

Water cleanup method destroys pervasive, cancer-causing 'forever chemicals'

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Graphical abstract. Credit: *Journal of Hazardous Materials Letters* (2022). DOI: 10.1016/j.hazl.2022.100072



An insidious category of carcinogenic pollutants known as "forever chemicals" may not be so permanent after all.

University of California, Riverside, chemical engineering and <u>environmental scientists</u> recently published new methods to chemically break up these <u>harmful substances</u> found in drinking water into smaller compounds that are essentially harmless.

The patent-pending process infuses <u>contaminated water</u> with hydrogen, then blasts the water with high-energy, short-wavelength ultraviolet light. The hydrogen polarizes <u>water molecules</u> to make them more reactive, while the light catalyzes <u>chemical reactions</u> that destroy the pollutants, known as PFAS or poly- and per-fluoroalkyl substances.

This one-two punch breaks the strong fluorine-to-carbon chemicals bonds that make these pollutants so persistent and accumulative in the environment. In fact, the molecular destruction of PFAS increased from 10% to nearly 100% when compared to other ultraviolet water-treatment methods, while no other undesirable byproducts or impurities are generated, the UCR scientists reported in a paper recently published in the *Journal of Hazardous Materials Letters*.

What's more, the cleanup technology is green.

"After the interaction, hydrogen will become water. The advantage of this technology is that it is very sustainable," said Haizhou Liu, an associate professor in UCR's Department of Chemical and Environmental Engineering and the corresponding author of the paper.

PFAS are a family of thousands of chemical compounds characterized by fully fluorinated carbon atoms with stubbornly strong chemical bonds



that last indefinitely in the environment—hence the moniker "forever chemicals." These compounds came into widespread use in thousands of consumer products starting in the 1940s because of their ability to resist heat, water, and lipids.

Examples of PFAS-containing products include grease-resistant paper wrappers and containers such as microwave popcorn bags, pizza boxes, and candy wrappers. They are also found in stain and water repellents used on carpets, upholstery, clothing; cleaning products; non-stick cookware; and paints, varnishes, and sealants, according to the U.S. Environmental Protection Agency.

Studies have linked exposure to certain levels of PFAS to many ill <u>health</u> <u>effects</u>, including increased risk for prostate, kidney, and testicular cancers, as well as decreased fertility or increased <u>high blood pressure</u> in <u>pregnant women</u>, developmental effects or delays in children, low birth weight, and accelerated puberty, according to the EPA.

Because of these health effects, federal and <u>state officials</u> are promulgating new cleanup standards for PFAS in drinking water and in groundwater below or emanating from toxic cleanup sites.

The technology developed by Liu's team involves emitting shortwavelength ultraviolet light into treatment tanks to excite water molecules.

"We are optimizing it by trying to make this technology versatile for a wide range of PFAS-contaminated source waters," Liu said. "The technology has shown very promising results in the destruction of PFAS in both drinking water and different types of industrial wastewater."

More information: Gongde Chen et al, Hydrogen-polarized vacuum ultraviolet photolysis system for enhanced destruction of perfluoroalkyl



substances, Journal of Hazardous Materials Letters (2022). DOI: 10.1016/j.hazl.2022.100072

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