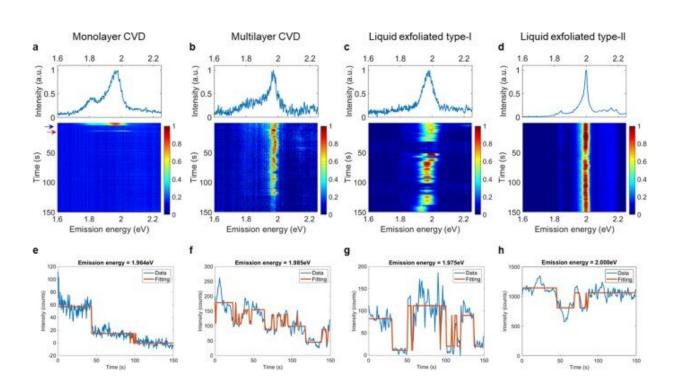


Prolonged photostability in hexagonal boron nitride quantum emitters



Photostability of emitters of three common commercial hBN samples in the air. Photoluminescence (PL) spectra at time 0 s (upper panels) and time-traces from 0 s to 150 s (lower panels) with zero-phonon lines (ZPLs) near 1.98 eV in asprepared (a) monolayer CVD, (b) multilayer CVD, and multilayer liquid exfoliated hBN with (c) type-I and (d) type-II emission. In the lower panel of (a), the blue arrow indicates a ZPL disappearing at approximately 5 s, while the red arrow highlights a new emission peak appearing at approximately 13 s, which is shifted +50 meV from the original ZPL, before completely losing emission signal. Representative time-dependent intensity at a fixed emission energy is plotted in blue lines of (e) monolayer CVD, (f) multilayer CVD, and multilayer liquid exfoliated hBN with (g) type-I and (h) type-II emission. Orange lines in (e)

April 11 2023, by Thamarasee Jeewandara



to (h) are fitting with a cluster analysis, details of which are presented in Supplementary Note 1 of the paper. In (e) monolayer CVD hBN, discrete bleaching steps are observed, where each step may correspond to an individual emitter. Credit: *Communications Materials* (2023). DOI: 10.1038/s43246-023-00345-8

Single-photon emitters are crucial building materials suited for optical quantum technologies. Among them, hexagonal boron nitride is a promising two-dimensional material that retains bright, roomtemperature single-photon emitters. However, photo-instability is an existing challenge to facilitate applications of these properties in practice.

In a new study published in *Communications Materials*, Sylvia Xin Li and a team of scientists at the department of Chemical Engineering, Massachusetts Institute of Technology, University of Texas, RIKEN center for Advanced Photonics, and the University of Oxford, revealed the possibility of photobleaching <u>hexagonal boron nitride</u> vacancy emitters to facilitate photostability suited for quantum applications.

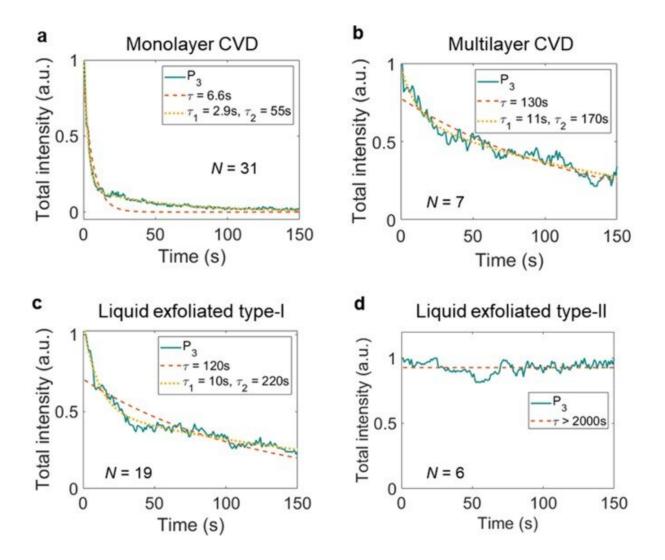
Designing quantum emitters

Quantum mechanics can be harnessed for real-world applications as one of the most intriguing and fastest growing technologies exploring quantum states of photons for <u>information processing and transmission</u>. Researchers have investigated a variety of solid-state single photon sources in <u>three-dimensional crystals</u>; however, the light emanating from these sources can be contained by the surrounding bulky dielectric environment to cause reduced efficiency of emission.

