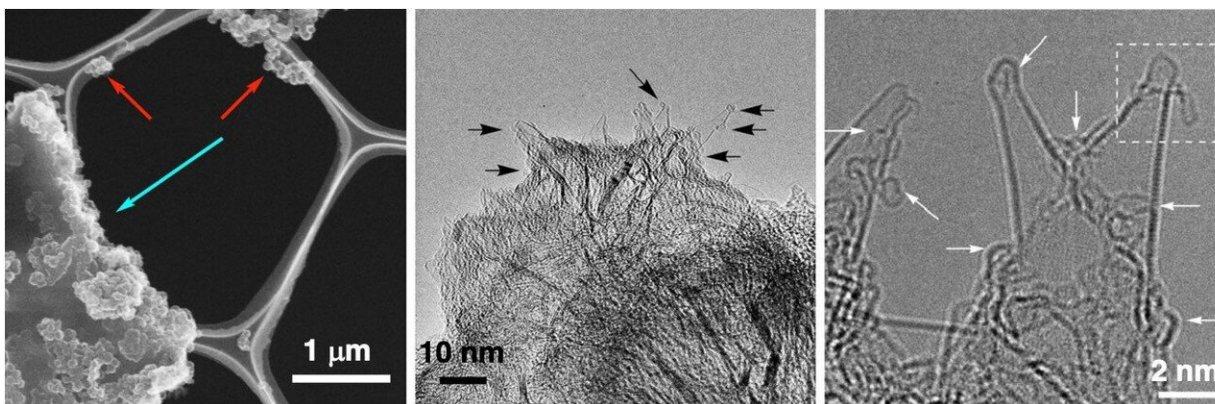


# Researchers produce first-ever videos of chemical synthesis at atomic resolution

August 23 2019



Electron microscope images of the test sample at different magnifications.  
Credit: (c) 2019 Nakamura et al.

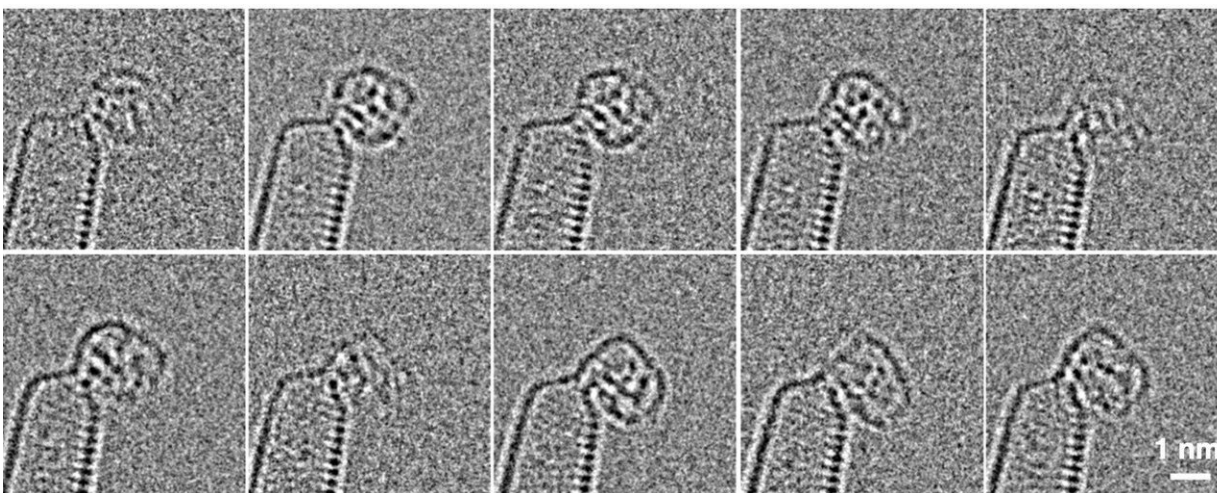
For the first time, researchers have managed to view previously inaccessible details of certain chemical processes. They have shown there are significant discrete stages to these processes that build on existing knowledge of chemical synthesis. These details could aid in the development of methods to synthesize chemicals with greater control and precision than ever before. Methods such as these could be useful in materials science and in drug development.

"Since 2007, physicists have realized a dream over 200 years old—the ability to see an individual atom," said Project Professor Eiichi

Nakamura. "But it didn't end there. Our research group has reached beyond this dream to create videos of [molecules](#) to see [chemical reactions](#) in unprecedented detail."

Nakamura's team from the University of Tokyo's Department of Chemistry seeks to master control of various chemical processes responsible for material synthesis. However, chemical synthesis is a complicated field of study.

"Conventional analytical methods such as spectroscopy and crystallography give us useful information about the outcomes of processes, but only hints about what takes place during them," explained Koji Harano, project associate professor in the Nakamura group. "For example, we are interested in metal-organic framework (MOF) crystals. Most studies look at the growth of these but miss the early stage of nucleation, as it is difficult to observe."

