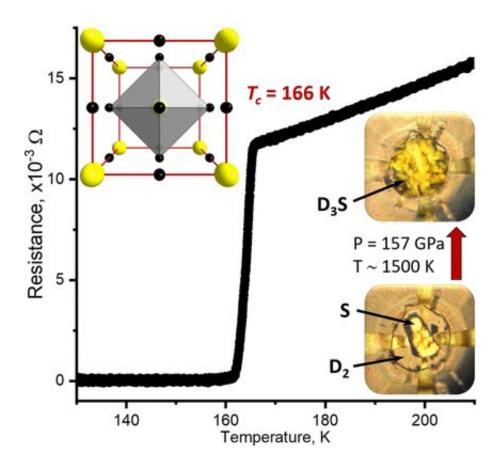


On the road to conductors of the future

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Credit: Angewandte Chemie International Edition

Superconducting wires can transport electricity without loss. This would allow for less power production, reducing both costs and greenhouse gasses. Unfortunately, extensive cooling stands in the way, because existing superconductors only lose their resistance at extremely low temperatures. In the journal *Angewandte Chemie*, scientist have now



introduced new findings about hydrogen sulfide in the H_3S form, and its deuterium analog D_3S , which become superconducting at the relatively high temperatures of -77 and -107 °C, respectively.

This is even true in comparison with the current front-runners, coppercontaining ceramics with transition temperatures that start at about -135 °C. Despite extensive research into sulfur/hydrogen systems, many important questions remain. Most importantly, superconducting hydrogen sulfide was previously produced from "normal" hydrogen sulfide, H₂S, which was converted into a metal-like state with a composition of H₃S under pressures of about 150 GPa (1.5 million bar). Such samples were inevitably contaminated by hydrogen-depleted impurities that can distort experimental results. To avoid this, researchers led by Vasily S. Minkov have now produced stoichiometric H₃S by heating elemental sulfur directly with an excess of hydrogen (H₂) with a laser, under pressure. They also produced samples made with deuterium (D₂)—an isotope of hydrogen.

The cause of the relatively high transition temperature of H₃S is its hydrogen atoms, which resonate with an especially high frequency within the crystal lattice. Because deuterium atoms are heavier than hydrogen, they resonate more slowly, so lower transition temperatures were expected for D₃S. The team at the Max-Planck Institute for Chemistry (Mainz, Germany), the University of Chicago (USA), and the Soreq Nuclear Research Center (Yavne, Israel) used a variety of analytical methods to refine the phase diagrams for H₃S and D₃S in relation to pressure and temperature, and to shed additional light on their superconducting properties.

At 111 to 132 GPa and 400 to 700 °C, the syntheses produced nonmetallic, electrically isolating structures (Cccm phases) that do not become a metal when cooled or pressurized further. They contain H_2 (or D_2) units within the crystal structure, which suppress superconductivity.



The desired superconducting structures, cubic Im-3m phases, were obtained by syntheses above 150 GPa at 1200 to 1700 °C. They are metallic and shiny with low electrical resistance. At 148 to 170 GPa, samples of Im-3m-H₃S had transition temperatures around -77 °C. The D₃S analogs had a transition temperature of about -107 °C at 157 GPa, which is significantly higher than expected. Decrease of pressure reversibly leads to an abrupt reduction of the transition temperature and loss of metallic properties. This is caused by rhombohedral distortions in the crystal structure (R3m phase). Heating under pressure irreversibly transforms the R3m phase into the Cccm phase. R3m is clearly a metastable intermediate phase that only occurs during decomposition.

In the future, the researchers hope to find other hydrogen-rich compounds that can be converted to metals without high pressures and become superconducting at room <u>temperature</u>.

More information: Vasily S. Minkov et al, A Boosted Critical Temperature of 166 K in Superconducting D 3 S Synthesized from Elemental Sulfur and Hydrogen, *Angewandte Chemie International Edition* (2020). DOI: 10.1002/anie.202007091

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