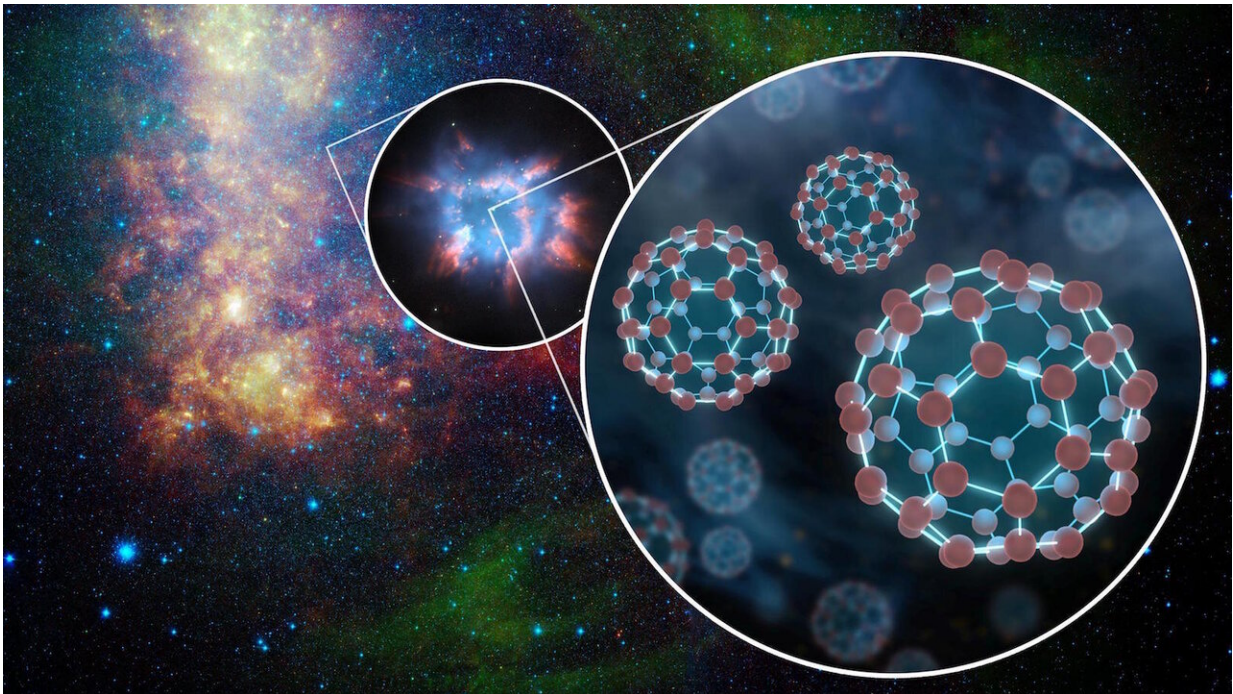


Interstellar iron isn't missing, it's just hiding in plain sight

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Carbon-chain molecules as complex as C₆₀ buckminsterfullerenes -- 'buckyballs' -- may form in space with the help of clustered iron atoms, according to new work by ASU cosmochemists. The work also explains how these iron clusters hide out inside common carbon-chain molecules. Credit: NASA/JPL-Caltech

Astrophysicists know that iron (chemical symbol: Fe) is one of the most abundant elements in the universe, after lightweight elements such as hydrogen, carbon, and oxygen. Iron is most commonly found in gaseous

form in stars such as the Sun, and in more condensed form in planets such as Earth.

