

New metal–organic framework can take on toxic sulfur dioxide gas

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Illustration of sulfur dioxide captured within the MFM-170 material, as revealed by experiments at Berkeley Lab's Advanced Light Source. Credit: Gemma Smith/Manchester University

An international team has developed a robust material that can



selectively take in toxic sulfur dioxide gas at record concentrations and preserve it for use in chemical production. The researchers verified its performance using a combination of techniques that included X-ray experiments at Lawrence Berkeley National Laboratory's (Berkeley Lab's) Advanced Light Source (ALS).

Sulfur dioxide emissions are typically produced by <u>power plants</u>, other <u>industrial facilities</u>, and trains, ships, and heavy equipment, and can be harmful to human health and the environment. The team developed porous, cagelike, stable copper-containing molecules known as <u>metalorganic frameworks</u> or MOFs that are designed to separate sulfur dioxide (SO₂) gas from other gases. The team exposed the MOF material, dubbed MFM-170, to simulated exhaust gases and found that it efficiently separated out SO₂ from the gas mixture at elevated temperatures even in the presence of water.

Existing techniques to remove SO_2 from pollution streams can produce a lot of solid and liquid waste and may only remove 60-95% of the toxic gas, researchers noted, while the MOF has been shown to eliminate SO_2 down to a level below 0.1 parts per million—or 99.99999% SO_2 -free. Their study was published Oct. 14 in the journal *Nature Materials*.

More information: Gemma L. Smith et al. Reversible coordinative binding and separation of sulfur dioxide in a robust metal–organic framework with open copper sites, *Nature Materials* (2019). DOI: 10.1038/s41563-019-0495-0

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