

Researchers publish 31,618 molecules with potential for energy storage in batteries

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Artist's impression of DIFFER's research on 31,618 molecules with potential for energy storage in redox flow batteries. The researchers used artificial intelligence and quantum chemical methods on supercomputers to predict approximately 300 properties per molecule. Credit: DIFFER/Süleyman Er

Scientists from the Dutch Institute for Fundamental Energy Research (DIFFER) have created a database of 31,618 molecules that could potentially be used in future redox-flow batteries. These batteries hold great promise for energy storage. Among other things, the researchers used artificial intelligence and supercomputers to identify the molecules' properties. Today, they publish their findings in the journal *Scientific Data*.

In recent years, chemists have designed hundreds of molecules that could potentially be useful in flow batteries for energy storage. It would be wonderful, researchers from DIFFER in Eindhoven (the Netherlands) imagined, if the properties of these molecules were quickly and easily accessible in a database. The problem, however, is that for many molecules the properties are not known. Examples of molecular properties are redox potential and water solubility. Those are important since they are related to the power generation capability and energy density of redox flow batteries.

To find out the still-unknown properties of molecules, the researchers performed four steps. First, they used a [desktop computer](#) and smart algorithms to create thousands of virtual variants of two types of molecules. These molecule families, the quinones and aza aromatics, are good at reversibly accepting and donating electrons. That is important for batteries. The researchers fed the computer with backbone structures of 24 quinones and 28 aza-aromatics plus five different chemically relevant side groups. From that, the computer created 31,618 different molecules.

In the second step, the researchers used supercomputers to calculate nearly 300 different properties of each molecule. The computer uses equations from quantum chemistry to do this. Because these formulas are difficult to solve, a powerful supercomputer is a handy tool.

In the third step, the researchers used [machine learning](#) to predict whether the molecules would be dissolvable in water.

The fourth and final step consisted of creating a both human- and machine-readable database. The database, called RedDB (from Redox DataBase), contains the molecules and their properties with convenient naming and description.

"When you work with [theoretical models](#) and machine learning, you obviously want to be confident in the results," says Süleyman Er, the leader of DIFFER's Autonomous Energy Materials Discovery research group. "This is why we used computer programs that have proven their excellence. For this purpose, we also implemented dedicated validation procedures."

Now that the database is public, researchers, including those outside DIFFER, can easily search for potentially interesting molecules for redox flow batteries. For instance, they can simply purchase or synthesize the molecules and research them further. Moreover, the researchers may use the [database](#) to improve their machine-learning models to speed up the design of high-quality [molecules](#) for [energy storage](#).

More information: Elif Sorkun et al, RedDB, a computational database of electroactive molecules for aqueous redox flow batteries, *Scientific Data* (2022). [DOI: 10.1038/s41597-022-01832-2](https://doi.org/10.1038/s41597-022-01832-2)

Provided by DIFFER

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