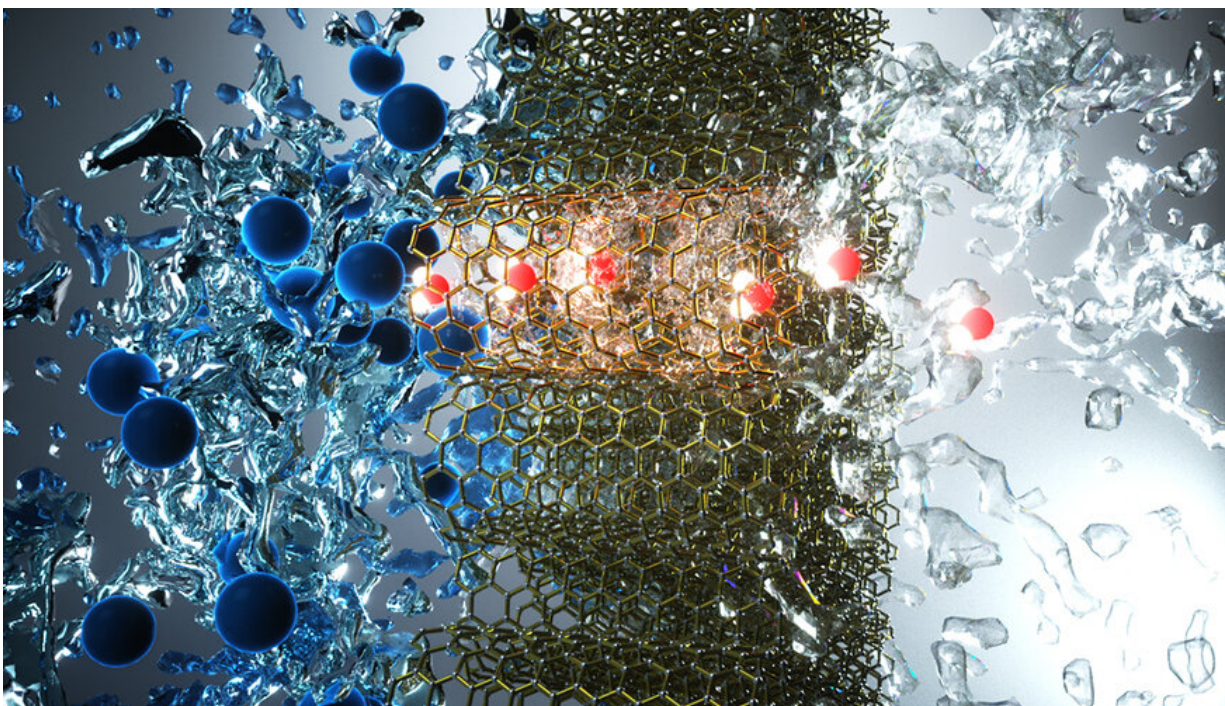


Carbon nanotubes developed for super efficient desalination

September 21 2020, by Anne M Stark



An artist's view of small-diameter carbon nanotubes that pass through water molecules (red and white) and reject ions (blue). High permselectivity of small-diameter nanotubes can enable advanced water desalination technologies. Credit: A. Noy, T. A. Pham, Y. Li, Z. Li, F. Aydin (LLNL). Illustration by Ella Maru Studios.

Membrane separations have become critical to human existence, with no better example than water purification. As water scarcity becomes more

common and communities start running out of cheap available water, they need to supplement their supplies with desalinated water from seawater and brackish water sources.

Lawrence Livermore National Laboratory (LLNL) researchers have created [carbon nanotube](#) (CNT) pores that are so efficient at removing salt from water that they are comparable to commercial desalination membranes. These tiny pores are just 0.8 nanometers (nm) in diameter. In comparison, a human hair is 60,000 nm across. The research appears on the cover of the Sept. 18 issue of the journal *Science Advances*.

The dominant technology for removing salt from water, [reverse osmosis](#), uses thin-film composite (TFC) membranes to separate water from the ions present in saline feed streams. However, some fundamental performance issues remain. For example, TFC membranes are constrained by the permeability-selectivity trade-offs and often have insufficient rejection of some ions and trace micropollutants, requiring additional purification stages that increase the energy and cost.

Biological water channels, also known as aquaporins, provide a blueprint for the structures that could offer increased performance. They have an extremely narrow inner pore that squeezes water down to a single-file configuration that enables extremely high water permeability, with transport rates exceeding 1 billion water molecules per second through each pore.

"Carbon nanotubes represent some of the most promising scaffold structures for artificial water channels because of the low friction of water on their smooth inner surfaces, which mimic the biological water channels," said Alex Noy, LLNL chemist and a lead co-author of the report.

The team developed CNT porins (CNTPs)—short segments of CNTs

that self-insert into biomimetic membranes—which form artificial water channels that mimic aquaporin channel functionality and intrachannel single-file water arrangement. Researchers then measured water and chloride ion transport through 0.8-nm-diameter CNTPs using fluorescence-based assays. Computer simulations and experiments using CNT pores in [lipid membranes](#) demonstrated the mechanism for enhanced flow and strong ion rejection through inner channels of carbon nanotubes.

"This process allowed us to determine the accurate value of water-salt permselectivity in narrow CNT pores," said LLNL materials scientist and lead co-author Tuan Anh Pham, who led the simulation efforts of the study. "Atomistic simulations provide a detailed molecular-scale view of [water](#) entering the CNTP channels and support the activation energy values."

More information: Yuhao Li et al. Water-ion permselectivity of narrow-diameter carbon nanotubes, *Science Advances* (2020). [DOI: 10.1126/sciadv.aba9966](https://doi.org/10.1126/sciadv.aba9966)

Provided by Lawrence Livermore National Laboratory

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