Scientists can overcome this challenge of color centers in two-



dimensional hosts such as transition metal dichalcogenides and hexagonal boron nitrides (hBN) via dielectric materials with significantly reduced thickness at the level of a single atomic layer to integrate with on-chip photonics. While single photon emission in transition metal dichalcogenides is only observed at room temperature, the material itself can host bright, room-temperature single-photon emitters to provide materials such as hexagonal boron nitride with unique photophysical properties to function at cryogenic temperatures.



Total intensity of P3 emission in the air. Since individual time-traces have discrete bleaching steps, the overall decay curves of total intensity are obtained



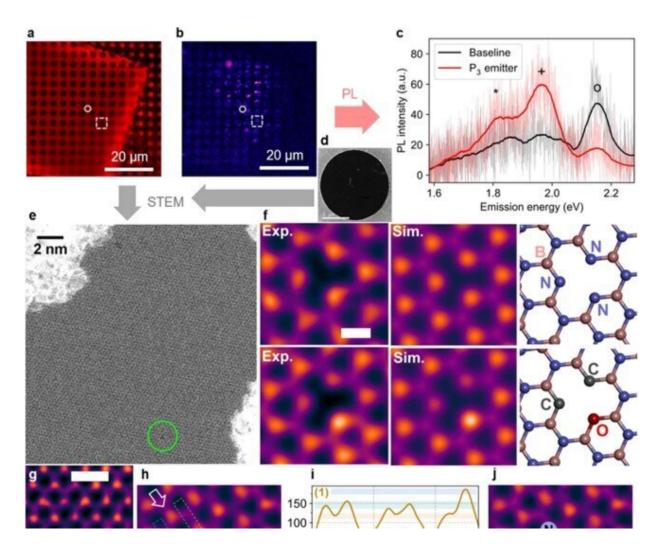
by summing up N time-traces. Here, N = 31, 7, 19 and 6 for (a) monolayer CVD hBN, (b) multilayer CVD hBN, (c) liquid exfoliated hBN with type-I emission and (d) liquid exfoliated hBN with type-II emission, respectively. Red dashed lines and yellow dotted lines are fitting with one exponential decay function, and the sum of two exponential decay functions, respectively. From the one-exponential fitting with 95% confidence interval, the bleaching lifetime in monolayer CVD hBN is 6.6 ± 0.6 s. From the two-exponential fitting with 95% confidence interval, the bleaching lifetime in monolayer CVD hBN is 6.6 ± 0.6 s. From the two-exponential fitting with 95% confidence interval, two bleaching lifetimes are extracted for monolayer CVD hBN, multilayer CVD and liquid exfoliated hBN with type-I emission, which are 2.9 ± 0.1 and 55 ± 3.2 s, 11 ± 4.1 and 170 ± 20 s, and 10 ± 1.6 s and 220 ± 30 s, respectively. Liquid exfoliated hBN with type-II emission has a bleaching lifetime longer than 2000 s, which is regarded as photostable. Credit: *Communications Materials* (2023). DOI: 10.1038/s43246-023-00345-8

Atomic-scale imaging

In this work, Li and colleagues quantitatively studied the photobleaching lifetimes of various hexagonal boron nitride samples and revealed them to be dominated via photochemical reactions with oxygen. The team showed bleaching to be significantly reduced by a simple strategy of stacking additional monolayers of the compound on the material.

The scientists characterized the photobleached monolayer with <u>annular</u> <u>dark-field scanning transmission electron microscopy</u> to achieve highquality atomic scale imaging of the hBN lattice. They combined X-ray photoelectron spectroscopy and annular dark-field scanning <u>transmission</u> <u>electron microscopy</u> to achieve high-quality atomic-scale imaging of the hexagonal boron nitride lattice.





Characterizing a bleached emitter with ADF-STEM. False-color optical micrograph (a) and wide-field photoluminescence imaging (b) of monolayer CVD hBN, suspended on a holey carbon film of TEM grid. Scale bars in (a) and (b) correspond to 20 µm. c Confocal photoluminescence spectrum, showing a zero-phonon line (+) near 1.98 eV (P3) and one-phonon sideband (*) from the suspended hBN monolayer, and a Raman response (o) from the carbon film of the grid. Light-colored and dark-colored lines correspond to the raw and smoothed data, respectively. P3 emission is in red, and the background spectrum in black. In (a) and (b), the circle denotes the position of the P3 spectrum. Note that only P3 emission is observed in the hBN covering this TEM grid. d Zoomed-out ADF-STEM image of a hole in the grid, which is the same position as the dotted square in (a) and (b). Scale bar corresponds to 1 µm. e ADF-STEM



