

A nanofiltration membrane with asymmetric selectivity toward enhanced water recovery



Figure 1. Schematic diagrams of (A) concurrent rejections of Ca^{2+} , SO_4^{2-} and PFOS with severe membrane scaling and (B) asymmetric rejections of Ca^{2+} , SO_4^{2-} and PFOS with reduced membrane scaling. Credit: Peking University

Groundwater contamination by emerging pollutants has drawn great concern in recent years. Emerging contaminants such as perfluoroalkyl substances (PFASs), especially perfluorooctanesulfonate (PFOS), were found to be widespread and persistent in groundwater systems due to their high chemical resistance to environmental degradation processes.

Nanofiltration (NF) has been applied as an effective separation technology to remove PFOS from groundwater. Despite the high removal efficiency of PFOS, scaling of NF membranes limits the water recovery during remediation. Furthermore, groundwater usually contains high levels of divalent ions such as <u>calcium</u>, magnesium, and sulfate that



are often well rejected by conventional NF membranes (see Fig. 1A). This will lead to severe <u>membrane</u> scaling (e.g., by <u>calcium sulfate</u> dihydrate, gypsum) and deplete the mineral content of the treated water, which further convert the water more corrosive and less suitable for drinking purposes.

To overcome the selectivity limitation of conventional NF membranes, a team from the College of Environmental Sciences and Engineering (CESE) of Peking University found a simple NaOH-promoted interfacial polymerization strategy that was proposed to tailor the membrane asymmetric selectivity of calcium over sulfate ions to reduce scaling potential towards high water recovery. The calcium can freely pass through the asymmetrically selective membrane while <u>sulfate</u> and PFOS are adequately removed (Fig. 1B).

This promoted interfacial polymerization reduced polyamide defects for better rejection and enabled the use of lower PIP concentrations. Enhanced hydrolysis of polyamide layer created a more negatively charged surface with larger pore sizes to achieve asymmetrical selectivity together with enhanced permeability.





Figure 2. (A) Pure water permeance, (B) Na₂SO₄, CaCl₂ and PFOS rejections, (C) Water flux decline curves for TFC–0.1 and NF270 membranes during scaling tests at different recoveries. Credit: Peking University

The water permeability was 2.1 times of that for commercial NF270 membrane, while exhibiting comparable perfluorooctanesulfonic acid rejections (>95%) (Fig. 2A and 2B). This membrane also achieved ~50% more water recovery than that for NF270 membrane under



simulated gypsum scaling condition, due to the selective passage of calcium (low rejection of $11.4 \pm 0.6\%$) (Fig. 2C).

The exemplified strategy of NaOH-promoted interfacial polymerization is facile and readily scalable, which demonstrated strong potential towards high water recovery in groundwater remediation contaminated by emerging pollutants.

More information: Wulin Yang et al, Facile synthesis of nanofiltration membrane with asymmetric selectivity towards enhanced water recovery for groundwater remediation, *Journal of Membrane Science* (2022). DOI: 10.1016/j.memsci.2022.121038

Provided by Peking University

Citation: A nanofiltration membrane with asymmetric selectivity toward enhanced water recovery (2022, December 7) retrieved 7 May 2024 from <u>https://phys.org/news/2022-12-nanofiltration-membrane-asymmetric-recovery.html</u>

This document is subject to copyright. Apart from any fair dealing for the purpose of private study or research, no part may be reproduced without the written permission. The content is provided for information purposes only.