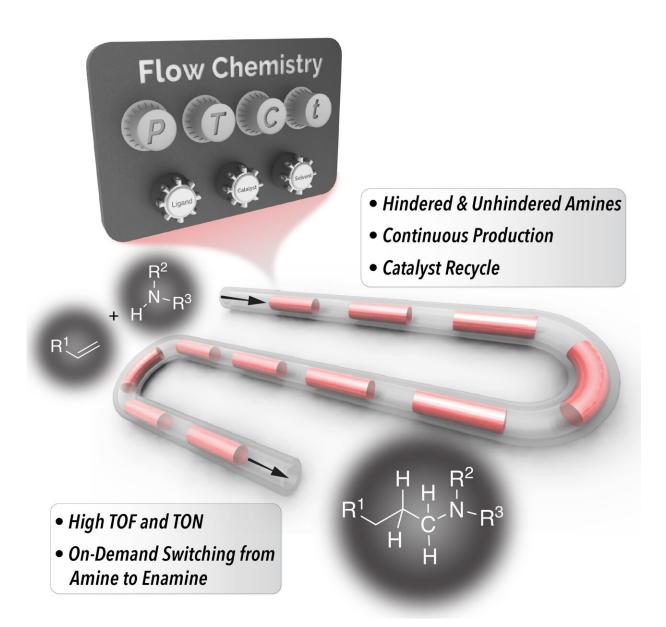


Researchers devise cheaper, faster way to continuously produce amines

May 4 2022, by Matt Shipman



Researchers at North Carolina State University have developed a faster, less



expensive technique for producing hindered amines – a class of chemicals used as building blocks in products ranging from pharmaceuticals and agrochemicals to detergents and organic light emitting diodes. The success of the new technique is made possible by two things. First, by using a continuous flow reactor that allows for continuous flow of both gases and liquids in a segmented flow format, the researchers were able to make the kinetics of the reaction far more efficient. Second, the new technique makes use of a co-catalyst – fluorinated benzoic acid – which reduces the amount of energy needed to perform some of the necessary reactions in the process. Credit: Milad Abolhasani, NC State University

Researchers at North Carolina State University have developed a faster, less expensive technique for producing hindered amines—a class of chemicals used as building blocks in products ranging from pharmaceuticals and agrochemicals to detergents and organic light emitting diodes.

"Hindered <u>amines</u> are used in a tremendous variety of products, but all of the existing techniques for producing these amines are complicated and expensive," says Milad Abolhasani, corresponding author of a paper on the new technique and an associate professor of chemical and biomolecular engineering at NC State. "We set out to develop a better method for synthesizing these hindered amines, and we were successful."

One of the less expensive techniques for producing hindered amines is hydroaminomethylation, or HAM. However, the <u>chemical industry</u> has largely avoided using HAM, because there are too many ways things can go wrong—leaving producers with undesirable chemicals instead of the functionalized amines they were trying to make. Researchers have improved the HAM process over the years. But all of the techniques for avoiding undesirable byproducts have meant extending the timeframe of the HAM process, so that it takes hours to perform all of the necessary reactions. Until now.



"We've developed a HAM technique that makes use of continuous <u>flow</u> reactor technologies to produce hindered amines more efficiently," Abolhasani says. "Our HAM process takes less than 30 minutes in most cases. The only products are hindered amines and water. And we are able to recycle the primary catalyst, rhodium/N-Xantphos, which further drives down costs."

The success of the new technique is made possible by two things. First, by using a continuous flow reactor that allows for continuous flow of both gases and liquids in a segmented flow format, the researchers were able to make the kinetics of the reaction far more efficient. Second, the new technique makes use of a co-catalyst—fluorinated benzoic acid—which reduces the amount of energy needed to perform some of the necessary reactions in the HAM process.

Ultimately, this technique drives down the cost of producing hindered amines using inexpensive feedstock, allowing users to produce them more quickly and with no toxic byproducts.

"By designing a cooperative catalyst system, we've demonstrated that the rate of the HAM reactions in our system can be 70 times higher than the existing state-of-the-art processes," says Malek Ibrahim, first author of the paper and a former postdoctoral researcher at NC State. "This process is also a good example for how flow chemistry platforms can improve catalyst turnover frequency, which is increasingly important as the price of rhodium catalysts goes up."

The new technique is particularly attractive for decentralized manufacturing operations, since the small footprint of the necessary equipment and its scalability allows users to efficiently produce hindered amines on site and on demand.

"What's more, the same technique can also be used to produce



enamines—which are other chemical <u>building blocks</u>—on demand, simply by tuning the solvents we use in the flow reactor," Ibrahim says. "You can literally switch back and forth between producing amines and enamines without having to stop the <u>production process</u>, since the only thing you're changing is the solvent mixture."

The researchers have filed a provisional patent on the new technique and are now looking for industrial partners to put the technique into widespread use.

The paper, "Recyclable Cooperative Catalyst for Accelerated Hydroaminomethylation of Hindered Amines in a Continuous Segmented Flow Reactor," will be published May 4 in the journal *Nature Communications*.

More information: Recyclable Cooperative Catalyst for Accelerated Hydroaminomethylation of Hindered Amines in a Continuous Segmented Flow Reactor, *Nature Communications* (2022). <u>DOI:</u> <u>10.1038/s41467-022-30175-0</u>

Provided by North Carolina State University

Citation: Researchers devise cheaper, faster way to continuously produce amines (2022, May 4) retrieved 28 April 2024 from <u>https://phys.org/news/2022-05-cheaper-faster-amines.html</u>

This document is subject to copyright. Apart from any fair dealing for the purpose of private study or research, no part may be reproduced without the written permission. The content is provided for information purposes only.