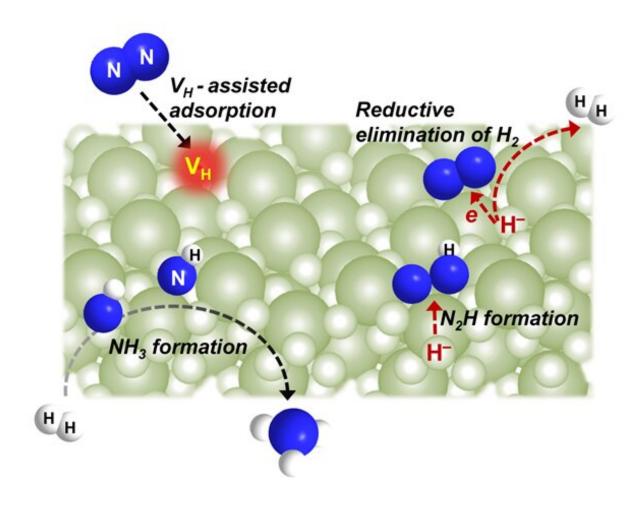


Transition-metal—free barium hydride mediates dinitrogen fixation and ammonia synthesis

September 5 2022, by Li Yuan



Transition-metal-free dinitrogen fixation mediated by barium hydride. Credit: Guan Yeqin



Ammonia is crucial for the manufacture of nitrogen fertilizers. Due to the high energy consumption of industrial ammonia production, the development of alternative materials and approaches for efficient N_2 reduction to ammonia under mild conditions is a long-sought goal.

Recently, a research group led by Prof. Chen Ping and Prof. Guo Jianping from the Dalian Institute of Chemical Physics (DICP) of the Chinese Academy of Sciences, in collaboration with Prof. Tejs Vegge from Technical University of Denmark, synthesized <u>ammonia</u> via a chemical looping process mediated by a transition-metal-free barium <u>hydride</u> (BaH₂) and revealed its mechanism.

This study was published in *Angewandte Chemie International Edition* on August 2.

Alkali or alkaline earth metal hydrides can fix N_2 , forming corresponding metal imides and H_2 . The metal imides then undergo hydrogenation to NH_3 and metal hydrides. However, the reaction mechanisms of N_2 activation, H_2 release and NH_3 formation over alkali hydrides are still unclear.

The researchers indicated that the creation of hydrogen vacancies played an important role in N_2 fixation process mediated by BaH_2 .

The creation of hydrogen vacancies led to the formation of multiple coordinatively unsaturated Ba sites, which were responsible for the adsorption and activation of N_2 . The hydridic hydrogen acted as an electron donor and facilitated the activation of N_2 with concurrent H_2 release.

They found that the process functionally resembled the molecular hydrido complexes and FeMo cofactor in nitrogenase. Both the hydridic hydrogen and gaseous H₂ were involved in the NH₃ formation process.



"This is a helpful model for understanding the activation and hydrogenation of N_2 to NH_3 mediated by alkali and alkaline earth metal hydrides, which is promising in future technologies for <u>nitrogen fixation</u> using transition-metal-free materials," said Prof. Chen.

More information: Yeqin Guan et al, Transition-Metal-Free Barium Hydride Mediates Dinitrogen Fixation and Ammonia Synthesis, *Angewandte Chemie International Edition* (2022). DOI: 10.1002/anie.202205805

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