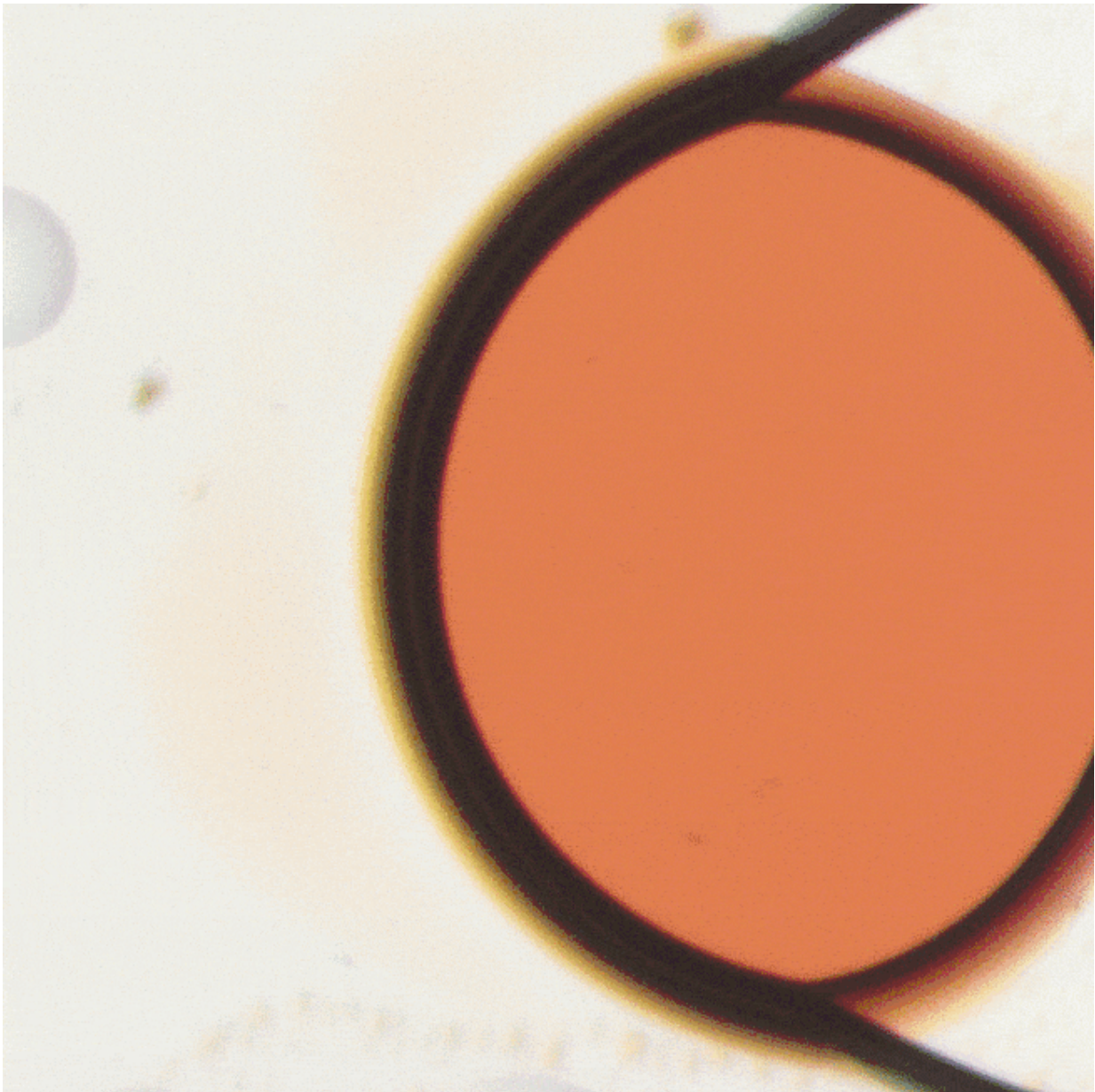


# Chemistry breakthrough could speed up drug development

May 8 2020

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BODIPY crystal growth. Credit: Newcastle University

