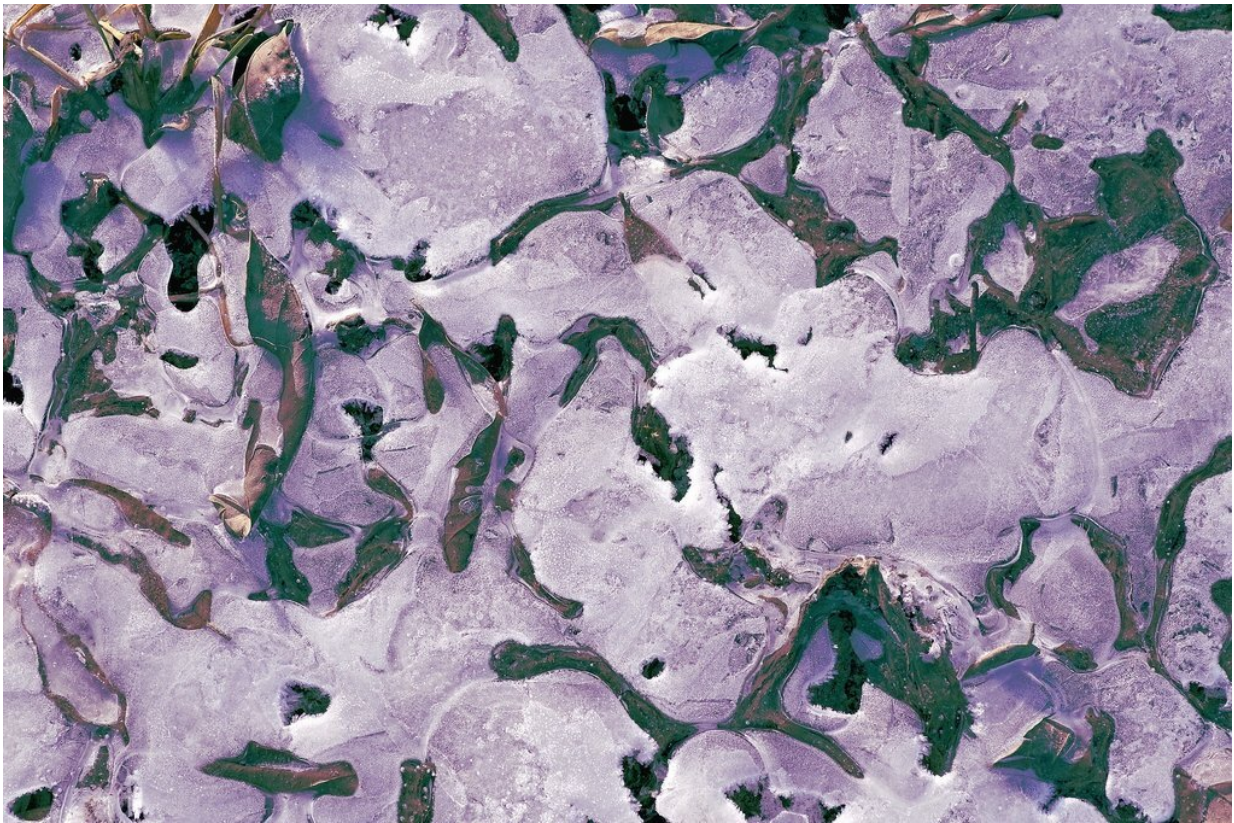


An investigation of thin liquid films at interfaces between ice and clay materials

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With their measurements, scientists can now better understand the processes occurring in permafrost soils at the boundary layer between ice and clay material. Credit: Pixabay / Arcaion

For ice, so-called 'surface melting' was postulated as early as the 19th

century by Michael Faraday: Already below the actual melting point, i.e. 0 °C, a thin liquid film forms on the free surface because of the interface between ice and air. Scientists led by Markus Mezger, group leader at the Max Planck Institute for Polymer Research (department of Hans-Jürgen Butt) and professor at the University of Vienna, have now studied this phenomenon in more detail at interfaces between ice and clay minerals.

In nature, this effect is particularly interesting in permafrost soils—i.e. soils that are permanently frozen. About a quarter of the land area in the northern hemisphere is covered by permafrost. These are composed of a mixture of ice and other materials. Microscopically thin platelets were formed over geological time by the weathering of [clay minerals](#). Similar to a sponge, a lot of [water](#) can enter the narrow slit pores between the thin platelets, be stored there, and freeze. Therefore, there is a lot of contact area between ice and [clay](#) minerals. For every gram of [clay mineral](#), there are about 10 square meters of surface area! This causes a comparatively high proportion of liquid water in the interfacially induced melt layer already below 0 °C.

The researchers have now investigated how fast the [water molecules](#) move in the thin melt layer at the boundary between ice and clay mineral. This value, known as self-diffusion, is directly linked to the viscosity of the water. For three different minerals, it has been shown that the viscosity of water in the interface-induced melt layer is sometimes significantly higher than that of ordinary water—i.e., the molecules are limited in their ability to move because the layer is more viscous. These results may help to better understand various phenomena in the future, such as the mechanical stability of permafrost, the transport of plant nutrients and pollutants, and geochemical reactions such as ion exchange processes at ice/mineral interfaces.

For their measurements, the Mainz scientists collaborated with partners

at the research reactors of the TU Munich and the Institut Laue-Langevin in Grenoble, France. The neutrons generated in the reactors there strike the sample at a certain speed. Similar to a ball bouncing back from a vehicle moving toward it at a higher speed, velocity [measurements](#) of the neutrons scattered from the sample allow conclusions to be drawn about the motion of the water molecules in the interface-induced premelting layer.

More information: Hailong Li et al, Water Mobility in the Interfacial Liquid Layer of Ice/Clay Nanocomposites, *Angewandte Chemie International Edition* (2020). [DOI: 10.1002/anie.202013125](https://doi.org/10.1002/anie.202013125)

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