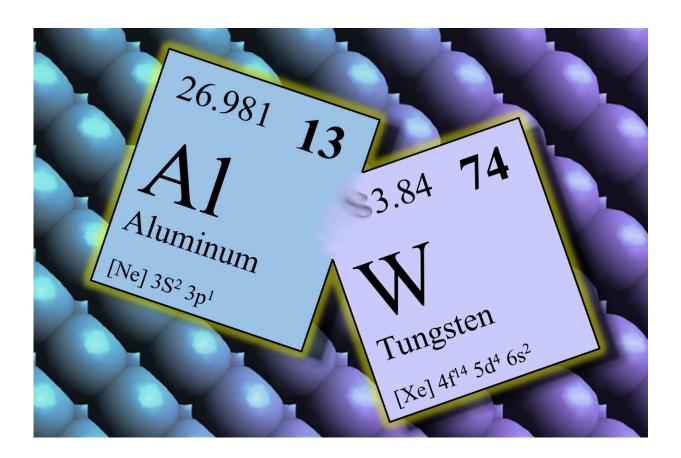


Energy researchers invent chameleon metal that acts like many others

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University of Minnesota researchers have invented a "catalytic condenser" that opens the door for new catalytic technologies using non-precious metal catalysts for important applications such as storing renewable energy, making renewable fuels, and manufacturing sustainable materials. Credit: Dauenhauer Group, University of Minnesota



A team of energy researchers led by the University of Minnesota Twin Cities have invented a device that electronically converts one metal so that it behaves like another for use as a catalyst in chemical reactions. The device, called a "catalytic condenser," is the first to demonstrate that alternative materials that are electronically modified to provide new properties can yield faster, more efficient chemical processing.

The invention opens the door for new catalytic technologies using nonprecious metal catalysts for important applications such as storing <u>renewable energy</u>, making renewable fuels, and manufacturing <u>sustainable materials</u>.

The research is published online in *JACS Au*, where it was selected as an Editor's Choice publication. The team is also working with the University of Minnesota Office of Technology Commercialization and has a provisional patent on the device.

Chemical processing for the last century has relied on the use of specific materials to promote the manufacturing of chemicals and materials we use in our everyday lives. Many of these materials, such as precious metals ruthenium, platinum, rhodium, and palladium, have unique electronic surface properties. They can act as both metals and metal oxides, making them critical for controlling <u>chemical reactions</u>.

The <u>general public</u> is probably most familiar with this concept in relation to the uptick in thefts of <u>catalytic converters</u> on cars. Catalytic converters are valuable because of the rhodium and palladium inside them. In fact, palladium can be more expensive than gold.

These expensive materials are often in short supply around the world and have become a major barrier to advancing technology.

In order to develop this method for tuning the catalytic properties of



<u>alternative materials</u>, the researchers relied on their knowledge of how electrons behave at surfaces. The team successfully tested a theory that adding and removing electrons to one material could turn the metal oxide into something that mimicked the properties of another.

"Atoms really do not want to change their number of electrons, but we invented the catalytic condenser device that allows us to tune the number of electrons at the surface of the catalyst," said Paul Dauenhauer, a MacArthur Fellow and professor of chemical engineering and materials science at the University of Minnesota who led the research team. "This opens up an entirely new opportunity for controlling chemistry and making abundant materials act like precious materials."

The catalytic condenser device uses a combination of nanometer films to move and stabilize electrons at the surface of the catalyst. This design has the unique mechanism of combining metals and metal oxides with graphene to enable fast electron flow with surfaces that are tunable for chemistry.

"Using various thin film technologies, we combined a nano-scale film of alumina made from low-cost abundant aluminum <u>metal</u> with graphene, which we were then able to tune to take on the properties of other materials," said Tzia Ming Onn, a post-doctoral researcher at the University of Minnesota who fabricated and tested the catalytic condensers. "The substantial ability to tune the catalytic and electronic properties of the catalyst exceeded our expectations."

The catalytic condenser design has broad utility as a platform device for a range of manufacturing applications. This versatility comes from its nanometer fabrication that incorporates graphene as an enabling component of the active surface layer. The power of the device to stabilize electrons (or the absence of electrons called "holes") is tunable with varying composition of a strongly insulating internal layer. The



device's active layer also can incorporate any base catalyst material with additional additives, that can then be tuned to achieve the properties of expensive catalytic materials.

"We view the catalytic condenser as a platform technology that can be implemented across a host of manufacturing applications," said Dan Frisbie, a professor and head of the University of Minnesota Department of Chemical Engineering and Materials Science and research team member. "The core design insights and novel components can be modified to almost any chemistry we can imagine."

The team plans to continue their research on catalytic condensers by applying it to precious metals for some of the most important sustainability and environmental problems. Several parallel projects are already in progress to store renewable electricity as ammonia, manufacture the key molecules in renewable plastics, and clean gaseous waste streams.

More information: Tzia Ming Onn et al, Alumina Graphene Catalytic Condenser for Programmable Solid Acids, *JACS Au* (2022). <u>DOI:</u> <u>10.1021/jacsau.2c00114</u>

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