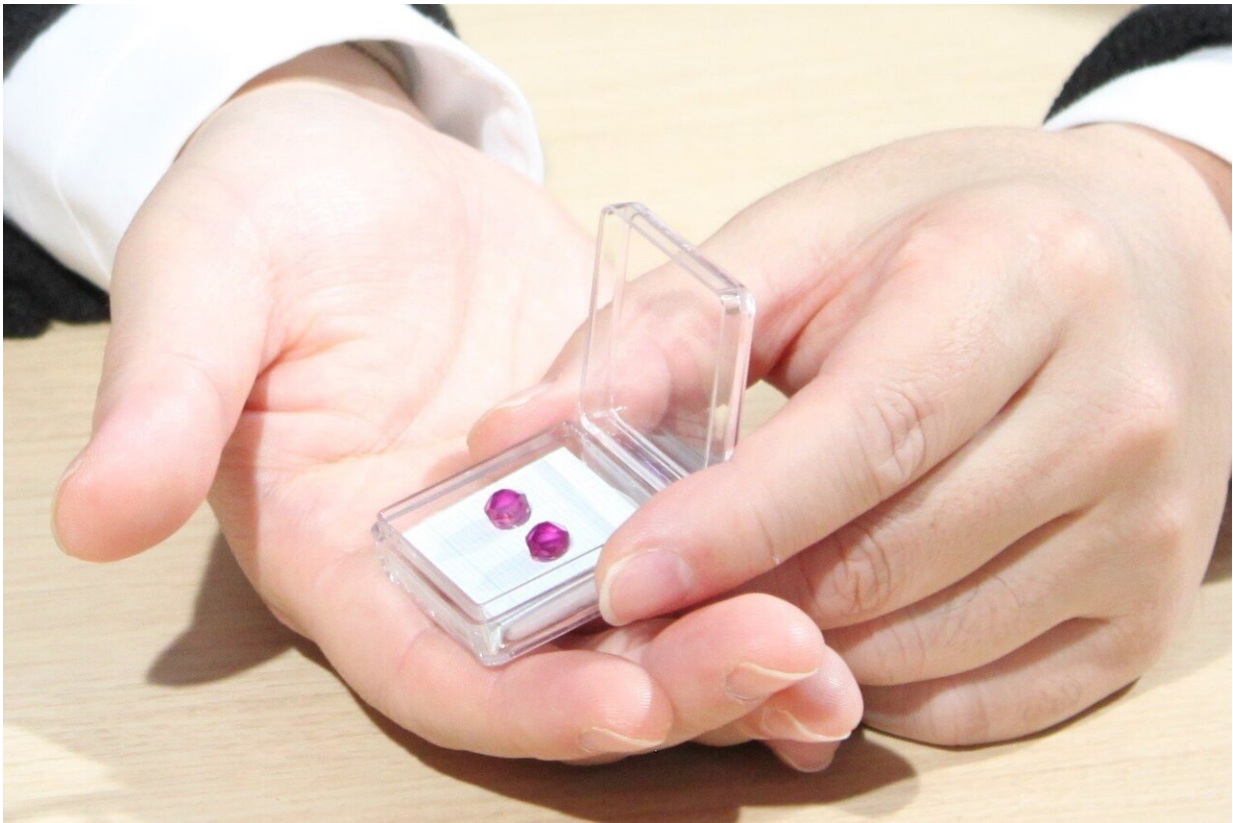


# Rubies on sapphire: Recipe for making crystals in flux

May 1 2020

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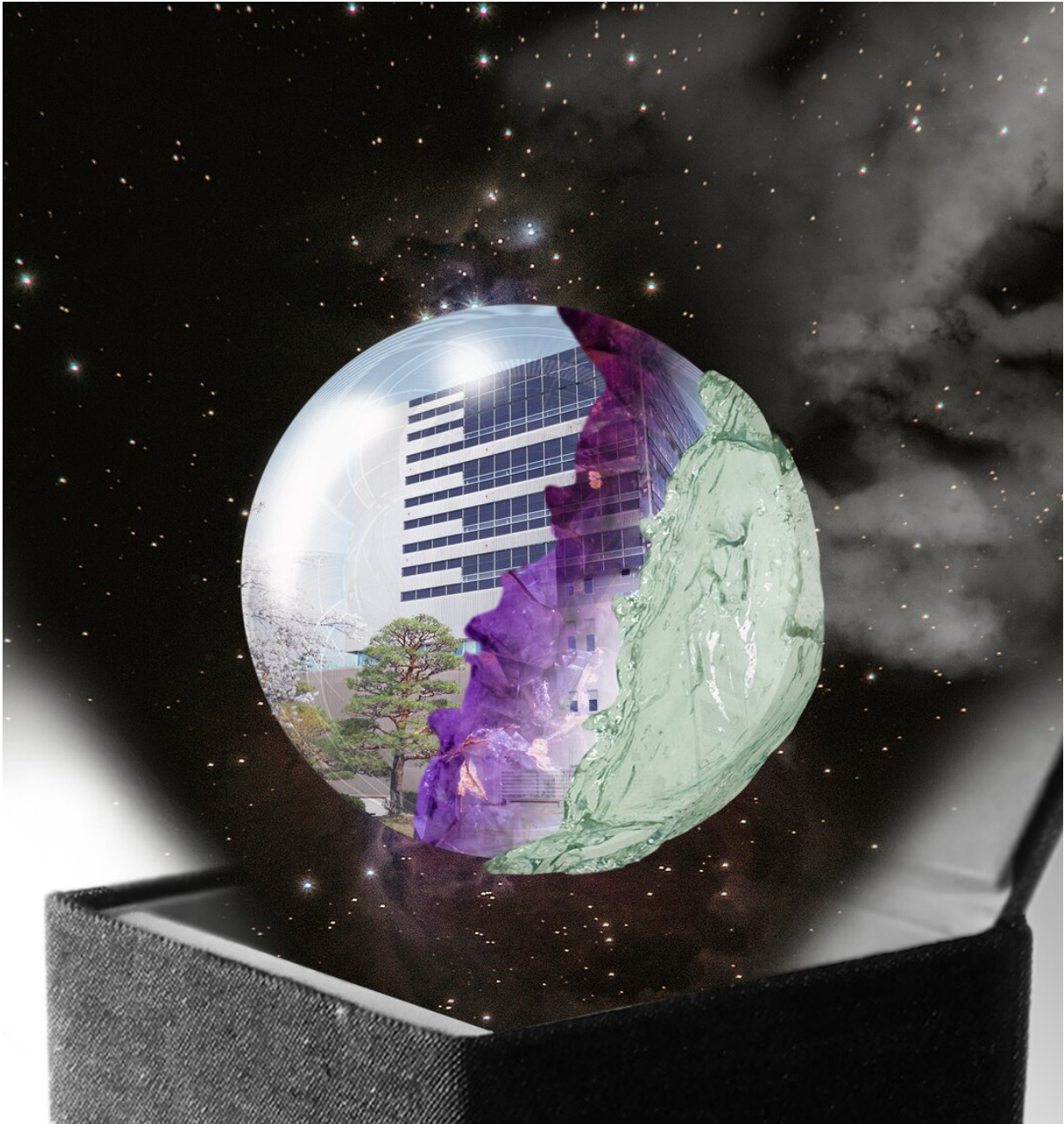
The effect of the holding temperature and solubility curve of rubies was elucidated, for  $\text{Al}_2\text{O}_3:\text{Cr}$  in  $\text{MoO}_3$  from 1050 to 1200. Credit: Katsuya Teshima Ph.D., Research Initiative for Supra-Materials, Shinshu University