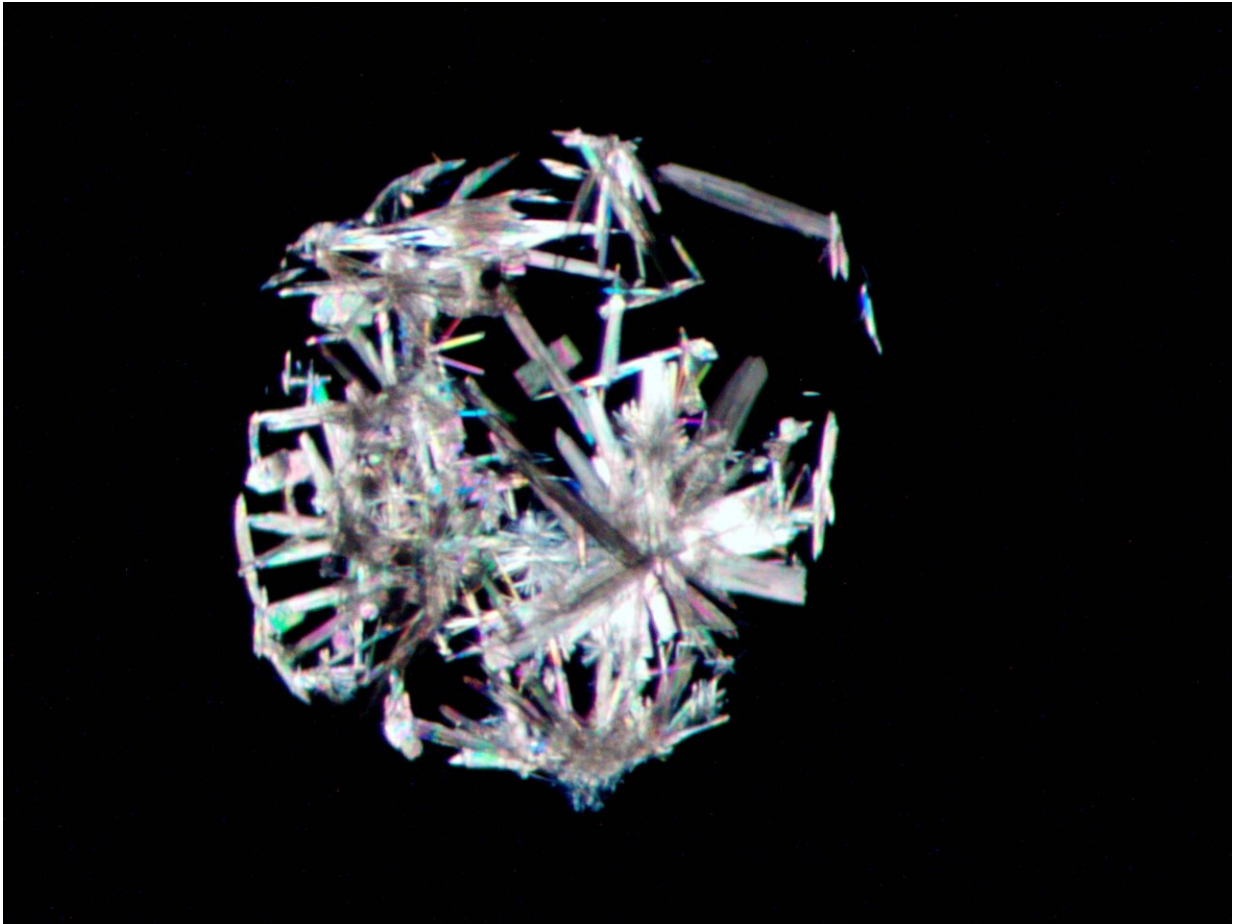
Scientists have successfully developed a new technique to reliably grow crystals of organic soluble molecules from nanoscale droplets, unlocking the potential of accelerated new drug development.

Chemistry experts from Newcastle and Durham universities, working in collaboration with SPT Labtech, have grown the small crystals from nanoscale encapsulated droplets. Their innovative method, involving the use of inert oils to control evaporative solvent loss, has the potential to enhance the [drug](#) development pipeline.

Whilst crystallization of organic soluble [molecules](#) is a technique used by scientists all over the world, the ability to do so with such small quantities of analyte is ground-breaking.

Through the use of this new method, called Encapsulated Nanodroplet Crystallisation (ENaCt), the researchers have shown that hundreds of crystallisation experiments can be set up within a few minutes. Each experiment involves a few micrograms of molecular analyte dissolved in a few nanoliters of organic solvent and is automated, allowing for rapid set up of hundreds of unique experiments with ease. Concentration of these nanodroplet experiments results in the growth of the desired high quality single crystals that are suitable for modern X-ray diffraction analysis.

Publishing their findings in the journal *Chem*, the team, led by Drs Michael Hall and Mike Probert, of Newcastle University, UK, successfully developed a new approach to molecular crystallization which allows access, within a few days, to high quality single crystals, whilst requiring only few milligrams of analyte.



Scientists have successfully developed a new technique to reliably grow crystals of organic soluble molecules from nanoscale droplets, unlocking the potential of accelerated new drug development. Credit: Dr Michael Hall, Newcastle University

### **Far-reaching impact**

Dr. Hall, senior lecturer in chemistry, Newcastle University, said: "We have developed a nanoscale crystallization technique for organic-soluble small molecules, using high-throughput liquid-handling robotics to undertake multiple [crystallisation](#) experiments simultaneously with

minimal sample requirements and high success rates.

"This new method has the potential to have far-reaching impact within the molecular sciences and beyond. Fundamental research will benefit from highly detailed characterization of new molecules, such as natural products or complex synthetic molecules, by X-ray crystallography, whilst the development of new drugs by the pharmaceutical industry will be accelerated, through rapid access to characterised crystalline forms of new active pharmaceutical ingredients."

Understanding these new crystalline forms, known as polymorphs, is essential to the successful generation of new pharmaceutical agents and drugs. The ability to investigate these forms quickly and on a vast scale, whilst minimizing the amount of analyte required, could be a key breakthrough enabled by the new ENaCT protocol.

Dr. Paul Thaw from SPT Labtech, added: "Enabling this work to develop a novel high-throughput method for single crystal X-ray diffraction on mosquito with the Newcastle team has been a pleasure. Having the ability to quickly screen organic soluble small molecules on the microgram scale will deliver valuable insight for both academic research and pharmaceutical drug design and validation."

Dr. Probert, senior lecturer in inorganic chemistry and head of crystallography, Newcastle University, commented "...this new approach to crystallization has the ability to transform the scientific landscape for the analysis of small molecules, not only in the drug discovery and delivery areas but also in the more general understanding of the crystalline solid state."

**More information:** Andrew R. Tyler et al. Encapsulated Nanodroplet Crystallization of Organic-Soluble Small Molecules, *Chem* (2020). [DOI: 10.1016/j.chempr.2020.04.009](https://doi.org/10.1016/j.chempr.2020.04.009)

Provided by Newcastle University

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