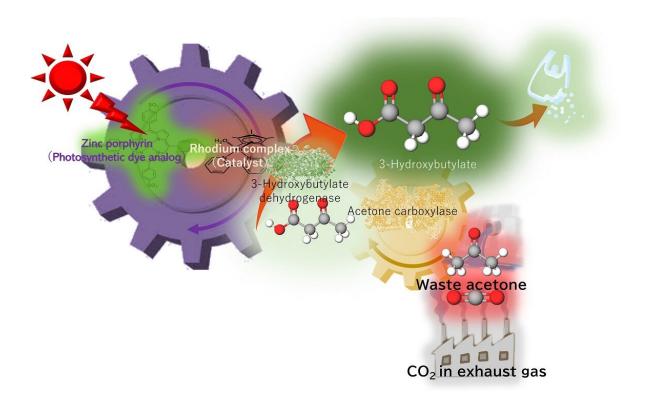


## Low concentration CO2 can be reused in biodegradable plastic precursor using artificial photosynthesis

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Using energy from light equivalent to sunlight the artificial photosynthesis system uses enzymes and a rhodium catalyst to produce a biodegradable plastic precursor. Now for the first time, the process works using low concentrations of  $CO_2$ , similar to exhaust gas, and waste acetone as raw materials. Credit: Yutaka Amao, OMU



Osaka Metropolitan University scientists have developed a process using artificial photosynthesis to successfully convert more than 60% of waste acetone into 3-hydroxybutyrate, a material used to manufacture biodegradable plastic. The results were obtained using low-concentration  $CO_2$ , equivalent to exhaust gas, and powered by light equivalent to sunlight for 24 hours.

The researchers expect that this innovative way of producing biodegradable plastic could not only reduce  $CO_2$  emissions but also provide a way of reusing laboratory and industrial waste acetone. Their findings have been published in the journal *Green Chemistry*.

Poly-3-hydroxybutyrate—a biodegradable plastic—is a strong waterresistant polyester often used in packaging materials, made from 3-hydroxybutyrate as a precursor. In previous studies, a research team led by Professor Yutaka Amao from the Research Center for Artificial Photosynthesis at Osaka Metropolitan University found that 3-hydroxybutyrate can be synthesized from  $CO_2$  and acetone with high efficiency, but this was only demonstrated at higher concentrations of  $CO_2$  or sodium bicarbonate.

This new study aimed to reuse waste acetone from permanent marker ink and low concentrations of  $CO_2$ —equivalent to <u>exhaust gas</u> from <u>power plants</u>, chemical plants, or steel factories. Acetone is a relatively inexpensive and reasonably harmless chemical used in many different laboratory settings, either for reactions or as a cleaning agent, which produces waste acetone. The acetone and  $CO_2$  acted as <u>raw materials</u> to synthesize 3-hydroxybutyrate using <u>artificial photosynthesis</u>, powered by light equivalent to sunlight.

"We focused our attention on the importance of using  $CO_2$  created by exhaust gas from thermal power plants and other sources to demonstrate the practical application of artificial photosynthesis," explained



Professor Amao.

After 24 hours, more than 60% of acetone had been successfully converted to 3-hydroxybutyrate. "In the future, we aim to develop artificial photosynthesis technology further, so that it can use acetone from liquid waste and as well as exhaust gas from the laboratory as raw materials," stated Professor Amao.

**More information:** Yu Kita et al, Visible-light-driven 3-hydroxybutyrate production from acetone and low concentrations of CO2 with a system of hybridized photocatalytic NADH regeneration and multi-biocatalysts, *Green Chemistry* (2023). <u>DOI:</u> 10.1039/D3GC00247K

Provided by Osaka Metropolitan University

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