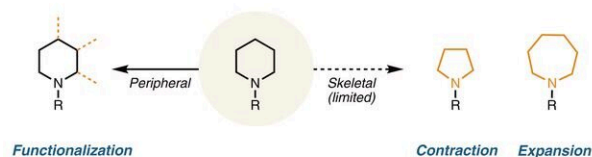


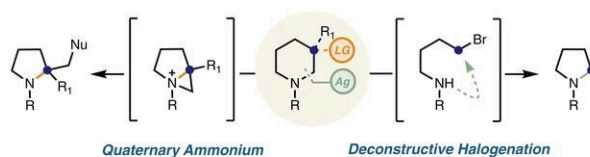
A reaction that removes sulfur, nitrogen or oxygen atoms from six-membered rings using only blue light

August 27 2021, by Bob Yirka

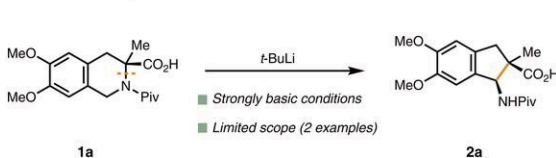
A Piperidine Diversification



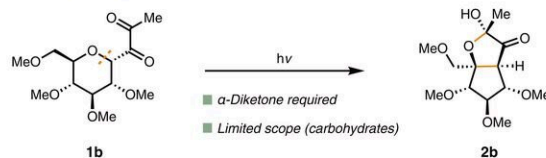
B Contraction Precedents



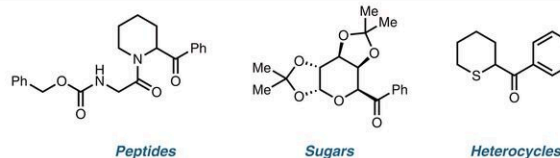
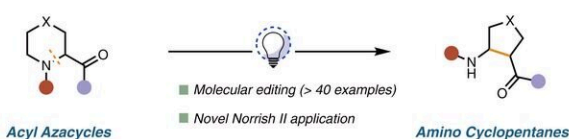
C Seebach (1993)



D Suarez (2008)



E This Work



Approaches to piperidine diversification. (A) Peripheral functionalization and skeletal remodeling. (B) Selected examples of ring contractions on piperidine frameworks. (C) Seminal report of Seebach and co-workers' unusual THIQ ring contraction (20). (D) Contraction of carbohydrates reported by Suárez and co-workers (21). (E) Norrish type II approach to piperidine skeletal framework modification (this work). Credit: *Science* (2021). DOI: 10.1126/science.abi7183

A team of chemists at the University of California, Berkeley, working

with a group at Merck & Co. Inc. has developed a reaction that can be used to remove a single sulfur, nitrogen or oxygen atom from a six-membered ring using only a blue light. In their paper published in the journal *Science* the group describes their reaction and possible uses for it in various applications.

In recent years, chemists have found ways to manipulate the [carbon rings](#) that are used to create a host of pharmaceutical and agrochemical products, but most are complex and difficult to conduct. They also typically involve using a lot of energy. In this new effort, the researchers have found a way to break the C–S, C–O and C–N bonds in saturated heterocycle rings to remove an atom, using only a [blue light](#)—afterward, the ring recloses, with one less atom.

To carry out a reaction, the researchers first expose an aromatic ketone group (which is attached to a heterocycle) to a blue LED [light](#). This sets off a Norrish reaction, wherein a carbon-heteroatom bond is cleaved, resulting in the opening of the ring and the ejection of an atom. This works because it results in the creation of a core radical. The second part of the reaction involves closure of the ring, which, the team notes, involves a procedure similar to a Mannich reaction. They note that for the reaction to work, the light wavelength had to match exactly with the carbonyl in the starting material. They also note that no [atoms](#) were removed from the molecule, nor were any added—the structure was simply rearranged in a way that left a five-membered ring.

The researchers note that despite its simplicity, the reaction is a game-changer for manipulating the building blocks of medicinal chemistry. They note also that the reaction could likely be used in combination with other reactions—offering a 1-2 punch sort of reaction to make more than one modification at a time. They demonstrated the effectiveness of their reaction by using it to edit multiple well-known drugs, such as mefloquine and rimiterol. They also noted that their new goal is to find a

way to add an aromatic ketone to a saturated heterocycle.

More information: Justin Jurczyk et al, Photomediated ring contraction of saturated heterocycles, *Science* (2021). [DOI: 10.1126/science.abi7183](https://doi.org/10.1126/science.abi7183)

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