material, such as resistance to breaking down when heated.

"For two materials to have the same structure and same properties at the network level, but have an almost order of magnitude difference in tearing, is quite rare," Johnson says.

The researchers are now investigating whether this approach could be used to improve the toughness of other materials, including rubber.

More information: Shu Wang et al, Facile mechanochemical



cycloreversion of polymer cross-linkers enhances tear resistance, *Science* (2023). <u>DOI: 10.1126/science.adg3229</u>. www.science.org/doi/10.1126/science.adg3229

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