Crystals can be made artificially but a lot of energy is used to melt the

ingredients together, and this can make them expensive. This problem can be overcome by using appropriate solvents. Called the flux method, crystals are grown in a crucible that contains solvents that allow the crystal to form with less energy because dissolution will happen more easily. Imagine having table salt and wanting to form crystals of a desirable structure. The salt can be heated to its melting point which would take a lot of energy, or, it can be dissolved in a solvent such as water, and the water can be evaporated at a much lower temperature than trying to heat the salt on its own. Finding the right conditions is key for having good crystal recipes.

Katsuya Teshima of Shinshu University is an expert on flux recipes for [crystal growth](#), and continues research to produce optimal [crystals](#) of desirable properties. With the recent paper published in *Crystal Growth & Design* Teshima, first author Shunsuke Ayuzawa and their team investigated rubies ( $\text{Al}_2\text{O}_3:\text{Cr}$ ) and its solubility curve in flux. The flux method of growing crystals is currently the preferable method of procuring crystals because it has negligible environmental impact. With this research, Teshima and his team at Shinshu University investigated the partial solubility curve of ruby crystal growth in Molybdenum Trioxide ( $\text{MoO}_3$ ).

The solubility curve is a graph that shows how much dissolves at what [temperature](#). It is crucial to know this information in order to grow crystals with precision. With this research, the solubility curve of  $\text{Al}_2\text{O}_3:\text{Cr}$  in  $\text{MoO}_3$  for 1050 to 1200 °C was obtained. Figuring out the solubility curve of this crystal which vaporizes at such high temperatures had been extremely difficult, but this team managed to figure out a way by using the guiding principle of epitaxial growth on a single crystal sapphire substrate to measure a small amount of ruby crystal growth. The key was to distinguish between the red ruby layer and the transparent sapphire layer as the flux evaporated.



Ruby crystal grown on sapphire substrate at the Nagano engineering campus, Shinshu University. Credit: Katsuya Teshima Ph.D., Research Initiative for Supra-Materials, Shinshu University

Ruby is the Latin word for red, and as the name suggests, is renowned for its beautiful pink and maroon color. Rubies are one of the five cardinal gems because they were rare in the past. Other cardinal gems are diamonds, emeralds, sapphire and amethyst. What is becoming more apparent is that rubies are not only nice to look at, but have properties that can be used in a variety of industries. Rubies have mechanical strength, excellent optical properties and are chemically stable. Single crystals of ruby make great optical devices. The first reported solid laser in 1960 was made of ruby crystal.

Unlike diamonds which are made of pure carbon, rubies are made of a mix of the mineral corundum, a crystalline form of aluminum oxide with a trace of chromium that gives it the red color. The holding temperature, which is the temperature at which the solution is kept, produces different types of rubies with differing properties. The newly discovered [solubility](#) curve will enable scientists to be more precise when making crystals.

Teshima, a university researcher, hopes to systemize the phase diagram of flux methods for all materials. Having relied on experience and knowledge in the past, he will conduct more research performed through analytical chemistry and computational science from a process chemistry standpoint. He believes that new substances can be discovered and created to perform new functions through the systemization of the [flux](#) method.

**More information:** Shunsuke Ayuzawa et al, Effect of Holding Temperature on Growth of Ruby Crystal Films via Molybdenum Trioxide Flux Evaporation–Solubility of Aluminum Oxide, Growth Rate, and Material Balance, *Crystal Growth & Design* (2020). [DOI: 10.1021/acs.cgd.9b01674](https://doi.org/10.1021/acs.cgd.9b01674)



Provided by Shinshu University

Citation: Rubies on sapphire: Recipe for making crystals in flux (2020, May 1) retrieved 25 April 2024 from <https://phys.org/news/2020-05-rubies-sapphire-recipe-crystals-flux.html>

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