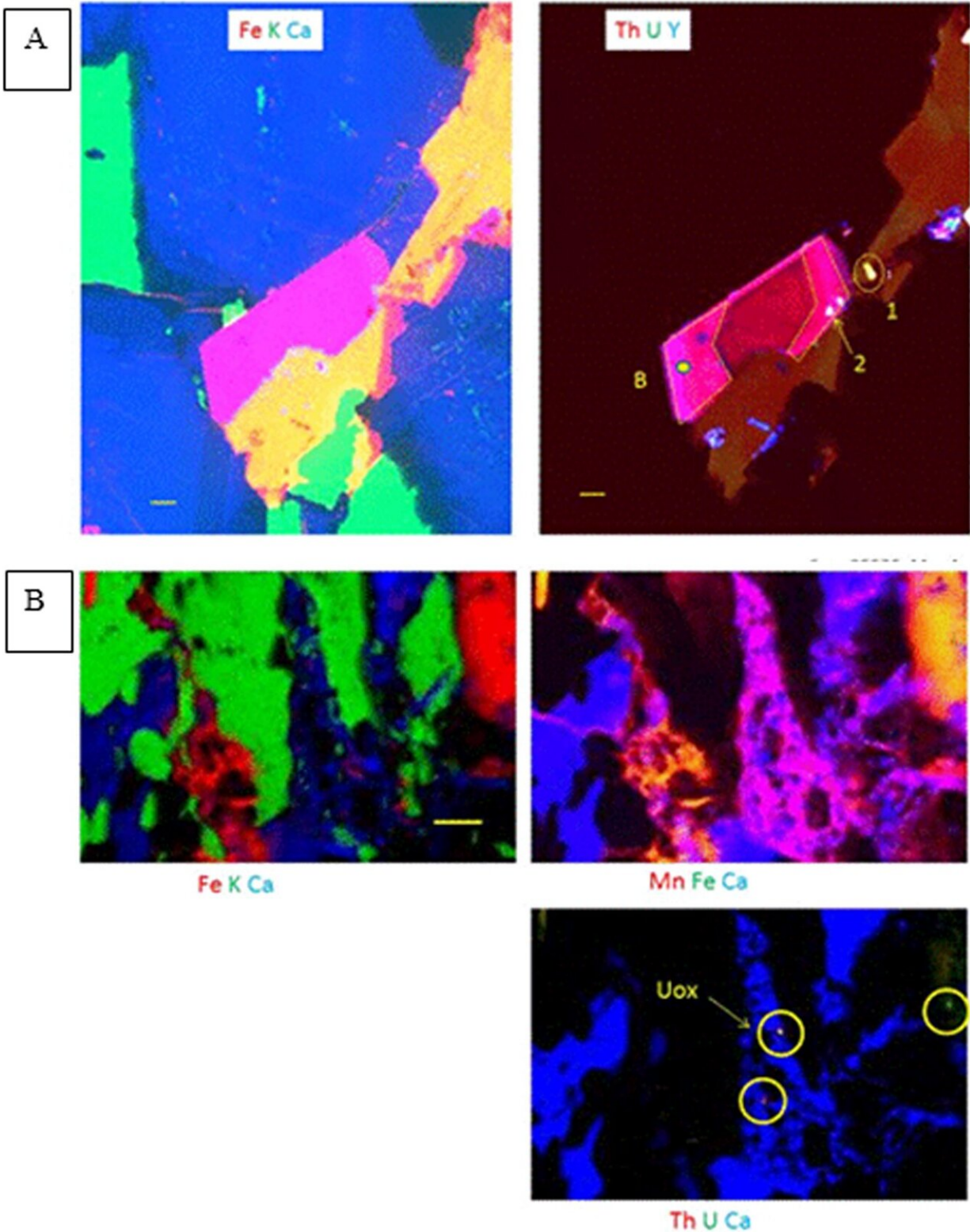


# Tiny crystals plug gaps and limit uptake of contaminants in rocks

May 19 2020, by [Follow Me On Twitter\(Opens In New Window\)](#) Ben Robinson

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Synchrotron microfocus XRF maps. (A) MIU-3/8 Left: Fe (red), K (green), and Ca (blue). Minerals are as follows: green = K-feldspar, blue = plagioclase

