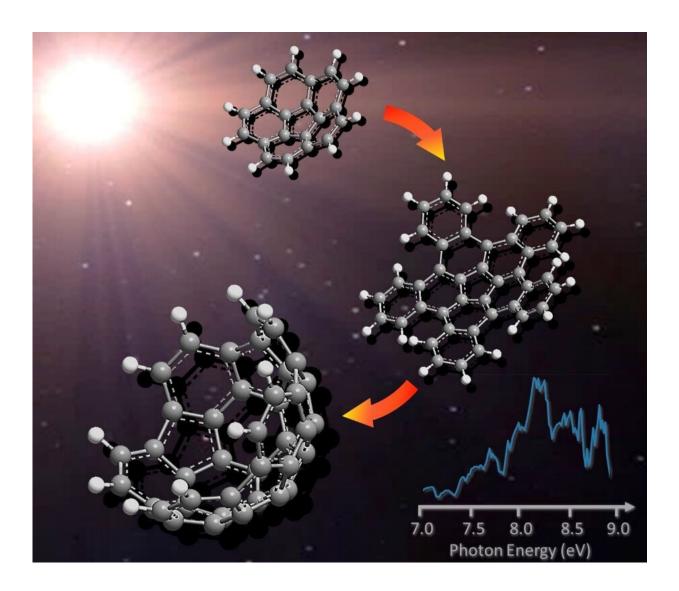


How football-shaped molecules occur in the universe

March 27 2023, by Bernd Müller



Credit: Shane Goettl/Ralf I. Kaiser



For a long time it has been suspected that fullerene and its derivatives could form naturally in the universe. These are large carbon molecules shaped like a football, salad bowl or nanotube. An international team of researchers using the Swiss SLS synchrotron light source at PSI has shown how this reaction works. The results have just been published in the journal *Nature Communications*.

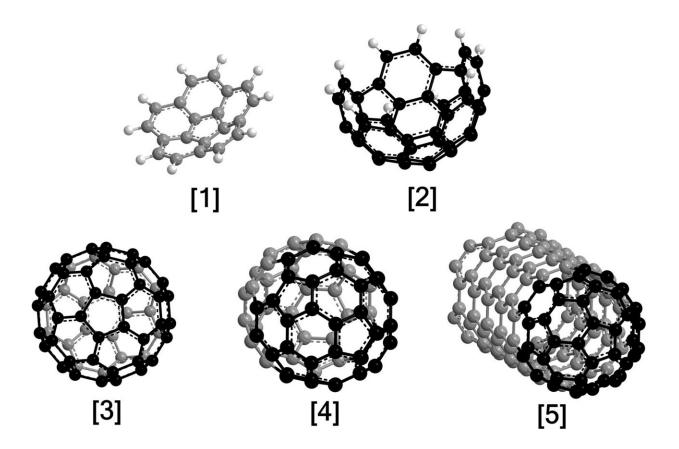
"We are stardust, we are golden. We are billion-year-old carbon." In the song they performed at Woodstock, the US group Crosby, Stills, Nash & Young summarized what humans are essentially made of: star dust. Anyone with a little knowledge of astronomy can confirm the words of the cult American band—both the planets and we humans are actually made up of dust from burnt-out supernovae and carbon compounds billions of years old. The <u>universe</u> is a giant reactor and understanding these reactions means understanding the origins and development of the universe—and where humans come from.

In the past, the formation of fullerenes and their derivatives in the universe has been a puzzle. These <u>carbon molecules</u>, in the shape of a football, bowl or small tube, were first created in the laboratory in the 1980s. In 2010 the infrared space telescope Spitzer discovered the C_{60} molecules with the characteristic shape of a soccer ball, known as buckyballs, in the planetary nebula Tc 1. They are therefore the biggest molecules to have been discovered to date known to exist in the universe beyond our solar system.

But how do they actually form there? A team of researchers from Honolulu (U.S.), Miami (U.S.) and Tianjin (China) has now completed an important reaction step in the formation of the molecules, with active support from PSI and the vacuum ultraviolet (VUV) beamline of the synchrotron light source Swiss SLS. "PSI offers unique experimental facilities and that's why we decided to collaborate with Patrick Hemberger at PSI," says Ralf Kaiser from the University of Hawaii in



Honolulu, the leading international researcher in this field.



Molecular structures of 3-dimensional carbonaceous nanostructures. The smallest nanobowl, corannulene, as well as the C40 nanobowl [2] emphasized as a molecular building block of Buckminsterfullerene (C_{60}), rugbyballene (C_{70}), and end-capped (5,5) armchair nanotubes. The C40 nanobowl carbons are black, the remaining carbons are gray, and the hydrogens are white. Credit: *Nature Communications* (2023). DOI: 10.1038/s41467-023-37058-y

A mini reactor for fullerene

Patrick Hemberger, a scientist working on the VUV beamline at PSI, has built a mini reactor for observing the formation of fullerene in real time.



A corannulene radical $(C_{20}H_9)$ is created in a reactor at a temperature of 1,000 degrees Celsius. This molecule looks like a salad bowl, as if it had been dissected from a C_{60} buckyball. This radical is highly reactive. It reacts with vinyl acetylene (C_4H_4) , which deposits a layer of carbon onto the rim of the bowl.

"By repeating this process many times, the molecule would grow into the end cap of a nanotube. We have managed to demonstrate this phenomenon in <u>computer simulations</u>," explains Alexander Mebel, Professor of Chemistry at Florida International University and one of the authors of the study. But that was not the researchers' only goal: "We wanted to show that this type of reaction is physically possible," Ralf Kaiser adds.

The reaction produces different isomers—molecules that all have the same mass, but slightly different structures. With standard mass spectrometry, all these variants produce the same signal. But the outcome is different when using photoelectron photoion coincidence spectroscopy, the method adopted by the team. "With this technique, the structure of the measurement curve allows conclusions to be drawn about each individual isomer," Patrick Hemberger explains.

Solving the puzzle of classic football-shaped molecules

"The universe contains a wild jungle of molecules and <u>chemical</u> <u>reactions</u>—not all of them can be distinctly classified in the signals from telescopes," says Ralf Kaiser. We already know from models that both corannulene and vinylacetylene exist in the universe. Now it has been possible to confirm that these molecules actually form the building blocks to fullerene. "That's why the experiment at PSI is so valuable for us."

But the successful publication in Nature Communications is not the end



of the story. The researchers want to conduct more experiments in order to understand how the classic buckyballs form in the universe, along with the football-shaped fullerene molecules with 60 carbon atoms and the minute nanotubes with even more atoms.

More information: Lotefa B. Tuli et al, Gas phase synthesis of the C_{40} nano bowl $C_{40}H_{10}$, *Nature Communications* (2023). DOI: <u>10.1038/s41467-023-37058-y</u>

Provided by Paul Scherrer Institute

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