

Scientists break the link between a quantum material's spin and orbital states

May 15 2020, by Glennda Chui



These balloon-and-disc shapes represent an electron orbital -- a fuzzy electron cloud around an atom's nucleus -- in two different orientations. Scientists hope to someday use variations in the orientations of orbitals as the 0s and 1s needed to make computations and store information in computer memories, a system known as orbitronics. A SLAC study shows it's possible to separate these orbital orientations from electron spin patterns, a key step for independently controlling them in a class of materials that's the cornerstone of modern information technology. Credit: Greg Stewart/SLAC National Accelerator Laboratory



In designing electronic devices, scientists look for ways to manipulate and control three basic properties of electrons: their charge; their spin states, which give rise to magnetism; and the shapes of the fuzzy clouds they form around the nuclei of atoms, which are known as orbitals.

Until now, electron spins and orbitals were thought to go hand in hand in a class of materials that's the cornerstone of modern information technology; you couldn't quickly change one without changing the other. But a study at the Department of Energy's SLAC National Accelerator Laboratory shows that a pulse of laser light can dramatically change the spin state of one important class of materials while leaving its orbital state intact.

The results suggest a new path for making a future generation of logic and <u>memory devices</u> based on "orbitronics," said Lingjia Shen, a SLAC research associate and one of the lead researchers for the study.

"What we're seeing in this system is the complete opposite of what people have seen in the past," Shen said. "It raises the possibility that we could control a material's spin and orbital states separately, and use variations in the shapes of orbitals as the 0s and 1s needed to make computations and store information in computer memories."

The international research team, led by Joshua Turner, a SLAC staff scientist and investigator with the Stanford Institute for Materials and Energy Science (SIMES), reported their results this week in *Physical Review B Rapid Communications*.

An intriguing, complex material

The material the team studied was a manganese oxide-based quantum material known as NSMO, which comes in extremely thin crystalline layers. It's been around for three decades and is used in devices where



information is stored by using a <u>magnetic field</u> to switch from one electron <u>spin state</u> to another, a method known as spintronics. NSMO is also considered a promising candidate for making future computers and memory storage devices based on skyrmions, tiny particle-like vortexes created by the magnetic fields of spinning electrons.

But this material is also very complex, said Yoshinori Tokura, director of the RIKEN Center for Emergent Matter Science in Japan, who was also involved in the study.



In SLAC experiments, scientists hit a quantum material with pulses of laser light (top) to see how this would affect zigzag patterns (middle) in its atomic lattice made by the spin directions of electrons (black arrows) and the orientations of electron orbitals (red balloon shapes). They were surprised to discover that the pulses disrupted the spin patterns while leaving the orbital patterns intact (bottom). This raises the possibility that spin and orbital states could be independently controlled to make much faster electronic devices. Credit: Greg Stewart/SLAC National Accelerator Laboratory



"Unlike semiconductors and other familiar materials, NSMO is a quantum material whose electrons behave in a cooperative, or correlated, manner, rather than independently as they usually do," he said. "This makes it hard to control one aspect of the electrons' behavior without affecting all the others."

One common way to investigate this type of material is to hit it with laser light to see how its electronic states respond to an injection of energy. That's what the research team did here. They observed the material's response with X-ray laser pulses from SLAC's Linac Coherent Light Source (LCLS).

One melts, the other doesn't

What they expected to see was that orderly patterns of <u>electron spins</u> and orbitals in the material would be thrown into total disarray, or "melted," as they absorbed pulses of near-infrared <u>laser light</u>.

But to their surprise, only the spin patterns melted, while the orbital patterns stayed intact, Turner said. The normal coupling between the spin and orbital states had been completely broken, he said, which is a challenging thing to do in this type of correlated material and had not been observed before.

Tokura said, "Usually only a tiny application of photoexcitation destroys everything. Here, they were able to keep the electron state that is most important for future devices—the orbital state—undamaged. This is a nice new addition to the science of orbitronics and correlated electrons."

Much as electron spin states are switched in spintronics, electron orbital states could be switched to provide a similar function. These orbitronic



devices could, in theory, operate 10,000 faster than spintronic devices, Shen said.

Switching between two orbital states could be made possible by using short bursts of terahertz radiation, rather than the magnetic fields used today, he said: "Combining the two could achieve much better device performance for future applications." The team is working on ways to do that.

More information: L. Shen et al, Decoupling spin-orbital correlations in a layered manganite amidst ultrafast hybridized charge-transfer band excitation, *Physical Review B* (2020). DOI: 10.1103/PhysRevB.101.201103

Provided by SLAC National Accelerator Laboratory

Citation: Scientists break the link between a quantum material's spin and orbital states (2020, May 15) retrieved 6 May 2024 from <u>https://phys.org/news/2020-05-scientists-link-quantum-material-orbital.html</u>

This document is subject to copyright. Apart from any fair dealing for the purpose of private study or research, no part may be reproduced without the written permission. The content is provided for information purposes only.