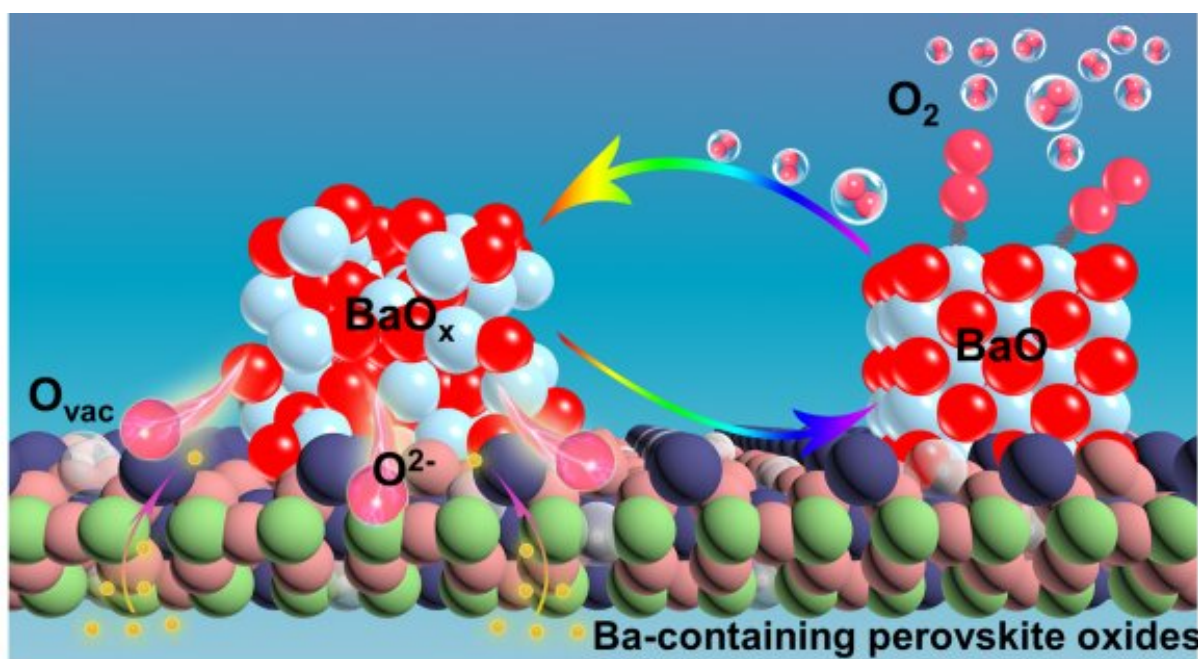


Mechanism of oxygen activation on barium-containing perovskite materials

April 25 2022, by Li Yuan



Oxygen activation mechanism on Ba-containing perovskite oxides. Credit: Zhu Yue

A research team led by Prof. Yang Weishen and Prof. Zhu Xuefeng from the Dalian Institute of Chemical Physics (DICP) of the Chinese Academy of Sciences (CAS) has revealed the mechanism of oxygen activation on Barium-containing perovskite materials.

The researchers discovered that BaO/BaO₂ nanoparticles precipitated on the surface of Ba-containing materials under high-temperature oxygen-rich conditions had an ultra-high activity for oxygen activation, which clarified the mechanism of high-temperature oxygen activation and transport on the surface of Ba-containing perovskite oxides.

This study was published in *Science Advances* on April 13.

In 2000, the DICP team invented an oxygen-permeable membrane material named Ba_{0.5}Sr_{0.5}Co_{0.8}Fe_{0.2}O_{3-δ} (BSCF). Due to its good catalytic activity towards oxygen activation, BSCF has become a representative material for oxygen permeation and has been widely used in solid oxide fuel cells, oxygen reduction reactions, and oxygen evolution reactions.

However, the essence of the good performance of the BSCF perovskite is still unclear.

In this study, the researchers analyzed the oxygen permeation process and found that the addition of Ba into perovskite oxides could accelerate the oxygen surface exchange reaction kinetics.

They identified the precipitation of BaO_x nanoparticles on the surface of BSCF materials in a high-temperature oxygen atmosphere by environmental electron microscopy, and proved that Ba-containing materials that could be precipitated or decomposed into BaO_x have high catalytic activity for the oxygen activation process.

Moreover, combined with DFT calculation, they found that the precipitated BaO_x nanoparticles could reduce the energy barriers of oxygen molecule adsorption and dissociation in the oxygen reduction and oxygen desorption of the oxygen evolution process, thus accelerating the oxygen exchange reaction kinetics at the gas-solid interface.

"This study provides a scientific basis for the design of [oxygen](#)-permeable membranes and electrocatalytic [materials](#)," said Prof. Yang.

More information: Yue Zhu et al, Oxygen activation on Ba-containing perovskite materials, *Science Advances* (2022). [DOI: 10.1126/sciadv.abn4072](#)

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