(mottled areas indicate plagioclase breakdown to phyllosilicate, especially in the more anorthitic cores), yellow = biotite, and pink = bastnaesite, a rare earth fluoride-carbonate,  $\text{CeCO}_3(\text{F})$ . Right: Th (red), U (green), and Y (blue). This crystal of bastnaesite is rich in Th. Discrete grains of a U-rich phase appear as small yellow areas, while blue acicular regions are rich in Y and may be a REE/Y rich phosphate phase. Yellow dot labelled 'B' is approximate location of Th-XANES and bastnaesite point analysis. Points labelled 1 and 2 correspond to AcO2 point analyses 1 and 2 in Table 2. Th XANES was also taken at the point labelled 1. (B) MIU-3/10 synchrotron microfocus XRF map. This image includes mineral infill associated with secondary fractures. False colour elemental map, with colour key provided as element symbols coloured to correspond to map colours. Top left major element map shows K-feldspar in green, iron oxyhydroxide in red, and calcite in blue. In the top right panel, the magenta region is calcite with enriched Mn concentrations. Bright orange shows the incorporation of Mn into Fe-oxyhydroxides. The lower right panel clearly shows the calcite precipitates and small regions within the infill that have high concentrations of both Th and U (yellow circles). The hotspot labelled Uox indicates the location of the oxidized U L-III XANES spectrum presented below. 100-micron scale bar. *Scientific Reports* DOI: 10.1038/s41598-020-65113-x

Research published today by a UK-based team of scientists has shown for the first time that the mobility of potentially harmful contaminants in crystalline rocks over long periods of time may be severely limited due to the presence of tiny crystals, meaning contaminant movement is likely to be focused to water-bearing fractures only.

Movement of contaminants through rocks below ground can act to spread contamination, an issue relevant to the geological disposal of some wastes. We undertake studies to enhance our understanding of how this process works, reduce uncertainties and further consider any potential risks it could pose.

These new results shed light on the [difficult problem](#) of how

contaminants may move over extremely long time periods and should improve our ability to calculate long term risks.

This study, published in the journal *Scientific Reports*, analyzed crystalline (granite) [rock samples](#) from an underground system in Japan and the results imply that in many cases the importance of '[rock matrix diffusion](#)' may be minimal. Additional analyses of a contrasting crystalline rock system (Carmenellis Granite, UK) corroborate these results.

These findings led by The University of Manchester, which apply to long-lived systems, build on previous laboratory and field studies over short periods of time which also suggested that contaminant mobility in crystalline rocks, such as granite, will be limited to short distances in parts of the rock that are away from large fractures.

This new work has examined rocks from ancient crystalline rock systems in Japan and the UK to show that even over long periods of geological time the movement of elements within such crystalline rock is indeed small, in large part because the formation of large quantities of small crystals during the aging of the rock acts to seal small openings and limit fluid access to only a few millimeters of the rock bordering fractures.

Professor Roy Wogelius, the senior author on this paper, commented: "We set out to test exactly what we could resolve in terms of fluid access to the matrix of these rocks and we were amazed at the extremely limited volume involved. But what was most amazing to us was the distribution of tiny crystals of carbonate minerals throughout what we usually think of as a uniform block of crystalline rock.

"Here, unexpected little crystals of calcite appear throughout the rock plugging up all the tiny openings. These crystals clog everything up and keep most of the fluid in large cracks with no access to smaller openings.

This effectively shuts down contaminant access to the rock mass, meaning any contaminant movement would likely focus in rock fractures only. "

**More information:** R. A. Wogelius et al. Mineral reaction kinetics constrain the length scale of rock matrix diffusion, *Scientific Reports* (2020). [DOI: 10.1038/s41598-020-65113-x](https://doi.org/10.1038/s41598-020-65113-x)

Provided by University of Manchester

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