

Microscopic computers: The wires of the future may be made of molecules

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Credit: Marius Christian Eriksen, Department of Physics, Chemistry and Pharmacy.

There are physical limits to how powerful computers can become if they

are to maintain their size. Molecular electronics can solve that problem, and now SDU researchers are contributing to this field with a new, efficient conducting material, based on molecules.

Our computers are becoming more and more powerful all the time. They also often become smaller—just think of what a standard smartphone can do today compared to just a few years ago.

But the development cannot last.

"With our current technology, we will soon reach the limit of how small the components within a computer can be," says Steffen Bähring from the Department of Physics, Chemistry and Pharmacy, University of Southern Denmark. He studies molecules and for this study he investigated how good they are at conducting electricity.

"The current technology based on silicon will reach the limit within the next 10 years and we do not yet have a technology ready to take over. But molecules are candidates to push the limit much further," he believes.

Together with international colleagues Jonathan L. Sessler (Texas, USA), Dirk M. Guldi (Erlangen, Germany) and Atanu Jana (Shanghai, China), he has just published a new scientific study on the composition of molecules in liquids and as crystalline materials which proved particularly interesting.

The study is published in the *Journal of The American Chemical Society*.

"We see really good conductivity qualities, which is an extremely important feature when talking about the development of [electronic devices](#) and computers of the future," he says.

He believes that if we want even more powerful computers than today, which also remain small, then the electronics have to transition to molecular dimensions, meaning the individual components will be under a nanometer in size.

The new 'molecular wire,' which the researchers describe in their article, is a good example and an elegant system, he believes.

Steffen Bähring explains the principle in the new molecular wire as follows:

"This is the first time that only neutral molecules, which are capable of recognizing and finding each other in solution, are used, thus forming a well-defined three-dimensional structure having semiconductor properties. By inserting different components, we can modify the conductivity and thereby control the system.

"Our system differs from previous ones, which are based on salts containing metals. These are not capable of forming different structures like our system.

"One challenge in building electronic devices from molecules is that the molecular wires must have satisfactory conducting properties. But there is also another challenge: stability.

"It's extremely difficult to control things this small, and when we talk about molecular electronics, stability is the biggest weakness. These are electroactive materials, and when you supply them with energy, the molecules will be charged, and any weaknesses will cause the molecules to break," says Bähring.

Such molecular instability is also known in the world we can see. An example is how the [molecules](#) in our skin change when the skin absorbs

energy from the sunlight if we do not protect it with sunscreen.

More information: Ramandeep Kaur et al, Semiconducting Supramolecular Organic Frameworks Assembled from a Near-Infrared Fluorescent Macrocyclic Probe and Fullerenes, *Journal of the American Chemical Society* (2020). [DOI: 10.1021/jacs.0c03699](https://doi.org/10.1021/jacs.0c03699)

Provided by University of Southern Denmark

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