image of a typical clean region of hBN. The green circle indicates a point defect. Scale bar corresponds to 2 nm. f Representative ADF-STEM analysis of monovacancies missing boron with atomic-scale resolution, showing experimental images (left column), simulated images (middle column) and their corresponding atomic structures (right column). The top row presents a monovacancy with NNN innermost edge atoms, while the bottom row shows OCC replacing NNN. g Averaged intensity of two consecutive frames under high magnification, showing a monovacancy missing boron, which has two O replacing inner most edge N, with higher uniformity in the image by noise reduction. Scale bar corresponds to 2 Å in (f) and 5 Å in (g), respectively. h A representative image of vacancy-free region under high magnification, the white arrow indicates the direction of line 1, 2 and 3. i Line profile analysis corresponding to line 1, 2, 3 in (h). The y-axis values indicate the intensity. j, ADF-STEM image from (h) with annotations, from the analysis in (i). Scale bars correspond to 5 Å in (h) and (j). Credit: Communications Materials (2023). DOI: 10.1038/s43246-023-00345-8

The photostability of emitters in various hexagon boron nitride (hBN) samples

The research team compared three commercially available hBN thin films, including monolayer <u>chemical vapor deposition</u> grown hexagon boron nitride on copper foil, multilayer chemical vapor deposition grown hBN on copper foil and multilayer-liquid-exfoliated hBN nanoflakes suspended in ethanol/water mixtures. The team transferred these samples into silicon dioxide substrates without any post-treatment and examined them with time-dependent photoluminescence measurements by using a <u>custom-built confocal spectroscopic microscope</u>.

Li and colleagues focused on the emitter species observed in all hBN samples and noted the presence of three different emission behaviors in monolayer and multilayer chemical vapor deposition samples. Multilayer



liquid exfoliated hexagonal boron nitride as-prepared without thermal annealing treatment showed two different emission modes of interest, which they characterized extensively.

Bleaching mechanisms in monolayer chemical vapor deposition hexagonal boron nitride (hBN)

The scientists studied the bleaching mechanisms of monolayer hBN emitters under a variety of atmospheres using a custom-built environmental chamber. They accomplished this by placing a monolayer sample of chemical vapor-deposition grown hexagonal boron nitride in a small sealed chamber with a quartz observation window and filled the gas chamber with a variety of gases, including nitrogen, oxygen and water vapor. Since many organic fluorophores are degraded with oxygen or water, the team filled the environmental chamber with nitrogen to minimize hydration.

While the nitrogen atmosphere did not mitigate bleaching completely when the team conducted bleaching experiments in an oxygen-rich atmosphere, they noted its influence in the process of photobleaching. The scientists further examined bleaching mechanisms of the material via <u>annular dark-field in the scanning transmission electron microscope</u> (ADF-STEM).



	1	2	3	4	5	6	7	8	9	10	11	12
Defect 1	Ν	Ν	\mathbf{N}	в	Ν	В	В	Ν	в	В	Ν	В
Defect 2					\mathbf{C}					в	С	С
Defect 3	N	С	0	В	Ν	В	В	Ν	В	В	Ν	В
Defect 4	Ν	С	0	В	Ν	В	С	Ν	В	В	Ν	В
Defect 5	0	С	Ν	В	С	В	С	Ν	В	В	Ν	В
Defect 6	O N	0	N	В	Ν	В	С	Ν	В	В	Ν	В

Atomic compositions of VB monovacancies. A nomenclature is introduced to label the heteroatom substitutions, which enumerates an edge of the innermost three N atoms clockwise as 1 to 3 highlighted in bold type, and the remaining B and N atoms as 4 to 12. Credit: *Communications Materials* (2023). DOI: 10.1038/s43246-023-00345-8

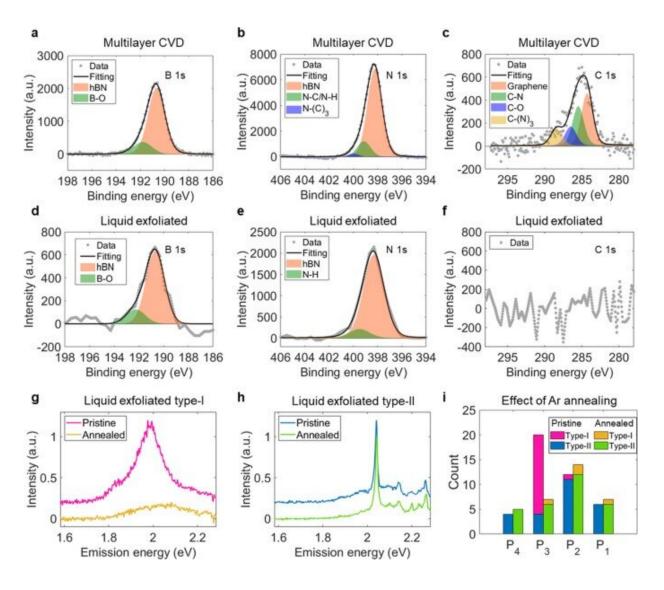
Chemical analysis and thermal annealing effects



