

Corrosion can improve materials' durability

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When it comes to the integrity of structural alloys, a little corrosion may sometimes be a good thing.



Cornell researchers used advanced atomic modeling to explore the ways environment can influence the growth of cracks in alloys such as aluminum and steel—knowledge that could help engineers better predict, and possibly postpone, the failure of structures. And by removing atoms from the tip of a crack, the modeling showed the researchers could prevent a crack from propagating, essentially improving the material's mechanical performance.

The team's paper, "Dissolution at a Ductile Crack Tip," published Oct. 1 in *Physical Review Letters*. The lead author is Wenjia Gu, Ph.D. '20.

"People have been modeling crack growth and fracture for a long time, but the actual process by which it occurs has not really been clear, at least for structural alloys in complicated environments," said Derek Warner, associate professor of civil and <u>environmental engineering</u> and the paper's senior author. "It can be a very large-scale phenomenon—big structures can fracture—but it can be controlled at the <u>atomic scale</u>, particularly when you look at environmental effects."

An environment has many different mechanisms by which it adversely impacts a material, among them dissolution, oxide formation, material redeposition and hydrogen embrittlement. Warner and his team in the Cornell Fracture Group chose to focus on dissolution, which can be found everywhere from corroded <u>metal</u> surfaces to eroded <u>human bone</u>.

The team created a series of atomic 2D models of a structural alloy, similar to aluminum and steel, that was ductile—i.e., pliable enough that it that wouldn't shatter when deformed, as glass does.

By running numerous simulations that stressed the material with a range of loading cycles, the researchers were able to see the different ways the atoms interacted. The researchers then began removing loosely bonded atoms from the surface, one at a time, and monitored the crack's



behavior.

They found that removing surface material inhibited the crack from growing.

"The proclivity of a crack to grow depends on how sharp it is," Warner said. "If you have a big round notch, it's unlikely to propagate like a crack. But if you have some sharp feature, like a slit cut with a knife, it is more likely to grow. So in this way, material removal, akin to what occurs during corrosion, can actually improve mechanical performance."

There is a corollary to this type of destruction-as-improvement in human biology, Warner noted. Osteoclasts, a type of bone cell, dissolve bone tissue as a way to promote bone growth and resist fracturing.

This approach could have plenty of practical applications, just by letting nature take its course.

"There are some situations where you would have an engineered structure, a structural alloy, and you could say well, it might actually be beneficial to let it corrode a little bit because it can blunt the cracks that are there already," Warner said.

The research is of particular interest for the Office of Naval Research, which funded the study, and its efforts to keep expensive aircraft in safe working condition amid the extreme ocean environment.

"When an aircraft lands on a carrier, that's within 30% of what you would call a crash landing every time it lands. You're operating in these tight margins. Then you let the thing sit on the deck of the aircraft carrier in the sun, getting whipped with salt water and corrosion," Warner said. "You know what happens if you leave your bike outside, right? With better modeling, they can make a better assessment of



whether it's safe to fly, and how often do we need to do maintenance to look for problems."

More information: Wenjia Gu et al, Dissolution at a Ductile Crack Tip, *Physical Review Letters* (2021). DOI: 10.1103/PhysRevLett.127.146001

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