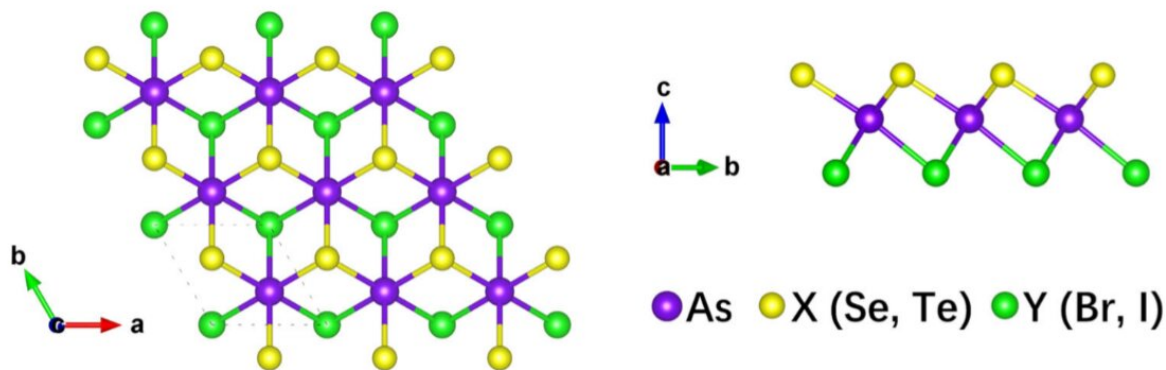


2D Janus materials could harvest abundant hydrogen fuel

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Top and side view of the optimized atomic structure of Janus AsXY (X = Se, Te; Y = Br, I) monolayers. Credit: *The European Physical Journal B* (2023). DOI: 10.1140/epjb/s10051-023-00486-2

Several studies have predicted that the water splitting reaction could be catalyzed by certain groups of 2D materials—each measuring just a few atoms thick. One particularly promising group are named 2D Janus materials, whose two sides each feature a different molecular composition.

Through new calculations detailed in *The European Physical Journal B*, Junfeng Ren and colleagues at Shandong Normal University in China

present a new group of four 2D Janus materials, which could be especially well suited to the task.

Since hydrogen releases an abundance of energy when combusted, with only water as a byproduct, it is now widely seen as an excellent alternative to fossil fuels. Splitting [water molecules](#) involves a [redox reaction](#), where [electrons](#) and holes participate in reduction and oxidation reactions.

Since they are excellent semiconductors, 2D Janus materials are particularly well suited to catalyzing this reaction. When an electron in a semiconductor's insulating [valence band](#) absorbs a photon, it is excited to the material's [conduction band](#), leaving behind a positively charged hole. In turn, these materials act as both source and acceptors of electrons—allowing redox reactions to occur more readily.

In their theoretical study, Ren's team examined a group of four of these materials: with one surface composed of either selenium or tellurium, and the other from either bromine or iodine—with both sides sandwiching a middle layer of arsenic. In these semiconductors, the energies of their valence and conduction bands were far enough apart to prevent electrons and holes from readily recombining: allowing them to combine electrons and holes to produce hydrogen and oxygen.

With all four materials displaying excellent stability and light absorption, the researchers believe they could be incredibly promising candidates for catalyzing the water splitting reaction. If these results can be reproduced in experiments, Ren's team hope the four materials could become a key element of the global effort to eliminate our [carbon emissions](#) in the next few decades.

More information: Jiali Wang et al, Two-dimensional Janus AsXY (X = Se, Te; Y = Br, I) monolayers for photocatalytic water splitting, *The*

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