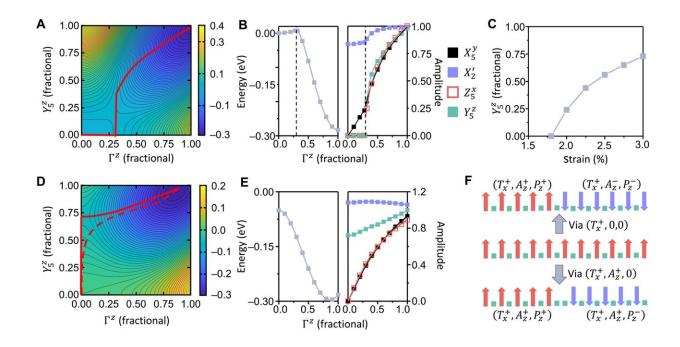


Stable, faster computer memory storage at the nanometer scale

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Strain stabilization of antipolar and ferroelectric phases of HfO_2 . (A) The energy of HfO_2 as a function of polar and antipolar modes under zero strain (tetragonal lattice parameters; all other modes are relaxed) shows that the tetragonal phase is metastable. (The mode amplitudes are normalized to the equilibrium orthorhombic phase.) (B) As the polar mode amplitude is increased [at zero strain; see red solid line in (A)], energy rises until an energy cusp [saddle point in (A)] is reached, beyond which large antipolar and polar mode amplitudes coexist. (C) For tensile strain ε_{xx} of > 1.9%, the tetragonal phase is no longer metastable, and the equilibrium antipolar mode amplitude at $\Gamma^z = 0$ is shown. (D) Under 3% tensile strain, the lowest-energy path from tetragonal to orthorhombic phase (red dashed line) is shown. The antipolar mode is unstable even at P = 0 (leading to the Pbcn phase), and after the antipolar mode amplitude increases



sufficiently, the polar mode then becomes unstable, leading to the orthorhombic AP + FE phase. (E) As the polar mode amplitude is increased from zero (at 3% tensile strain), the energy decreases to the equilibrium AP + FE phase. (F) Switching an up-polarized orthorhombic variant (T_x^+, A_z^+, P_z^+) to the down-polarized variants (T_x^+, A_z^+, P_z^-) or (T_x^+, A_z^-, P_z^-) could generate domain walls consisting of two (fast-switching domain wall) or one (slow-switching domain wall) nonpolar layer(s), respectively. These two switching paths will pass through the Pbcn and the tetragonal phases, respectively. The second switching path is not energetically favorable, because the stability of antipolar mode requires additional energy to switch antipolar and polar mode together. Credit: *Science Advances* (2022). DOI: 10.1126/sciadv.add5953

Unlike in humans, when computer "brains" evolve, they get smaller and smaller. This is because the components that perform calculations and consolidate stored information work more efficiently when there are more of them tightly packed on a chip.

Yet when the chip feature sizes get too small, say, to the <u>nanometer scale</u>, their physical and <u>material properties</u> can change, rendering them less reliable at doing their jobs. In the last decade, scientists have made great strides in uncovering new substances that instead become increasingly stable as they scale down, hinting at the promise of smaller storage devices that can be integrated onto silicon computer processing units (CPUs) to increase speed and functionality.

One such compound is hafnium dioxide (HfO₂), a material that was found to retain a desirable property, known as ferroelectricity, even at the few-nanometer scale (~2nm). When a ferroelectric material is exposed to a strong enough external electric field, it becomes strongly electrically polarized, which is a state where the material has plus-minus charge dipoles in alignment. What's great about <u>ferroelectric materials</u> is that this polarization persists, even if the external electric field is



removed, analogous to how an iron nail can become permanently magnetized. This persistent polarization means that the material remembers the last direction it was electrically polarized.

What makes HfO₂ special is that it can rapidly switch between an up or down mode—corresponding to the ones and zeroes computers use—at reduced dimensions and then retain this information until it is switched again. But how it's able to achieve this feat has remained a mystery.

Now, a group of researchers led by Andrew M. Rappe, the Blanchard Professor of Chemistry in the School of Arts & Sciences, has uncovered how HfO₂ retains its ferroelectric <u>phase</u> in these conditions and explains how it remains stable.

Their research, published in *Science Advances*, details how HfO₂ undergoes a two-step transition resulting in a change in the arrangement of its atoms when grown on a thin film. This allows it to "transition from one phase, which isn't very useful, to a special one that could be useful for the next generation of information storage devices," says co-first author of the paper, Songsong Zhou, a postdoctoral researcher in the School of Arts & Sciences.

"The popular belief explaining the mechanism of this phase transition was that it was either a simple single proper phase transition, or a rare and complicated improper phase transition," says Zhou. "However, we were able to present a third alternative: Tension from being grown on a thin film and an unconventional change in HfO₂'s polarization state are linked together to drive a wholly new reaction that induces an antiferroelectric state that actually stabilizes HfO₂'s ferroelectric state."

The ability to have a material be both ferroelectric and antiferroelectric was a major surprise finding. The researchers were under the impression that these were competing states because antiferroelectric materials have



their charges alternate between up and down, as opposed to the unidirectional ferroelectric charges. "Our model presents a new framework for understanding phase transitions for materials capable of retaining polarization states at the nanometer scale," says co-first author of the paper, Jiahao Zhang, a sixth-year Ph.D. student in the chemistry department.

"HfO₂ and a few other materials are competing to become successful computer memory materials, but all of them currently have problems," says Rappe. "In offering a greater insight into the mechanism of ferroelectricity in HfO₂, our work addresses some of these issues and paves the way for developing the next generation of materials that could someday soon integrate both processing and memory onto a single chip."

Next, the researchers will build on their models as they continuously merge experimental and theoretical insights to harness the nanomaterials world.

More information: Songsong Zhou et al, Strain-induced antipolar phase in hafnia stabilizes robust thin-film ferroelectricity, *Science Advances* (2022). DOI: 10.1126/sciadv.add5953

Provided by University of Pennsylvania

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