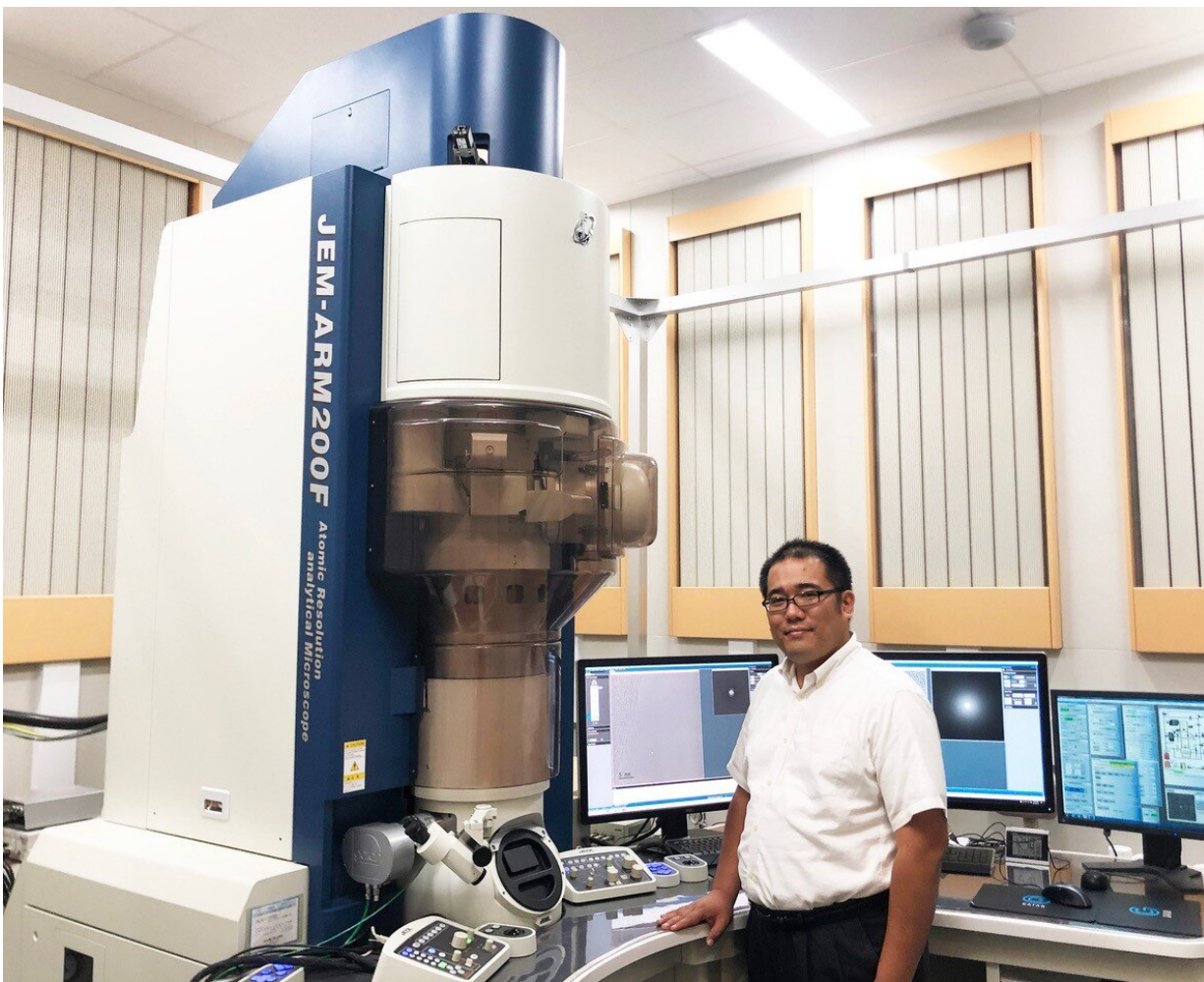


A video showing for the first time the cubic molecule critical to MOF crystals.  
Credit: (c) 2019 Nakamura et al.

The transient stages of complex chemical reactions are difficult to study as there are multiple intermediate processes that occur between the start and end of most reactions. In principle, the individual stages could be seen, but in reality, it was impossible to isolate the products at each stage and to see how these changed with time. Nakamura, Harano and team spent over 10 years on this problem, and have developed a method called molecular electron microscopy.

"It was a problem of two parts," said Harano. "On the large scale, there was an engineering challenge to combine a uniquely high-resolution electron microscope with a fast and sensitive imaging sensor for continuous video imaging; while on the small scale, we had to devise a way to capture the molecules of interest and hold them in place so the camera could catch the action."

To isolate and secure particular molecules, the team used a specially modified carbon nanotube. This would snag a passing molecule and hold it in place, but crucially, would not interfere with the reactions of that molecule. This way, every stage of the reaction would take place at the tip of the nanotube, which in turn was held in place at the focal point of the electron microscope. The resulting data can be made into real-time videos of the reactions.



Project Associate Professor Harano at the controls of the research group's unique electron microscope. Credit: (c) 2019 Harano et al.

"What surprised us very much in the beginning was that our plan actually worked. It was a complex challenge, but we first visualized these molecular videos in 2013," said Harano. "Between then and now, we worked to turn the concept into a useful tool. Our first success was to visualize and describe a cube-shaped molecule, which is a crucial intermediate form that occurs during MOF synthesis. It took a year to convince our reviewers what we found is real."

This is just the first step toward the ability to gain control over [chemical synthesis](#) in a precise and controlled manner—a term the researchers call "rational [synthesis](#)." It's important to observe details of reactions as they progress so they can effectively be reverse-engineered. The dream 200 years ago was to see an atom; the dream now is to control molecules in order to create things like synthetic minerals for construction, or even new drugs to save lives.

**More information:** Junfei Xing, Luca Schweighauser, Satoshi Okada, Koji Harano, Eiichi Nakamura. Atomistic structures and dynamics of prenucleation clusters in MOF-2 and MOF-5 syntheses. *Nature Communications*. [DOI: 10.1038/s41467-019-11564-4](https://doi.org/10.1038/s41467-019-11564-4)

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