Li and colleagues conducted X-ray photon spectroscopy to identify different elements and their chemical states with the materials in contrast to elemental analysis with <u>inductively coupled plasma</u> techniques. To accomplish this, they first sputtered the sample surfaces with an argon ion beam to remove any adsorbed hydrocarbons. As the timeframe of Argon sputtering increased, the atomic concentration of carbon on the multilayer surface decreased to reach a stable value. The outcomes were consistent with ADF-STEM imaging outcomes of the monolayer compound.

With higher surface carbon concentrations, the researchers grew atomic layers of <u>hybridized boron nitride and graphene domains</u> with methane and ammonia borane as precursors. They then synthesized hBN from a lab-built, low-pressure chemical vapor deposition reactor with typical hBN crystals of triangular shape, and observed the materials with scanning electron microscopy. The team also noted the presence of polygonal sheets in similar regions, which they credited to the presence of graphene by analyzing the regions with <u>Raman spectroscopy</u>.





XPS characterization of different hBN samples and annealing effect on liquid exfoliated hBN. XPS spectra of hBN samples that are pre-cleaned with Ar ion sputtering, showing B 1 s (a, d), N 1 s (b, e), and C 1 s (c, f). While (a) to (c) represent multilayer CVD hBN, (d) to (f) are results of multilayer liquid exfoliated hBN. The C concentration is higher in the CVD sample, mainly in the form of sp2 C-C, which corresponds to graphene domains. Note that C-N is different from C-(N)3, with details in Supplementary Fig. 18 and Supplementary Table 3. g Photoluminescence (PL) spectra of type-I emission in liquid exfoliated hBN, before (pink) and after (yellow) annealing at 850 °C in Ar for 30 min. h PL spectra of type-II emission in liquid exfoliated hBN, with sharper zero-phonon lines (ZPLs), before and after annealing at 850 °C in Ar for 30 min. i Number of each emitter species Pi (i = 1, 2, 3, 4) in liquid exfoliated hBN,



before (left bars) and after (right bars) 850 °C annealing in Ar. Pink and yellow colors indicate broad ZPLs, corresponding to type-I emission. Blue and green indicate sharp ZPLs, corresponding to type-II emission. Before annealing, P3 has a large percentage of type-I emitters, which disappear almost completely after annealing. All type-II emitters remain the same after annealing, regardless of the emission species Pi (i = 1, 2, 3, 4). Credit: *Communications Materials* (2023). DOI: 10.1038/s43246-023-00345-8

Outlook

In this way, Sylvia Xin Li and a research team in engineering, chemistry and photonics, identified a key to improve the photostability of emitters in hexagonal boron nitride materials that effectively shielded oxygen from the environment to optimize carbon substitution in the hexagonal boron nitride (hBN) lattice. They noted photobleaching of hBN emitters to be in direct contact with air, and dominated by photochemical reactions emitting defects with oxygen. The team mitigated the effects of bleaching by shielding molecular oxygen and by introducing a nitrogen atmosphere, or by stacking additional hexagonal boron nitride layers on the material.

This strategy improved the average emitter lifetime by about 20 times. The team compared the experimental results across a range of imaging methods and thermal annealing to verify the outcomes to provide indepth insights to the structural origins of hBN quantum emission. The knowledge gained about defect engineering hexagonal <u>boron nitride</u> will be insightful across broader fields of study to tune the electric properties of materials within 2D integrated devices and in <u>2D quantum materials</u> for future sensing.

More information: Sylvia Xin Li et al, Prolonged photostability in hexagonal boron nitride quantum emitters, *Communications Materials*



(2023). DOI: 10.1038/s43246-023-00345-8

Toan Trong Tran et al, Quantum emission from hexagonal boron nitride monolayers, *Nature Nanotechnology* (2015). DOI: 10.1038/nnano.2015.242

© 2023 Science X Network

Citation: Prolonged photostability in hexagonal boron nitride quantum emitters (2023, April 11) retrieved 24 April 2024 from <u>https://phys.org/news/2023-04-prolonged-photostability-hexagonal-boron-nitride.html</u>

This document is subject to copyright. Apart from any fair dealing for the purpose of private study or research, no part may be reproduced without the written permission. The content is provided for information purposes only.