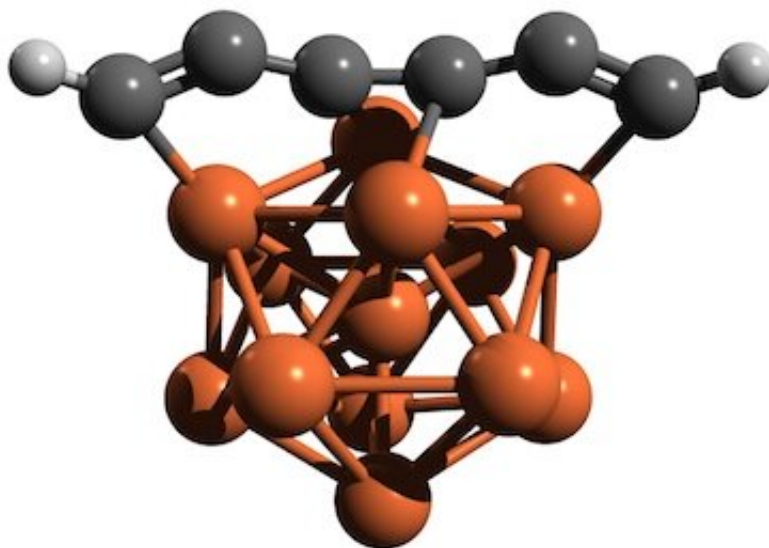
Iron in interstellar environments should also be common, but astrophysicists detect only low levels of the gaseous kind. This implies that the missing [iron](#) exists in some kind of solid form or molecular state, yet identifying its hiding place has remained elusive for decades.

A team of cosmochemists at Arizona State University, with support from the W.M. Keck Foundation, now claims that the mystery is simpler than it seems. The iron isn't really missing, they say. Instead it's hiding in plain sight. The iron has combined with [carbon molecules](#) to form [molecular chains](#) called iron pseudocarbynes. The spectra of these chains are identical with the much more common chains of carbon molecules, long known to be abundant in interstellar space.

The team's work was published late in June in the *Astrophysical Journal*.

"We are proposing a new class of molecules that are likely to be widespread in the interstellar medium," said Pilarasetty Tarakeshwar, research associate professor in ASU's School of Molecular Sciences. His coauthors, Peter Buseck and Frank Timmes, are both in ASU's School of Earth and Space Exploration; Buseck, an ASU Regents Professor, is also in the School of Molecular Sciences with Tarakeshwar.

The team examined how clusters containing only a few atoms of metallic iron might join with chains of carbon molecules to produce molecules combining both elements.



Iron pseudocarbonyls are likely widespread in the interstellar medium, where extremely cold temperatures would lead carbon chains to condense on the Fe clusters. Over eons, complex organic molecules would emerge from these Fe pseudocarbonyls. The model shows a hydrogen-capped carbon chain attached to an Fe₁₃ cluster (iron atoms are reddish brown, carbon is gray, hydrogen is light gray). Credit: P. Tarakeshwar/ASU

Recent evidence obtained from stardust and meteorites indicate the widespread occurrence of clusters of iron atoms in the cosmos. In the extremely cold temperatures of interstellar space, these iron clusters act as deep-freeze particles, enabling carbon chains of various lengths to stick to them, thus producing different molecules from those that can occur with the gaseous phase of iron.

Said Tarakeshwar, "We calculated what the spectra of these molecules would look like, and we found that they have spectroscopic signatures nearly identical to carbon-chain molecules without any iron." He added

that because of this, "Previous astrophysical observations could have overlooked these carbon-plus-iron molecules."

That means, the researchers say, the missing iron in the [interstellar medium](#) is actually out in plain view but masquerading as common carbon-chain molecules.

The new work may also solve another longstanding puzzle. Carbon chains with more than nine atoms are unstable, the team explains. Yet observations have detected more complex carbon molecules in [interstellar space](#). How nature builds these complex carbon molecules from simpler carbon molecules has been a mystery for many years.

Buseck explained, "Longer carbon chains are stabilized by the addition of iron clusters." This opens a new pathway for building more complex molecules in space, such as [polyaromatic hydrocarbons](#), of which naphthalene is a familiar example, being the main ingredient in mothballs.

Said Timmes, "Our work provides new insights into bridging the yawning gap between molecules containing nine or fewer [carbon](#) atoms and complex molecules such as C60 buckminsterfullerene, better known as 'buckyballs.'"

More information: Pilarisetty Tarakeshwar et al, On the Structure, Magnetic Properties, and Infrared Spectra of Iron Pseudocarbonyls in the Interstellar Medium, *The Astrophysical Journal* (2019). [DOI: 10.3847/1538-4357/ab22b7](#)

Provided by Arizona State University

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