

## **Researchers reveal degradation mechanism** of non-precious metal catalysts for fuel cells

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Graphical abstract. Credit: *Applied Catalysis B: Environmental* (2022). DOI: 10.1016/j.apcatb.2022.121290

The development of efficient non-precious metal catalysts (NPMCs) can help reduce the cost of fuel cells and accelerate their commercialization.

At present, non-precious metal catalysts suffer from poor stability in proton exchange membrane fuel cells, and their degradation mechanism at the molecular scale is unclear.

Recently, researchers led by Profs. Sun Gongquan, Wang Suli and Wang Junhu from the Dalian Institute of Chemical Physics (DICP) of the



Chinese Academy of Sciences (CAS) revealed the degradation mechanism of Fe-N-C-type NPMCs.

The study was published in *Applied Catalysis B: Environmental* on March 4.

The researchers found that D1 was the main active site for <u>oxygen</u> reduction reaction and the demetallation was responsible for the degradation of Fe-N-C. The degradation was initially rapid but slow afterward, which could be explained by the different activity and stability of various  $FeN_4$  sites.

Moreover, they revealed that the adsorption of oxygenated intermediate and the motivation of electric field could further weaken the Fe-N bond in FeN<sub>4</sub>, leading to more severe demetallation. The faster degradation of Fe-N-C under higher potential was due to the stronger <u>adsorption</u> and field intensity.

This work provides insight into the degradation mechanism of Fe-N-C at the <u>molecular level</u>. It can work as the guidance for future development of stable NPMCs in <u>fuel cell</u> devices.

**More information:** Xinlong Xu et al, Investigation on the demetallation of Fe-N-C for oxygen reduction reaction: The influence of structure and structural evolution of active site, *Applied Catalysis B: Environmental* (2022). DOI: 10.1016/j.apcatb.2022.121290

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